

INTRODUÇÃO À FÍSICA NUCLEAR

FNC404

1º semestre de 2011

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endereço eletrônico do curso:

PROGRAMA

1) INTRODUÇÃO: O QUE É A FÍSICA NUCLEAR- SUAS QUESTÕES - SUA FRONTEIRA – SUAS APLICAÇÕES

2) O PROGRAMA:

BLOCO 1: INTRODUÇÃO

BLOCO 2: GRANDEZAS MACROSCÓPICAS DO NÚCLEO

BLOCO 3: FÍSICA NUCLEAR DE ALTAS ENERGIAS : FÍSICA HADRÔNICA

BLOCO 4: MODELOS NUCLEARES

BLOCO 5: REAÇÕES NUCLEARES

BLOCO 6: ACELERADORES E INSTRUMENTAÇÃO NUCLEAR

3) PROJETO E/OU TRABALHO DE FIM DE CURSO

PROVAS

PRIMEIRA PROVA (P1): 29 MARÇO

SEGUNDA PROVA (P2): 6 MAIO

TERCEIRA PROVA (P3): 17 JUNHO

**PROJETO E/OU TRABALHO DE FIM DE CURSO:
NOTA TT**

MEDIA: $M = 1/3 (P_1 + P_2 + P_3) + \alpha .TT$

$M > 5 \rightarrow$ APROVADO

$M < 3 \rightarrow$ REPROVADO

$3 < M < 5 \rightarrow$ REC se $1/3 (M + 2REC) > 5 \rightarrow$ APROVADO

Referências

Nuclear and Particle Physics

(W.S.C. Williams) - Clarendon Press-Oxford

Física Nuclear

(Theo Mayer-Kuckuk) - Fundação Calouste Gulbenkian

Introductory Nuclear Physics

Kenneth S. Krane – John Wiley & Sons

Introduction to Nuclear Reactions

(G.R. Satchler) – MacMillan Education Ltd

Introdução à Física Nuclear

(K.C. Chung) – editora UERJ

BLOCO 1

INTRODUÇÃO

EVOLUÇÃO DO UNIVERSO:

BIG BANG - HADRONIZAÇÃO - FORMAÇÃO DAS ESTRELAS

A MATÉRIA NUCLEAR

A NUCLEO-SÍNTESE DOS ELEMENTOS NAS ESTRELAS:

OS CICLOS p-p , CNO, O SOL, PROCESSOS r, rp ...

O NÚCLEO E SUA ESTRUTURA : TABELA PERIÓDICA

O CAMINHO INVERSO: COLISÕES DE ÍONS PESADOS RELATIVÍSTICOS

APLICAÇÕES DA FÍSICA NUCLEAR

TECNOLOGIA NUCLEAR

Referências para o bloco 1

<http://particleadventure.org/particleadventure/>

<http://ie.lbl.gov/education/glossary/glossaryf.htm>

<http://www.cpepweb.org/>

<http://books.nap.edu/books/0309062764/html/index.html>

<http://webphysics.ph.msstate.edu/jc/library/>

<http://WWW.PHYS.VIRGINIA.EDU/CLASSES/252/home.html>

http://WWW.PHYS.VIRGINIA.EDU/CLASSES/252/Nuclear_Notes/nuclear_notes.html

<http://www.overture.com/d/search/p/netscape/?Keywords=rutherford+scattering+applets>

<http://www2.slac.stanford.edu/vvc/Default.htm>

<http://www2.slac.stanford.edu/vvc/Default.htm>

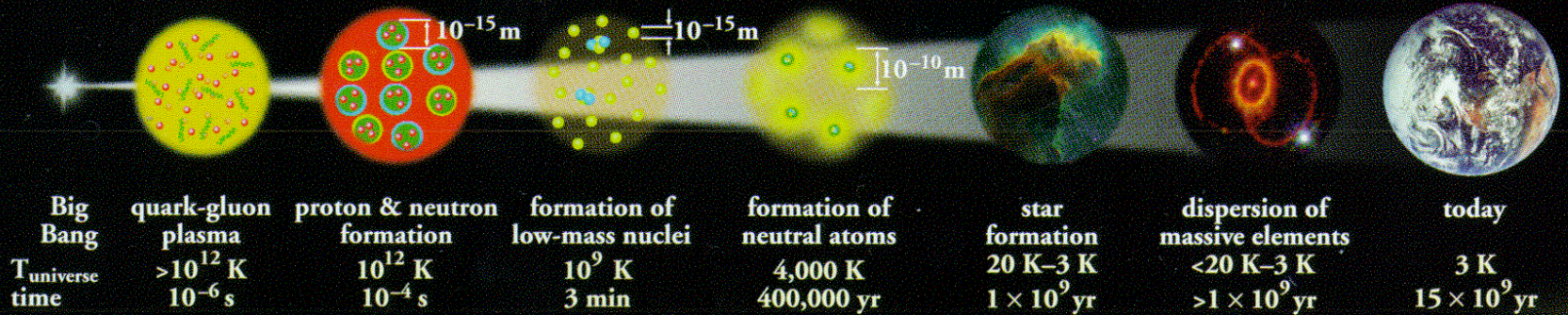
<http://home.a-city.de/walter.fendt/phe/decayseries.htm>

http://www.phys.virginia.edu/classes/109N/more_stuff/Applets/rutherford/rutherford2.html

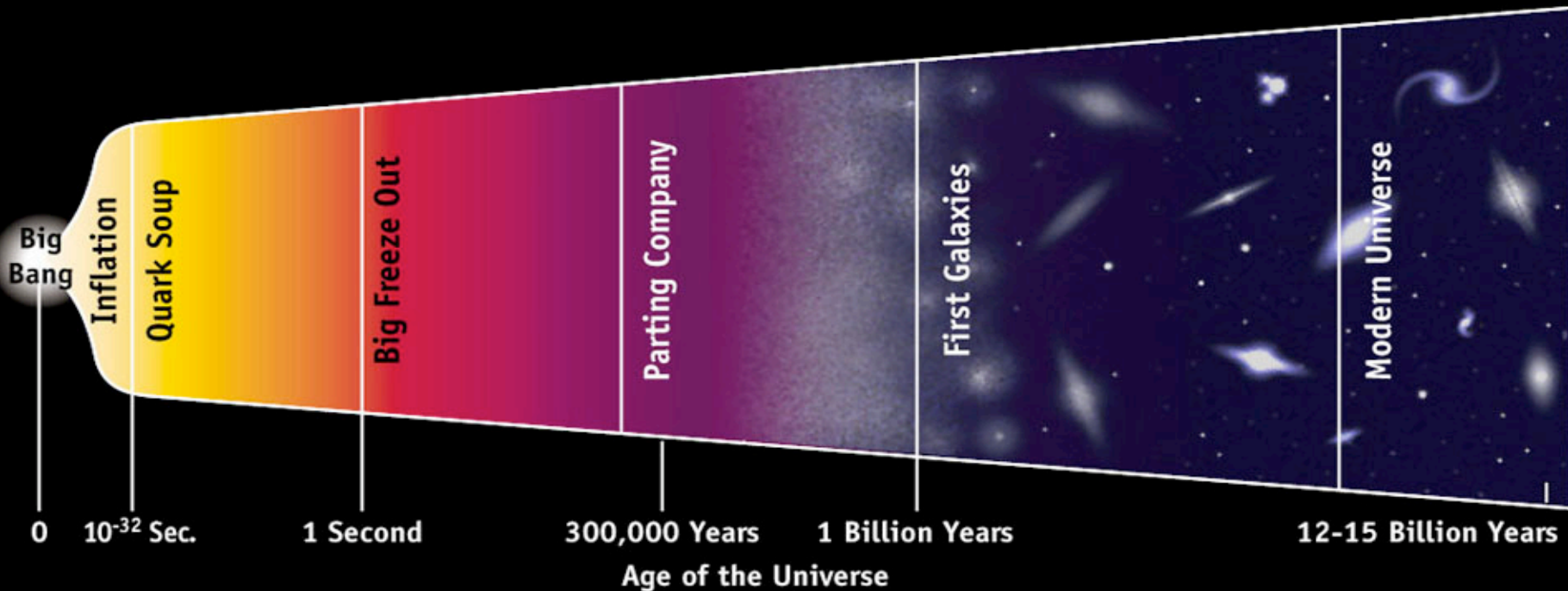
http://www.phys.virginia.edu/classes/109N/more_stuff/Applets/rutherford/rutherford.html

Expansion of the Universe

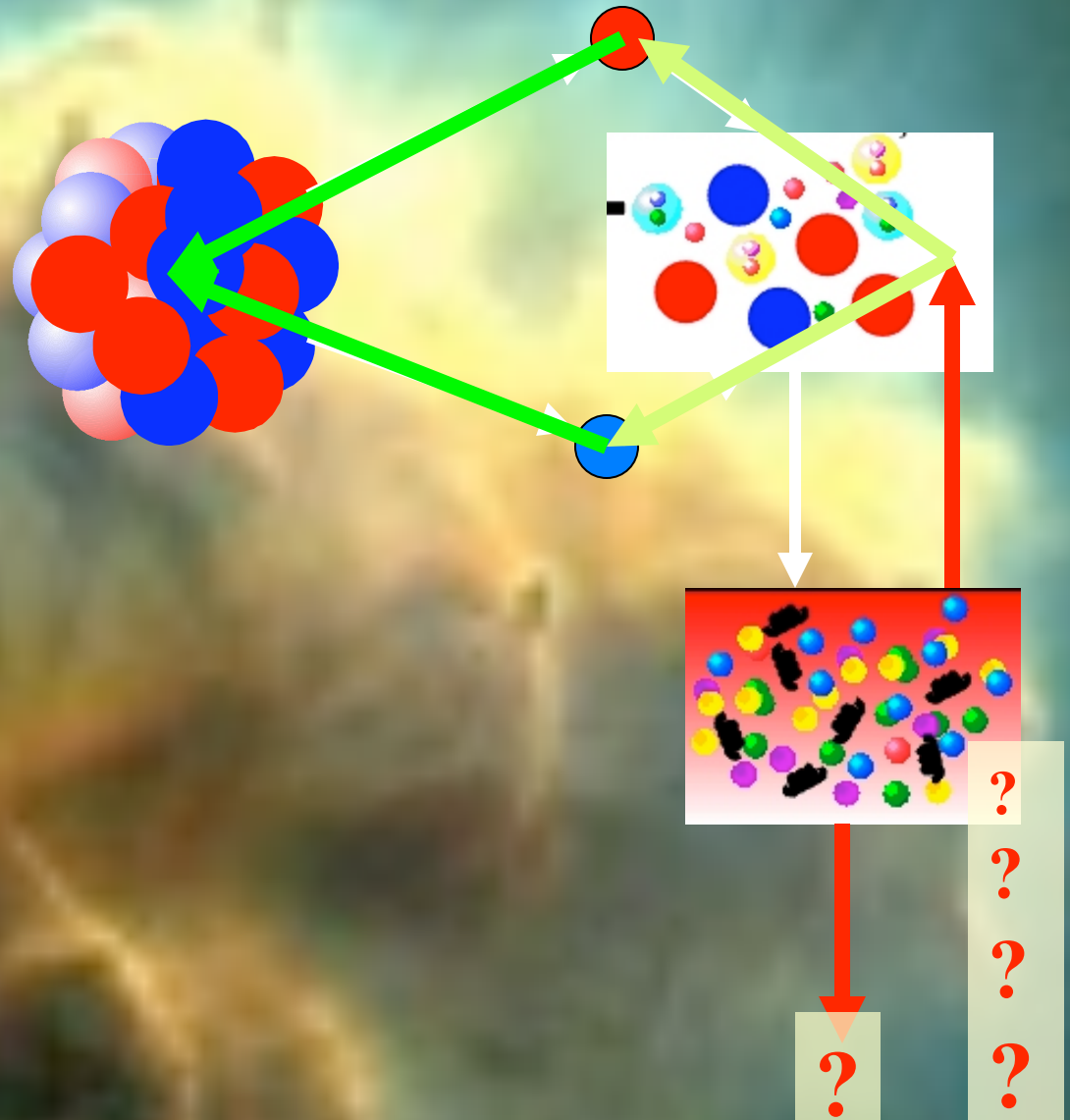
After the Big Bang, the universe expanded and cooled. At about 10^{-6} second, the universe consisted of a soup of quarks, gluons, electrons, and neutrinos. When the temperature of the Universe, T_{universe} , cooled to about 10^{12} K, this soup coalesced into protons, neutrons, and electrons. As time progressed, some of the protons and neutrons formed deuterium, helium, and lithium nuclei. Still later, electrons combined with protons and these low-mass nuclei to form neutral atoms. Due to gravity, clouds of atoms contracted into stars, where hydrogen and helium fused into more massive chemical elements. Exploding stars (supernovae) form the most massive elements and disperse them into space. Our earth was formed from supernova debris.



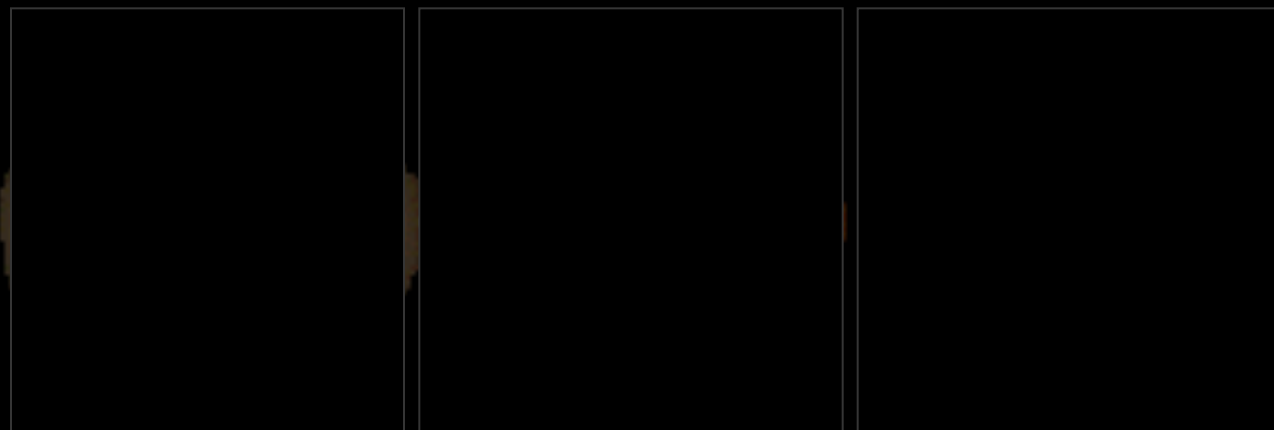
← Radius of the Visible Universe →



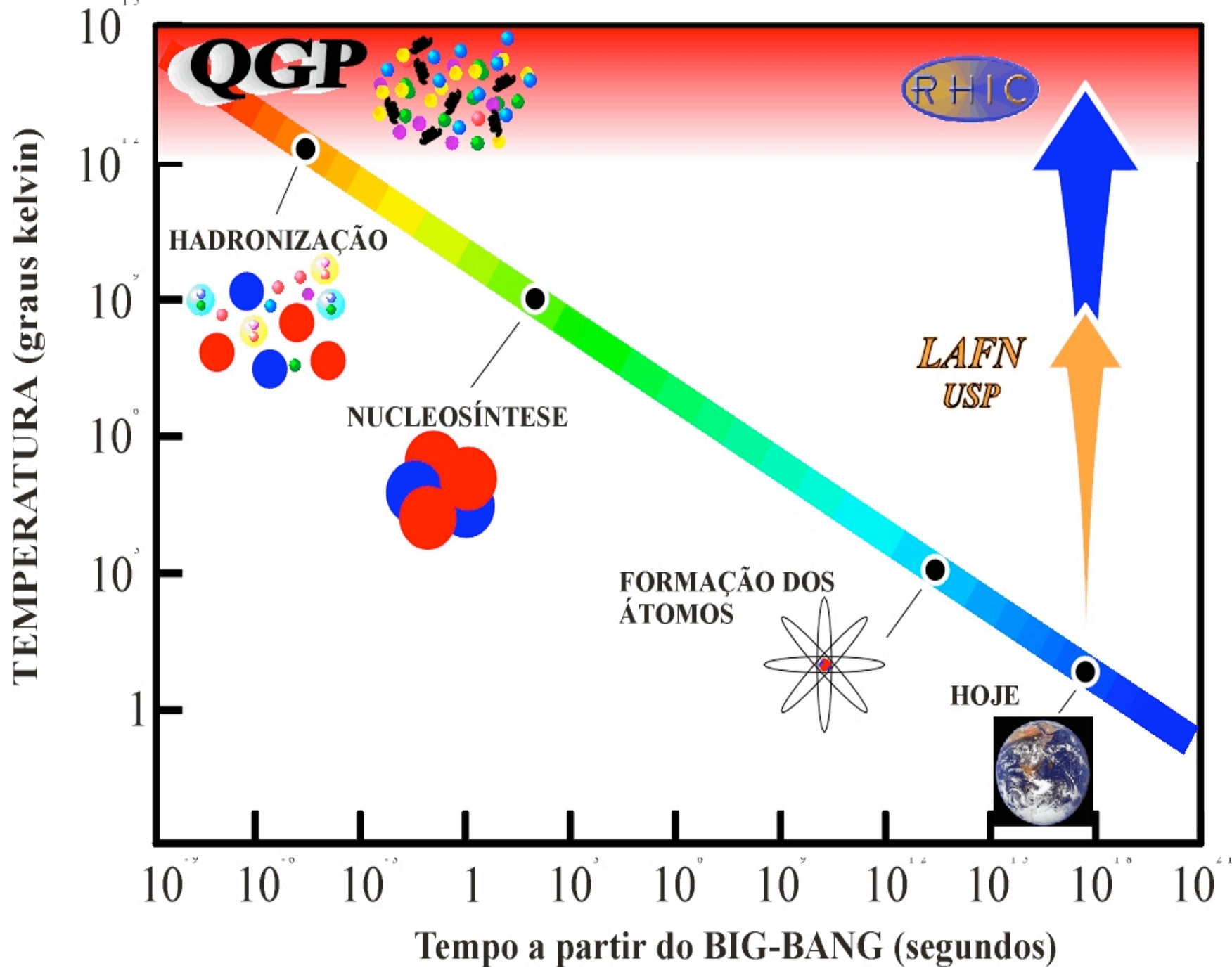




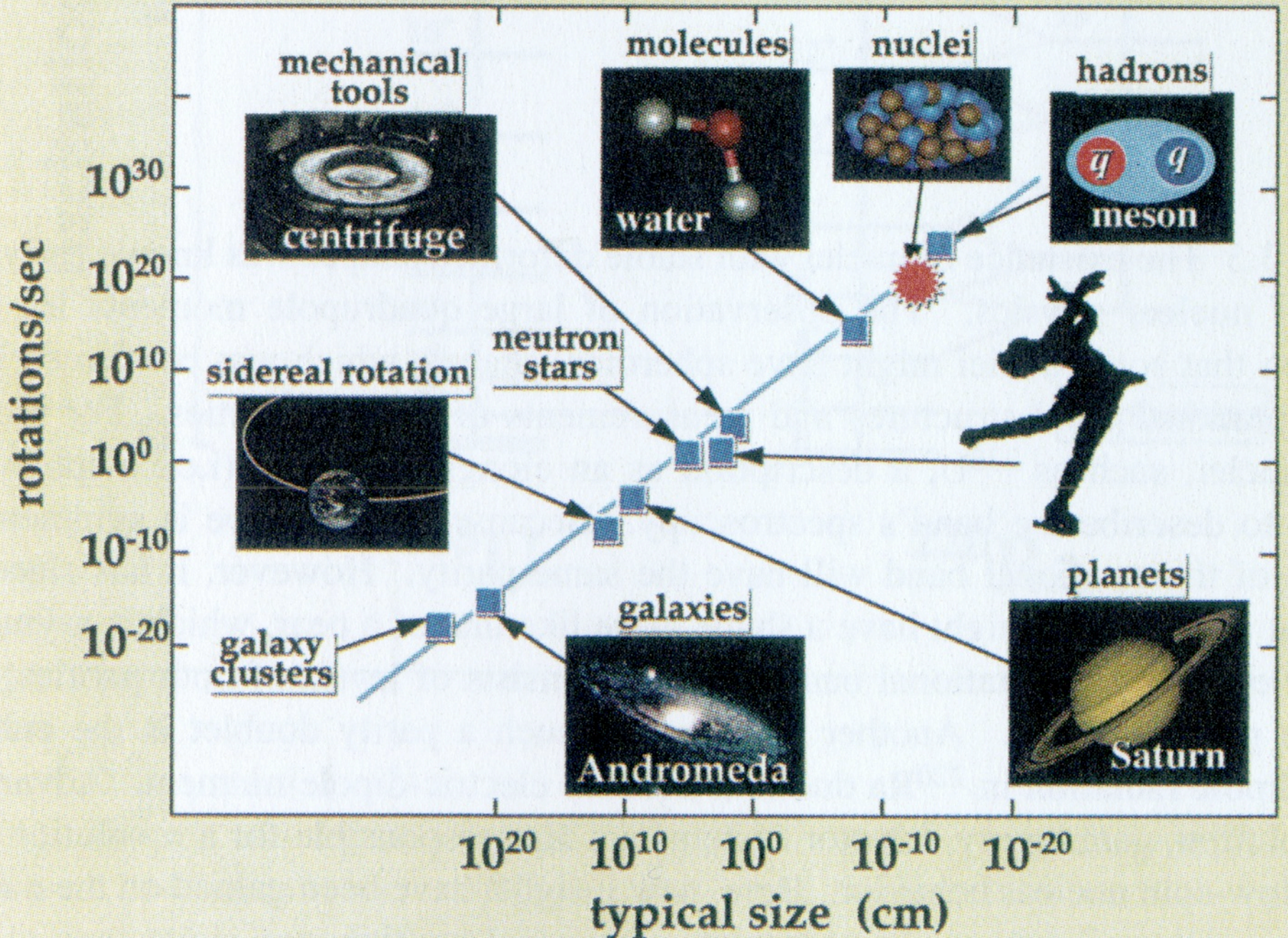
nebula



increasing temperature



rotations in the universe





NGC 6543

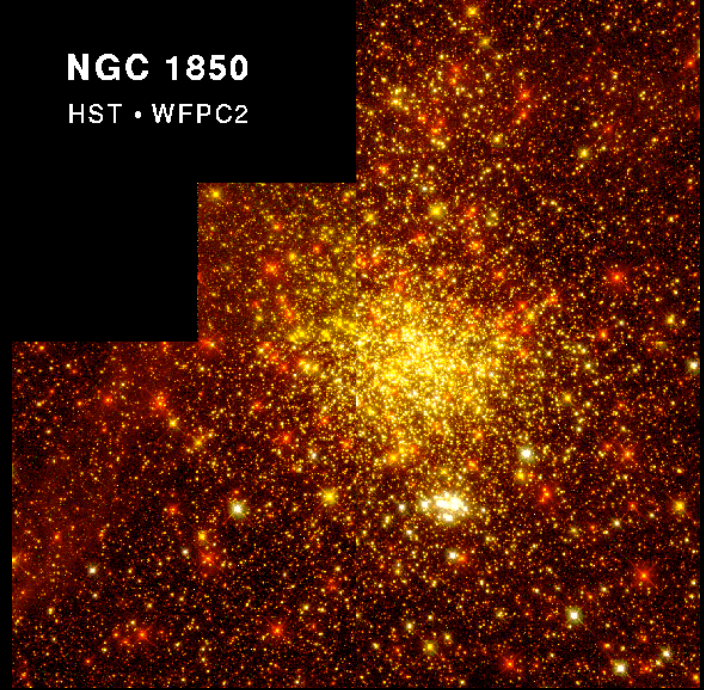
PR95-01a - ST ScI OPO - January 1995 - P. Harrington (U.MD), NASA

HST • WFPC2

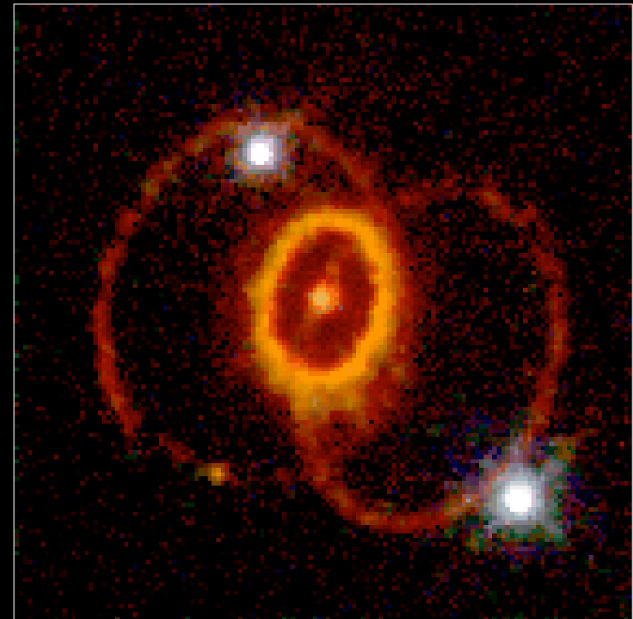
12/13/94 zgl

NGC 1850

HST • WFPC2



Supernova 1987A Rings

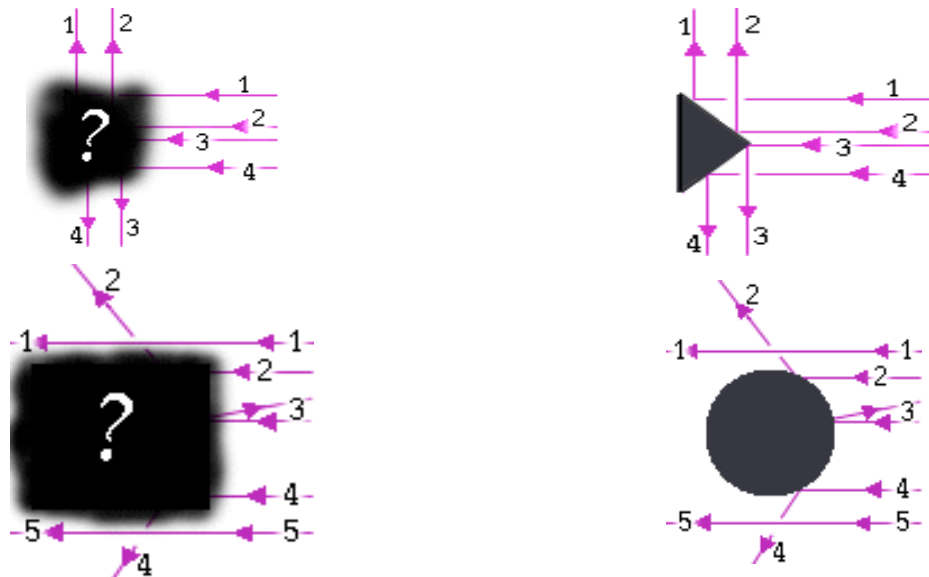


Hubble Space Telescope



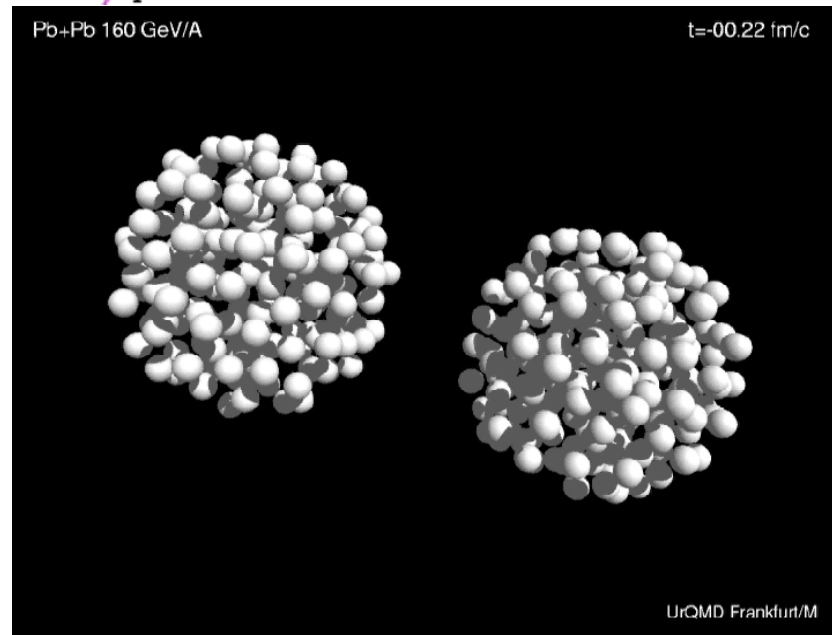
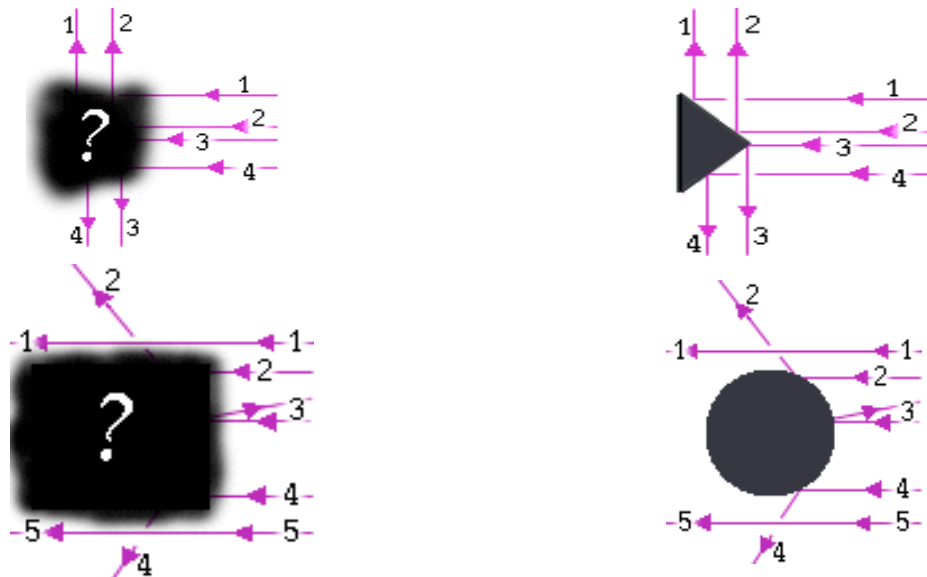
COMO SE FAZ FÍSICA NUCLEAR

DO PONTO DE VISTA EXPERIMENTAL

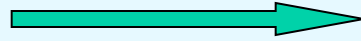


COMO SE FAZ FÍSICA NUCLEAR

DO PONTO DE VISTA EXPERIMENTAL



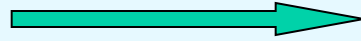
NUCLEO DE THOMSON



ANIMAÇÃO

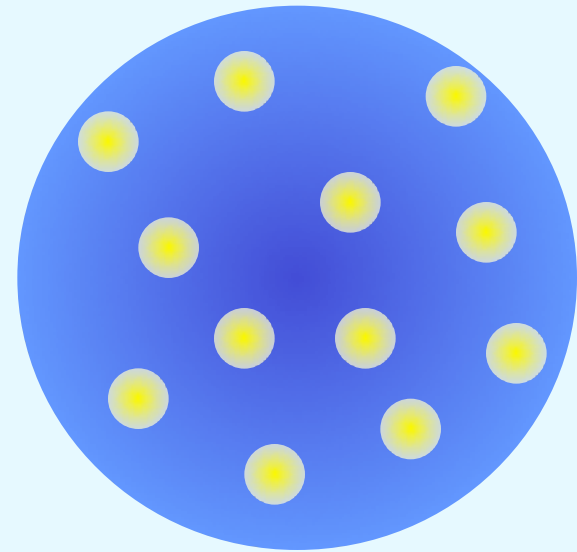
http://galileoandeinstein.physics.virginia.edu/more_stuff/Applets/rutherford/rutherford2.html

NUCLEO DE THOMSON



ANIMAÇÃO

http://galileoandeinstein.physics.virginia.edu/more_stuff/Applets/rutherford/rutherford2.html

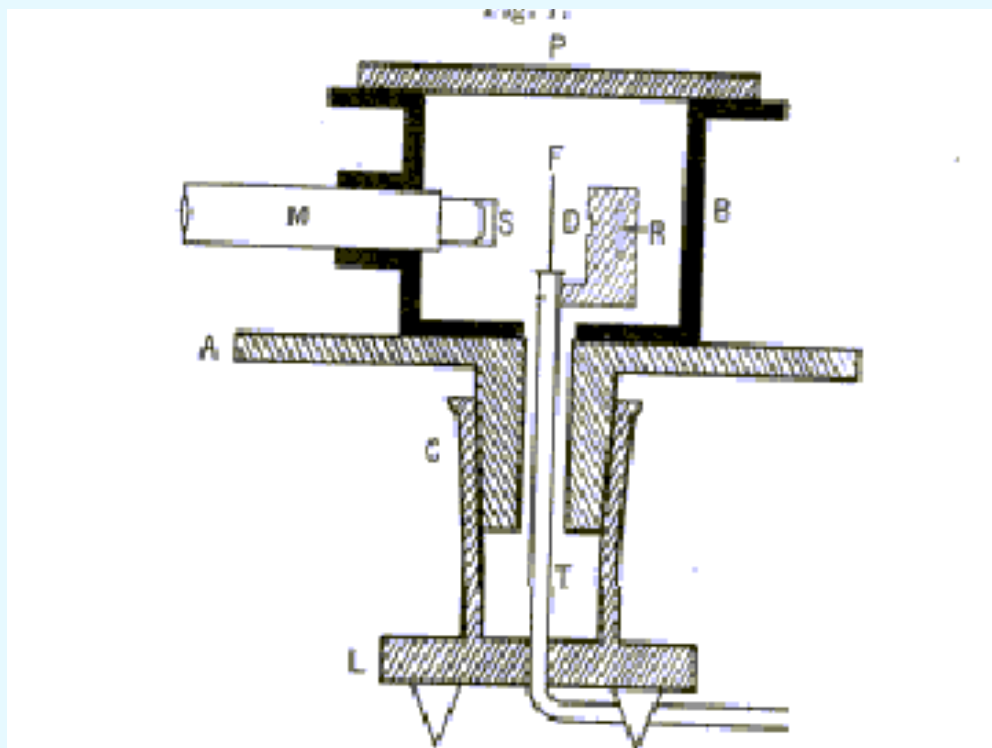


^{24}Mg

NUCLEO DE RUTHERFORD



ANIMAÇÃO



http://galileoandeinstein.physics.virginia.edu/more_stuff/Applets/rutherford/rutherford.html

<http://www.nat.vu.nl/~pwgroen/projects/sdm/applets.htm>

http://www.nat.vu.nl/~pwgroen/sdm/hyper/anim/anim_DI.html

<http://physics.uwstout.edu/physapplets/>

<http://micro.magnet.fsu.edu/electromag/java/rutherford/>

Disproof of the Pudding

The back scattered alpha-particles proved fatal to the plum pudding model. A central assumption of that model was that both the positive charge and the mass of the atom were more or less uniformly distributed over its size, approximately 10⁻¹⁰ meters across or a little more. It is not difficult to calculate the magnitude of electric field from this charge distribution. (Recall that this is the field that must scatter the alphas, the electrons are so light they will jump out of the way with negligible impact on an alpha.)

To be specific, let us consider the gold atom, since the foil used by Rutherford was of gold, beaten into leaf about 400 atoms thick. The gold atom has a positive charge of 79e (balanced of course by that of the 79 electrons in its normal state). Neglecting these electrons -- assume them scattered away -- the maximum electric force the alpha will encounter is that at the surface of the sphere of positive charge,

$$E \cdot 2e = \frac{1}{4\pi\epsilon_0} \cdot \frac{79e \cdot 2e}{r_0^2} = 9 \cdot 10^9 \cdot \frac{158 \cdot (1.6 \cdot 10^{-19})^2}{10^{-20}} = 3.64 \cdot 10^{-6} \text{ newtons}$$

If the alpha particle initially has momentum p , for small deflections the angle of deflection (in radians) is given by $(\Delta p)/p$, where Δp is the sideways momentum resulting from the electrically repulsive force of the positive sphere of charge. Assuming the atomic sphere itself moves negligibly -- it is much heavier than the alpha, so this is reasonable -- the trajectory of the alpha in the inverse square electric field can be found by standard methods. It is the same mathematical problem as finding the elliptic orbits of planets around the sun. Replacing inverse square attraction with inverse square repulsion changes the orbit from an ellipse (or a hyperbola branch swinging around the sun for a comet) to a hyperbola branch lying on one side of the center of repulsion.

Note that since the alpha particle has mass 6.7×10^{-27} kg, from $F = ma$, the electric force at the atomic surface above will give it a sideways acceleration of 5.4×10^{20} meters per sec per sec (compare $g = 10!$). But the force doesn't have long to act - the alpha is moving at 1.6×10^7 meters per second. So the time available for the force to act is the time interval a particle needs to cross an atom if the particle gets from New York to Australia in one second.

The time $t_0 = 2r_0/v = 2 \times 10^{-10} / 1.6 \times 10^7 = 1.25 \times 10^{-17}$ seconds.

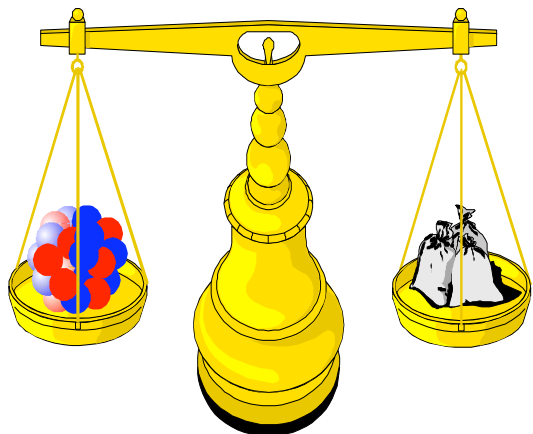
Thus the magnitude of the total sideways velocity picked up is the sideways acceleration multiplied by the time,

$1.25 \times 10^{-17} \times 5.4 \times 10^{20} = 6750$ meters per second.

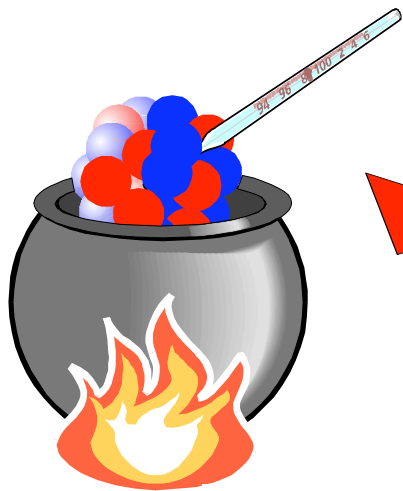
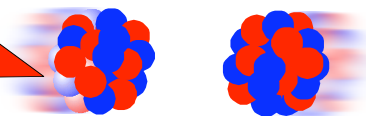
This is a few ten-thousandths of the alpha's forward speed, so there is only a very tiny deflection. Even if the alpha hit 400 atoms in succession and they all deflected it the same way, an astronomically improbable event, the deflection would only be of order a

In fact, one can get a clear idea of how much deflection comes about without going into the details of the trajectory. Outside the atom, the repulsive electrical force falls away as the inverse square. Inside the atom, the force drops to zero at the center, just as the gravitational force is zero at the center of the earth. The force is maximum right at the surface. Therefore, a good idea of the sideways deflection is given by assuming the alpha experiences that maximal force for a time interval equal to the time it takes the alpha to cross the atom -- say, a distance $2r_0$. degree. Therefore, the observed deflection through *ninety* degrees and more was completely inexplicable using Thomson's pudding model!

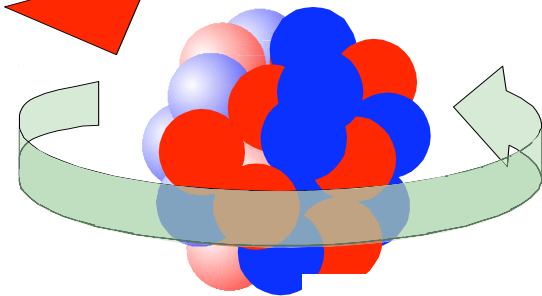
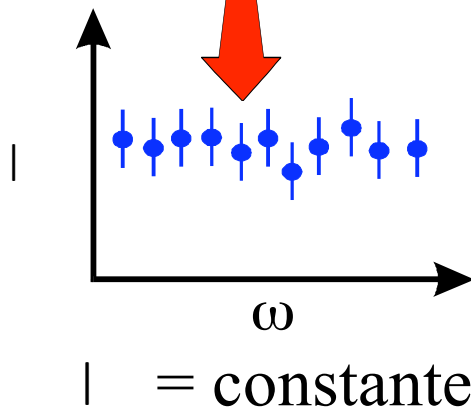
$$\rho \sim 10^{18} \text{ Kg/m}^3$$



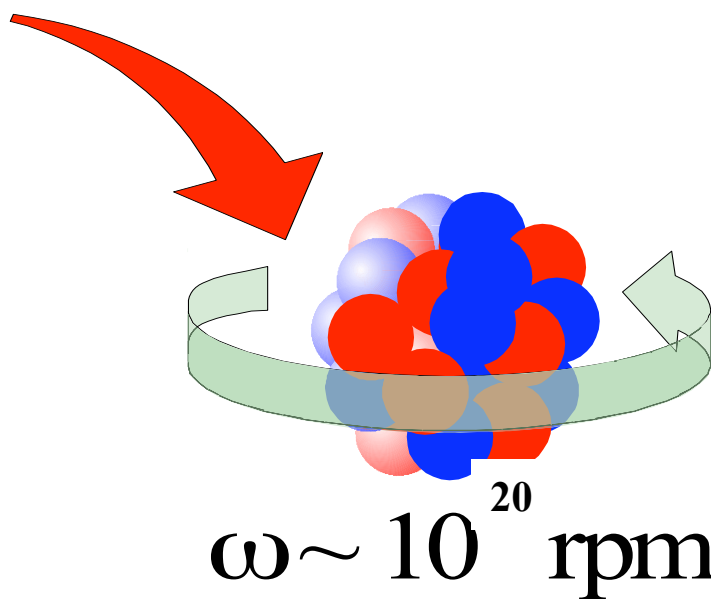
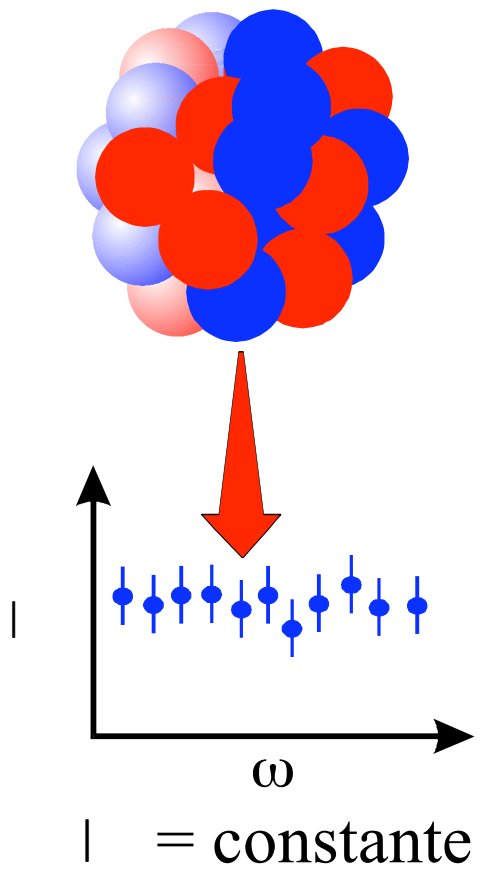
$$\tau = 10^{-16} - 10^{-22} \text{ s}$$

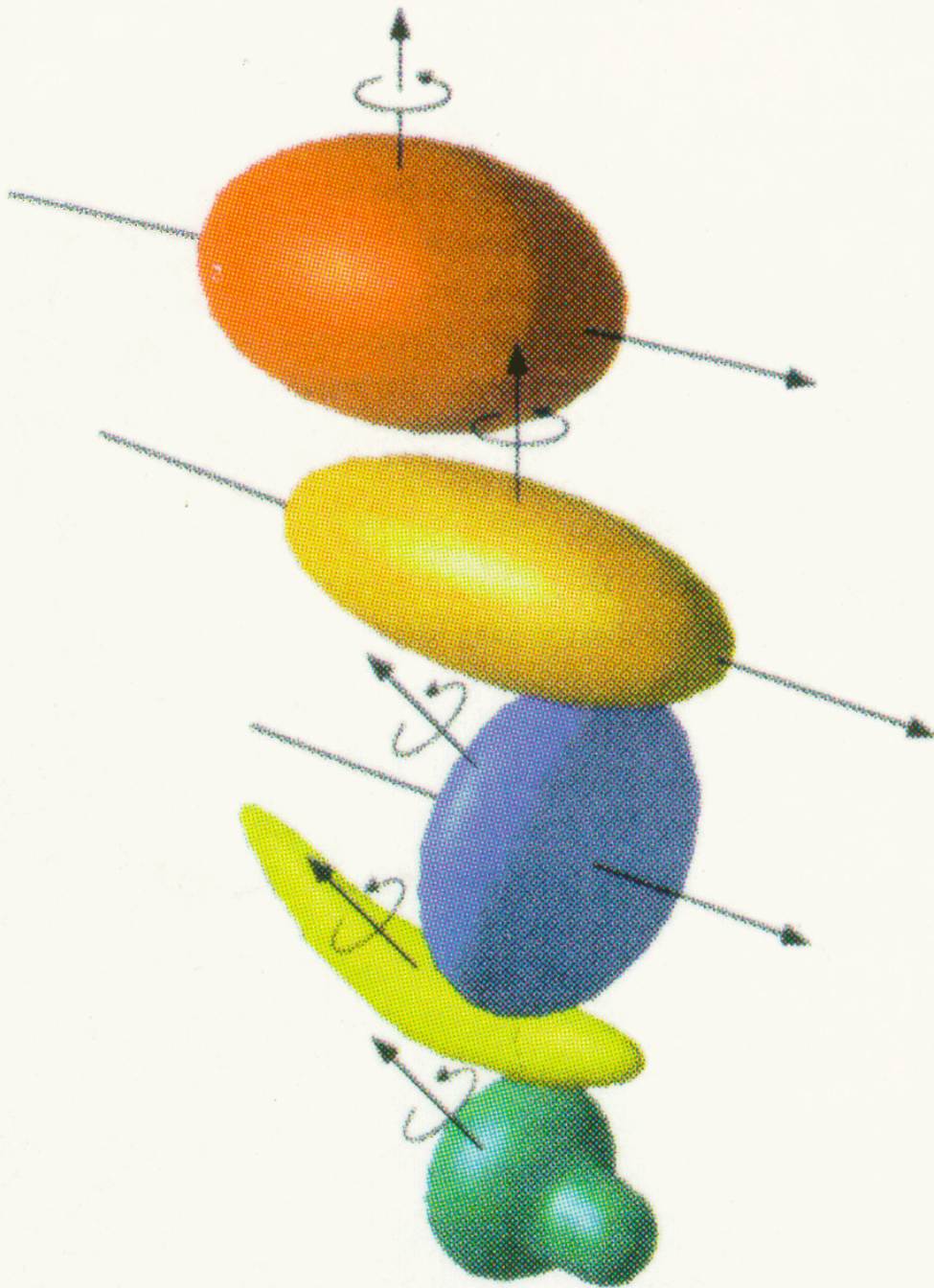


$$T \sim 10^{11} \text{ K}$$

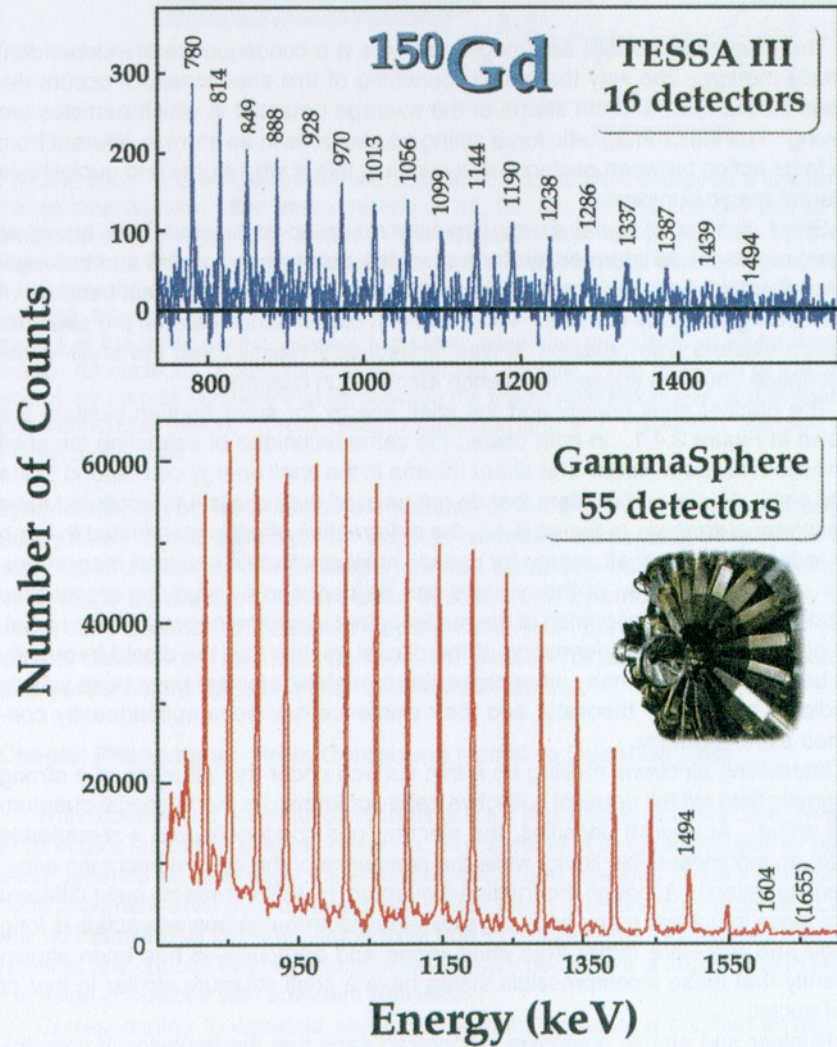


$$\omega \sim 10^{20} \text{ rpm}$$



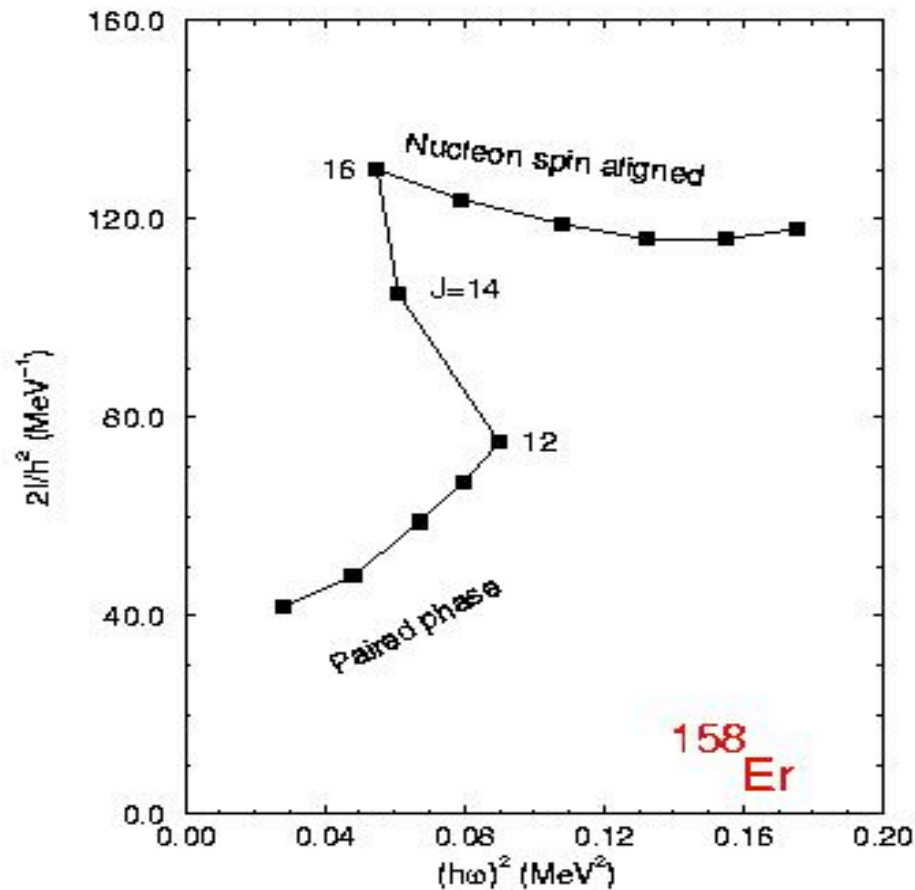


INCLUR DIAGRAMA DE NIVEIS



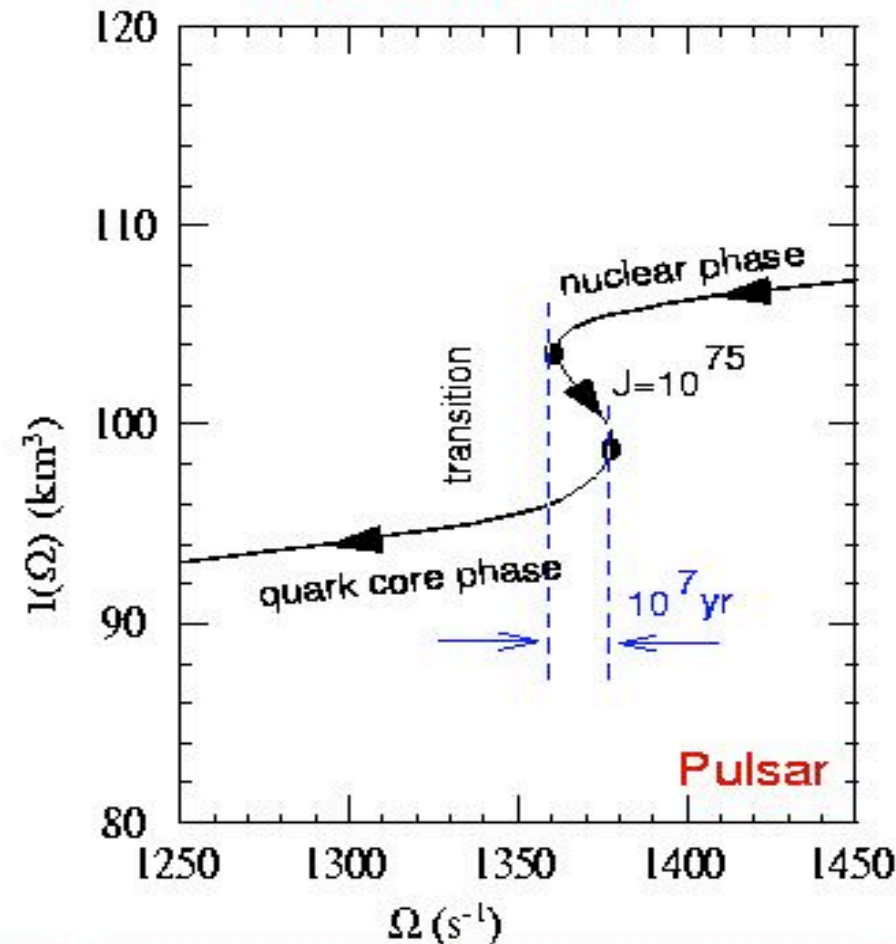
Quark Deconfinement during Pulsar Spin-Down

Deformed Nucleus



Backbending predicted in 1960 by Mottelson and Valatin (phase transition from spin aligned state at high Ω to pair-correlated superfluid at low Ω)

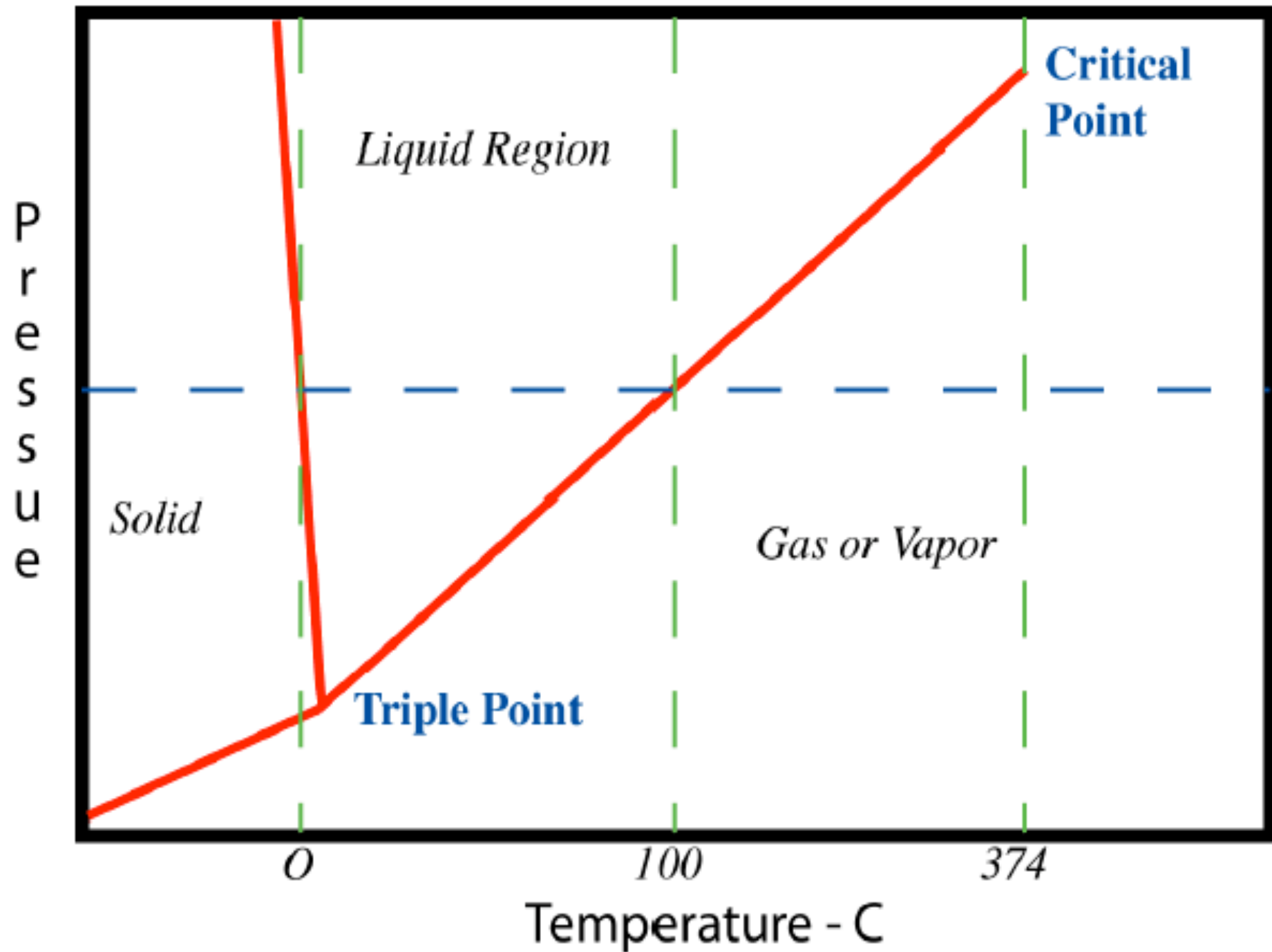
Neutron Star



Backbending in pulsar may result if phase transition occurs.

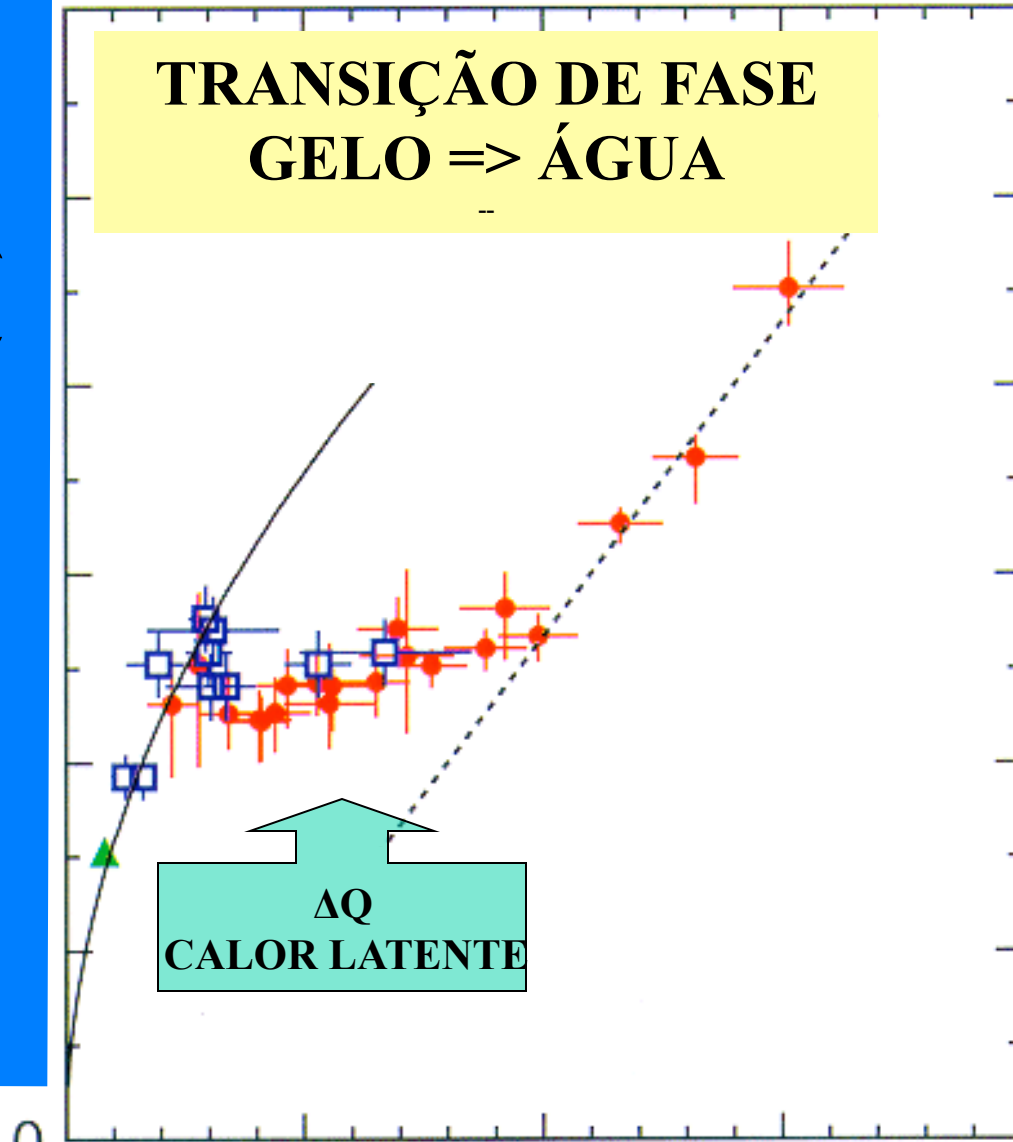
1) Átomo de Bohr: diagrama de níveis e espectro de raios X
<http://www.walter-fendt.de/ph14e/bohrh.htm>

2) Espectro nuclear de raios Gamma

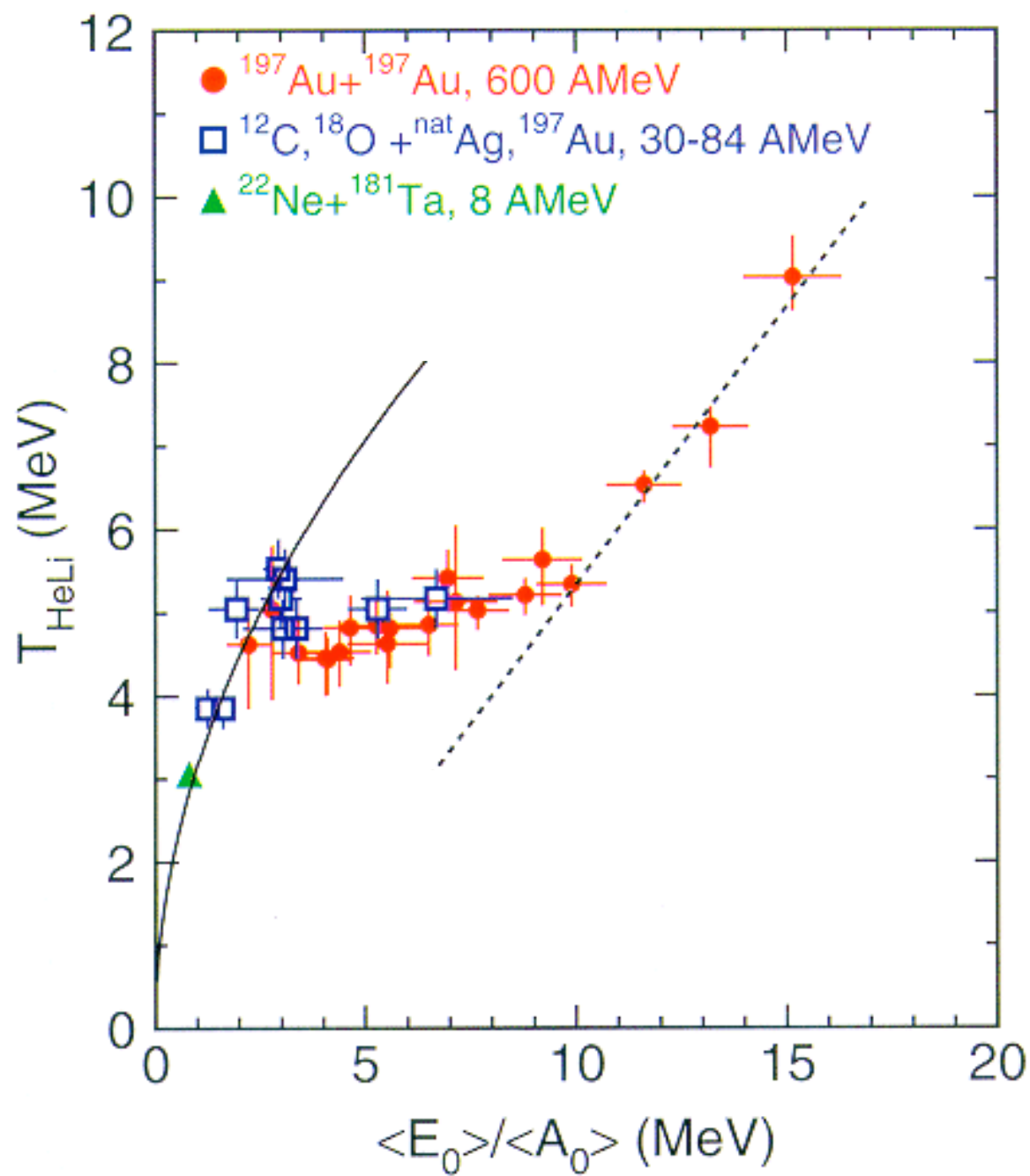


TEMPERATURA (K)

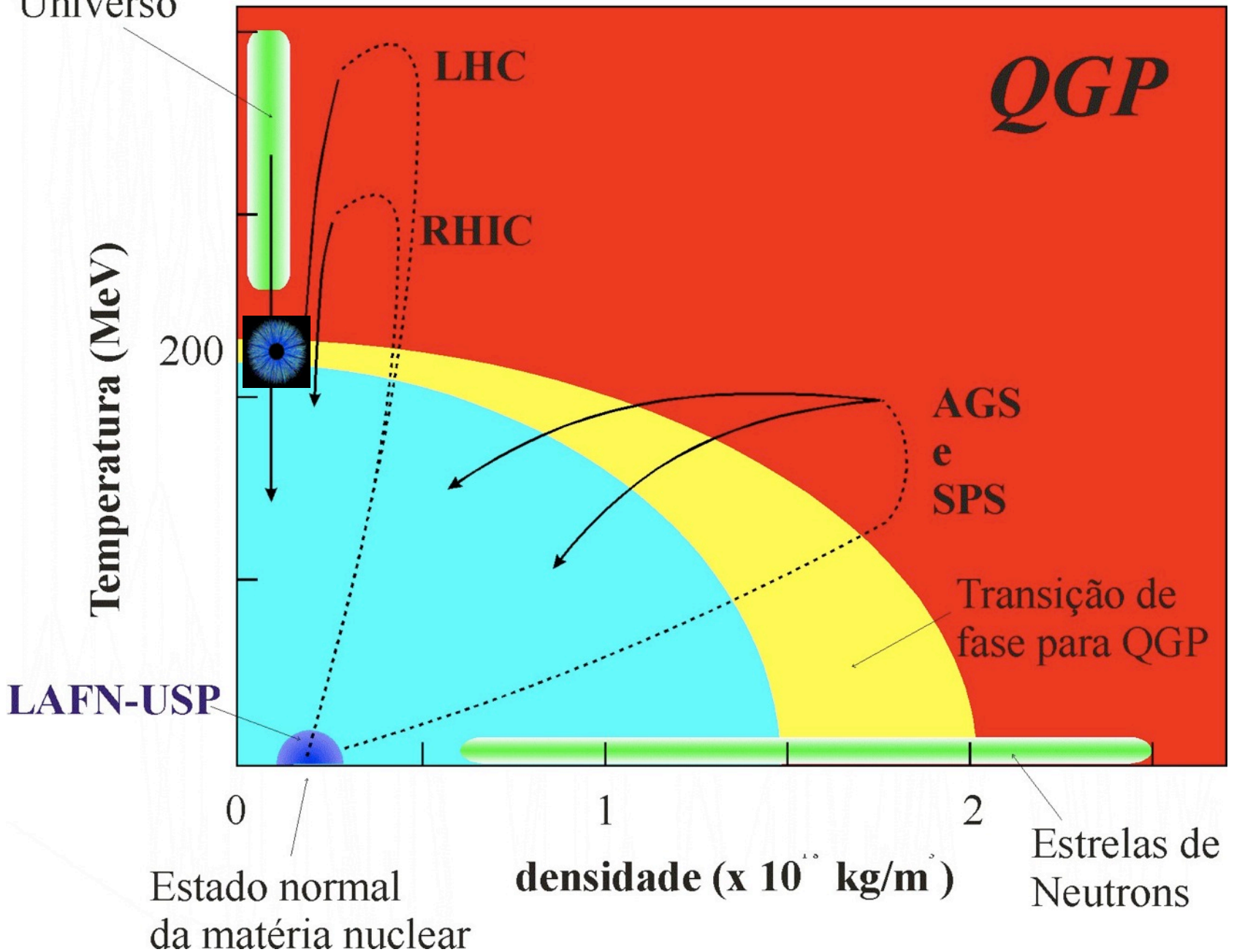
TRANSIÇÃO DE FASE
GELO => ÁGUA



QUANTIDADE DE CALOR (joules)



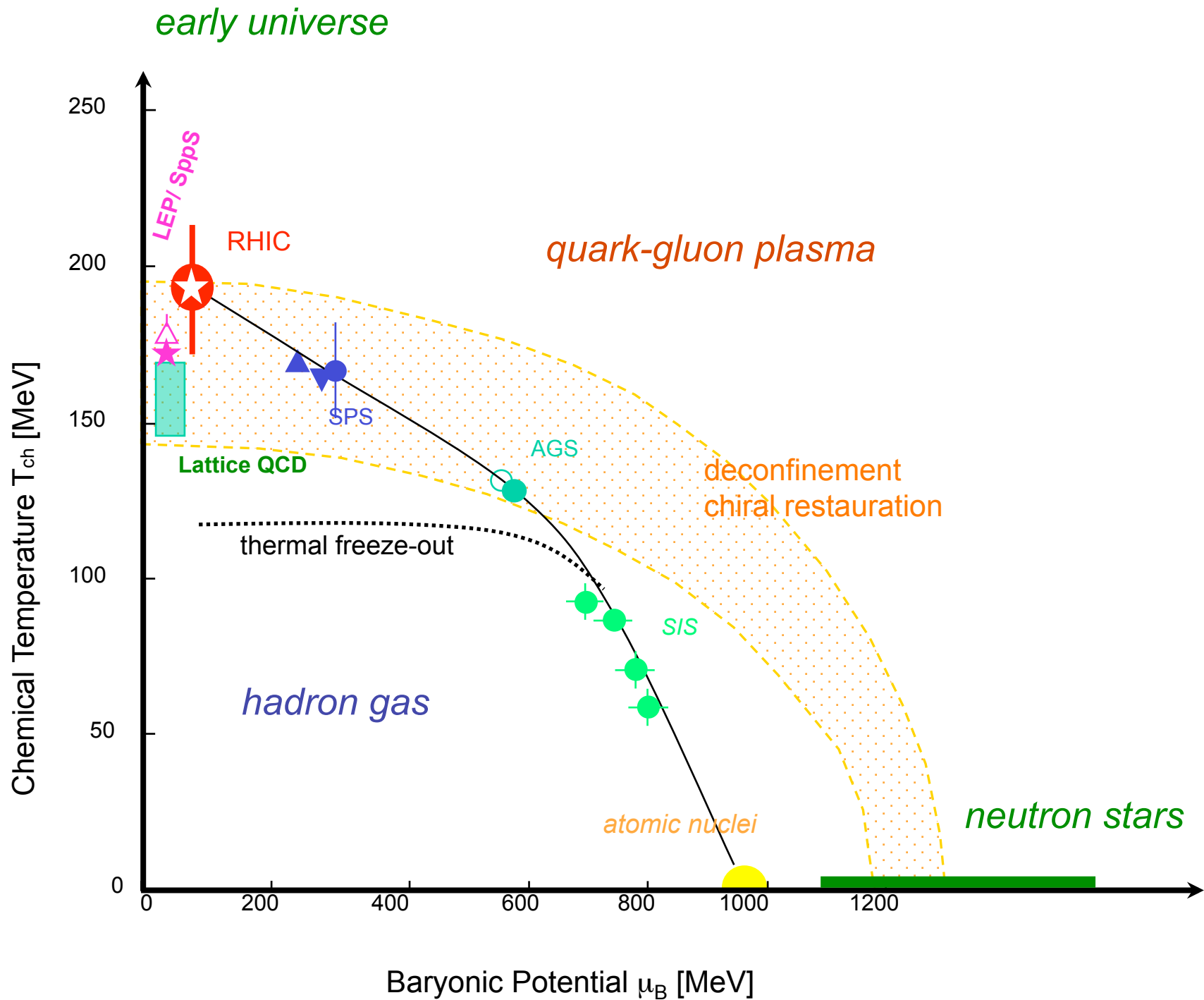
Início do Universo



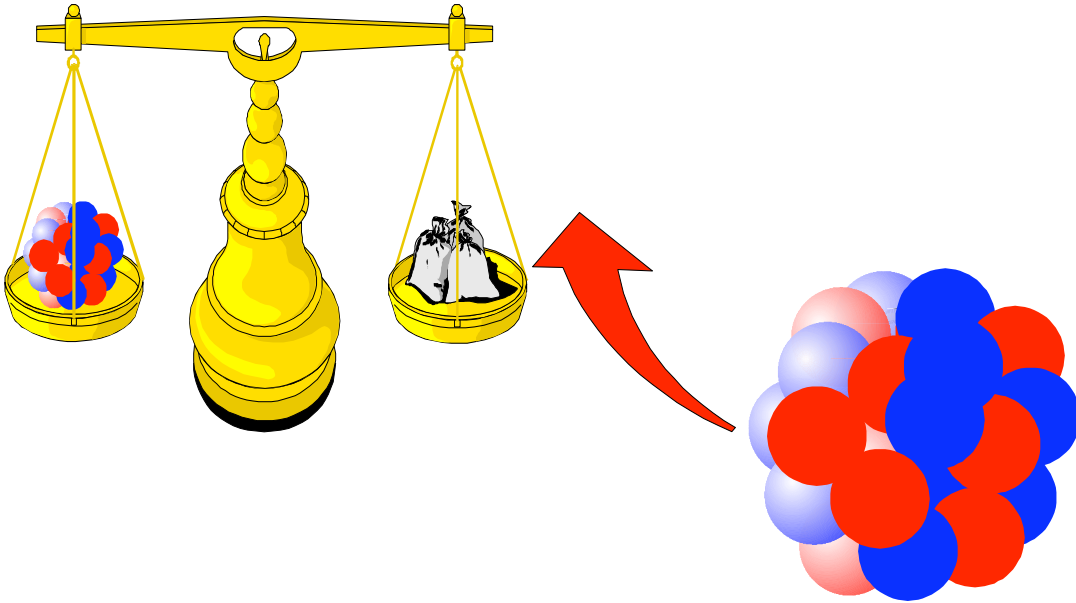
Estado normal da matéria nuclear

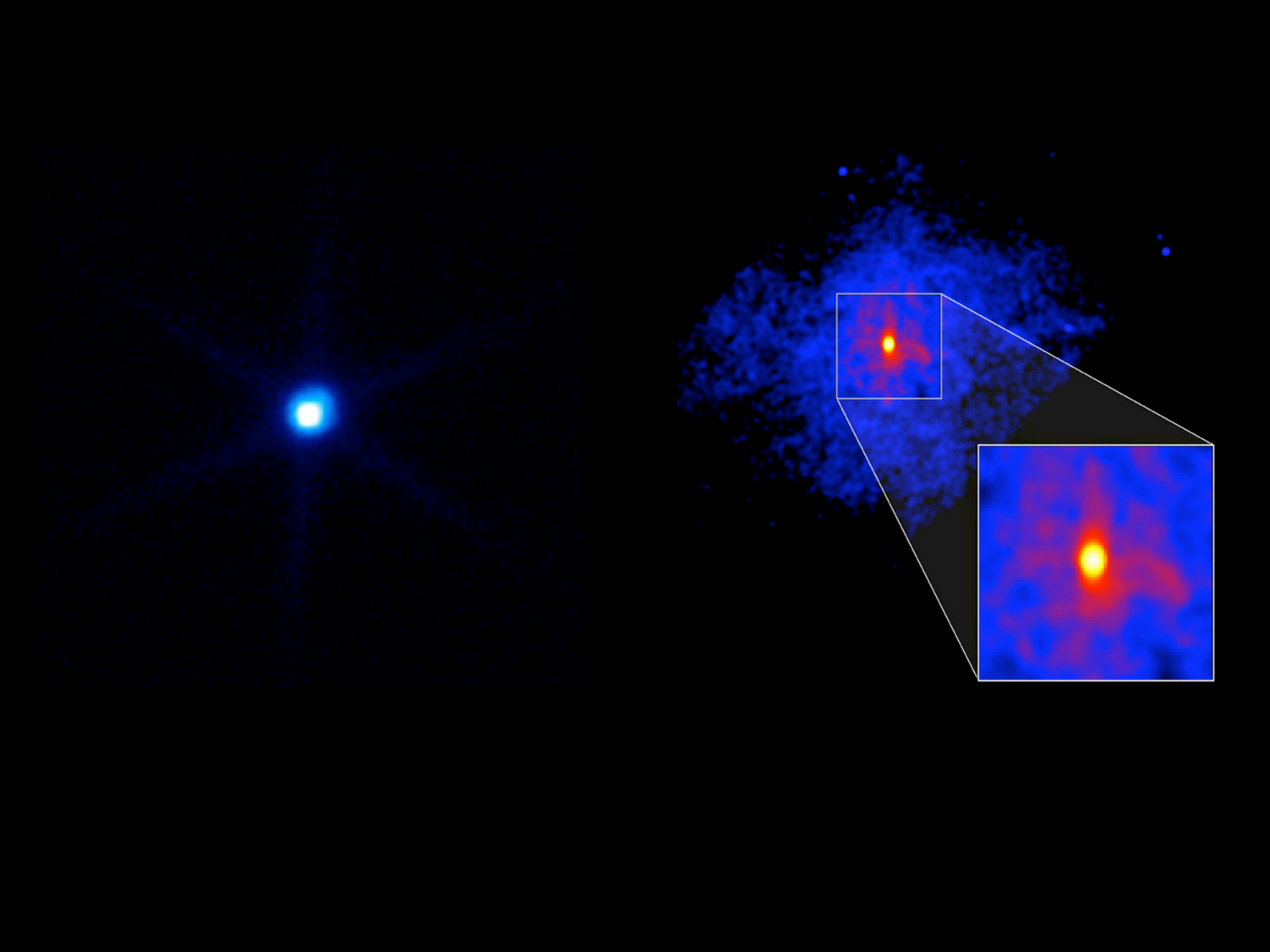
densidade ($\times 10^{14} \text{ kg/m}^3$)

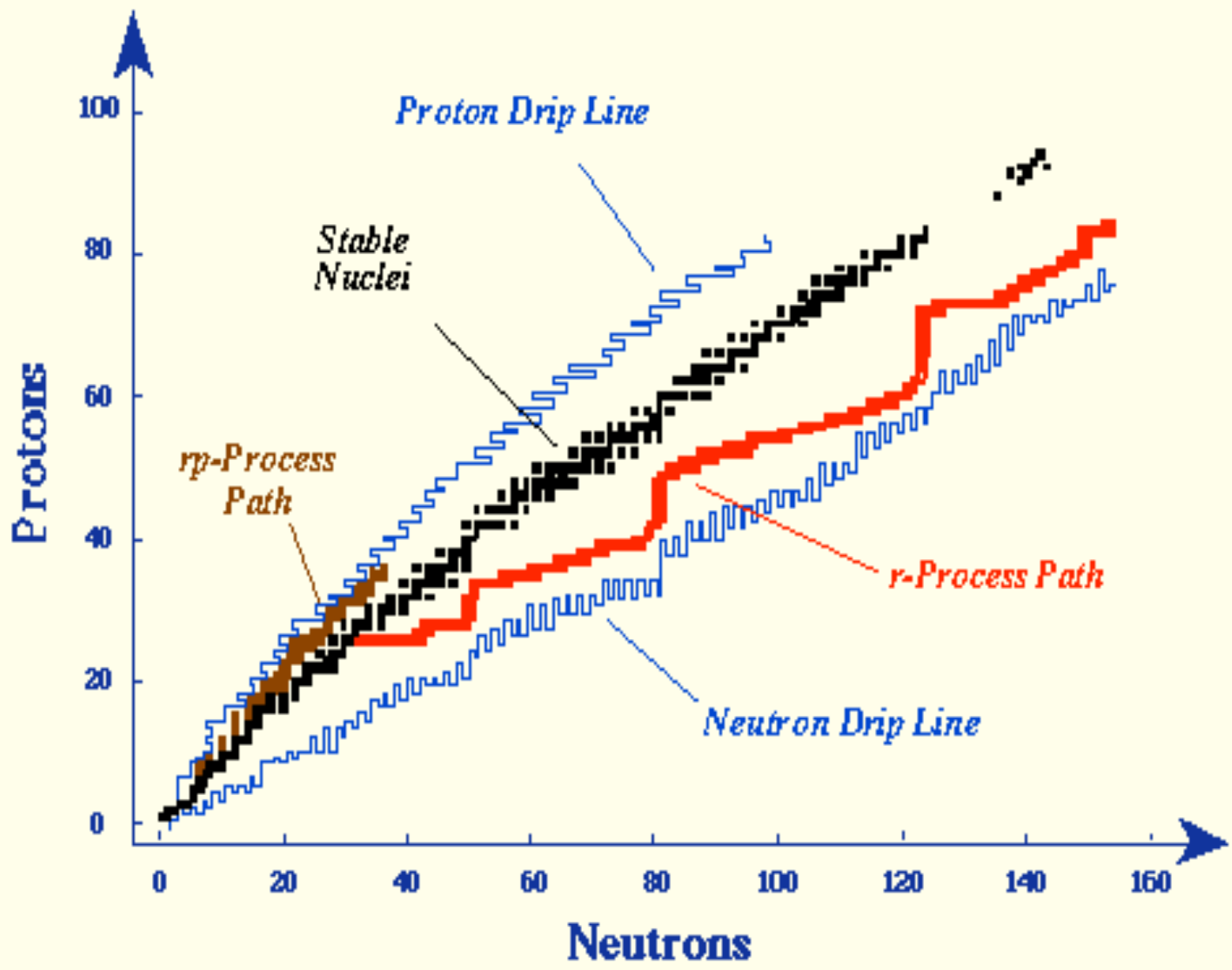
Estrelas de Neutrons



$$\rho \sim 10^{18} \text{ Kg/m}^3$$







COLISÃO DISTANTE



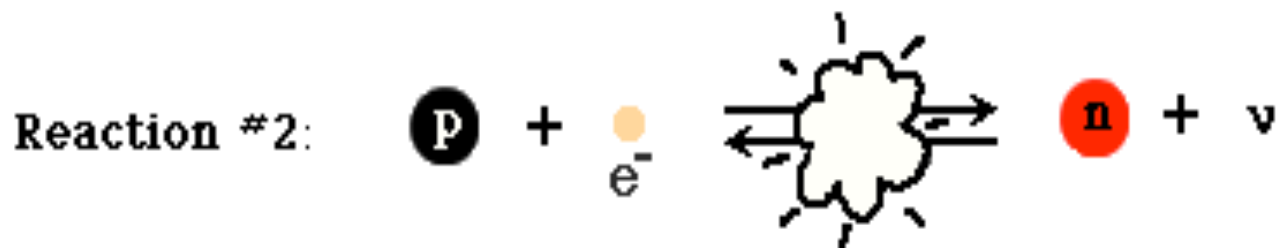
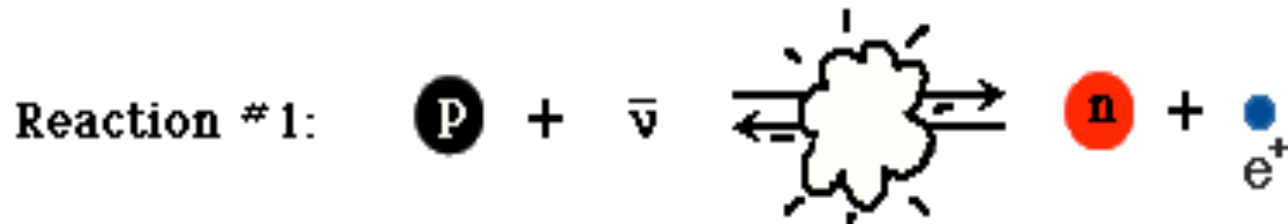
COLISÃO RASANTE



COLISÃO FRONTAL



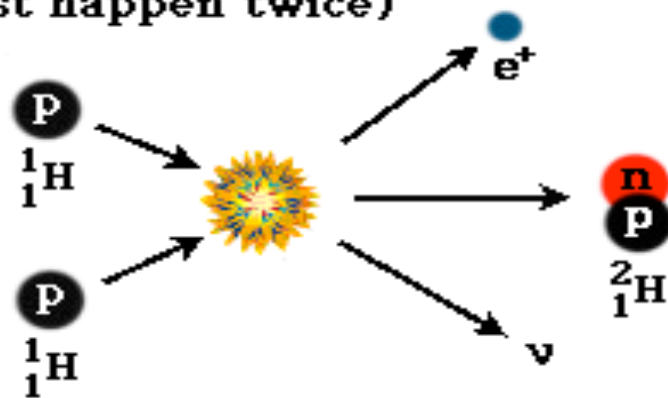
proton/neutron conversions



(The double arrows indicate these reactions go both ways.)

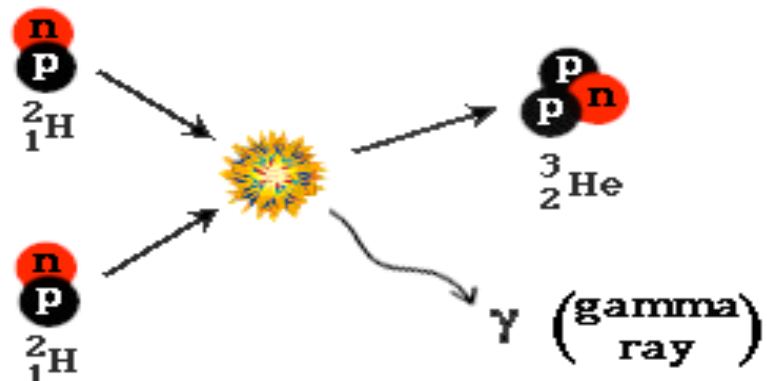
Step 1

(must happen twice)

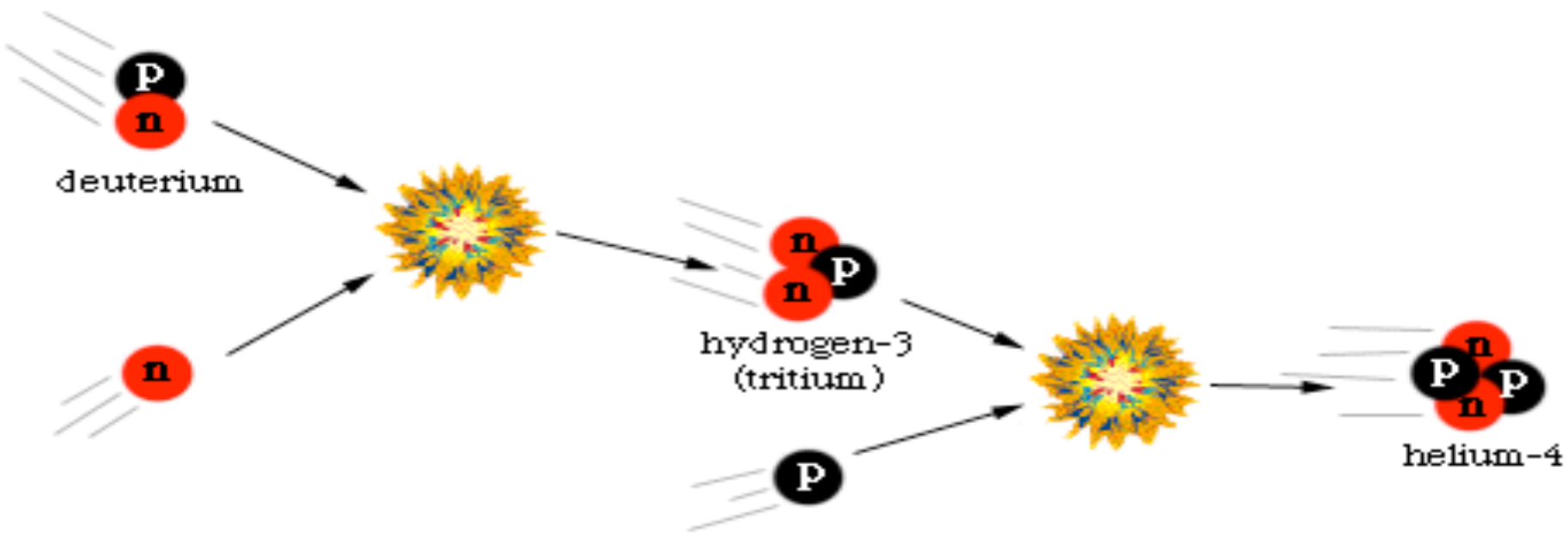
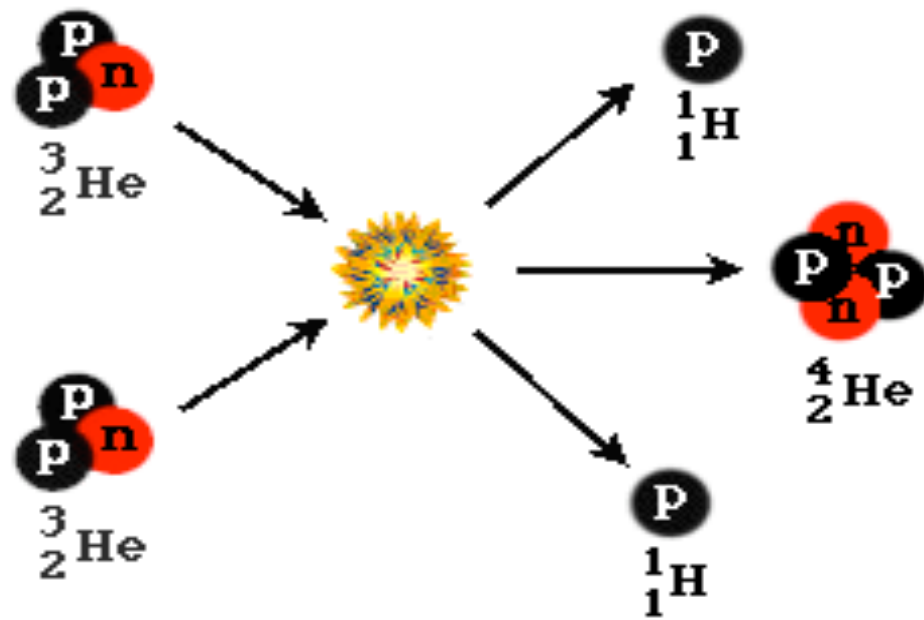


Step 2

(must happen twice)



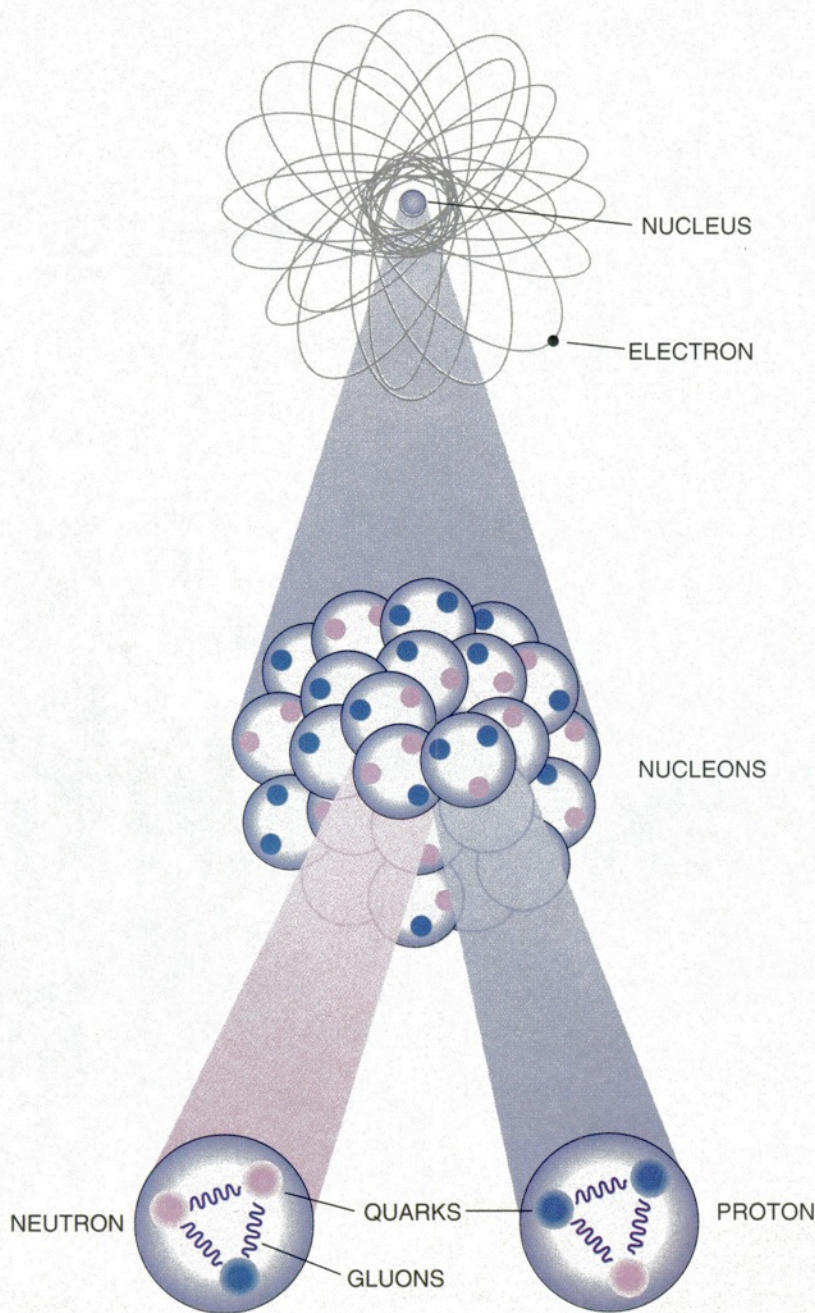
Step 3:



$p+p+n+n \equiv \text{ALPHA}$
 $d+t \equiv n + \text{ALPHA}$



$n = 1.00866$ u.m.a.
 $p = 1.0079$ u.m.a.
 $d = 2.01410$ u.m.a.
 $t = 3.01860$ u.m.a.
 ${}^4\text{He} = 4.00260$ u.m.a.
 ${}^6\text{Li} = 6.01512$ u.m.a.
 ${}^{12}\text{C} = 0.00000$ u.m.a.



$$M_{(\text{proton})} = M_p = 938.27 \text{ MeV}$$

$$M_{(\text{neutron})} = M_n = 939.56 \text{ MeV}$$

$$M_{(\text{eletron})} = M_e = 0.511 \text{ MeV}$$

$$M_{(\text{átomo de Hidrogênio})} = M_H = 938.58 \text{ MeV}$$

$$M_p + M_e = 938.27 + 0.511 = 938.78 \text{ MeV}$$

$$M_H = 938.58 \text{ MeV}$$

$$M_{(16O)} = 14\,899.17 \text{ MeV}$$

$$8M_p + 8M_n = 15\,022.64 \text{ MeV}$$

$$\Delta M = 123.47 \text{ MeV}$$

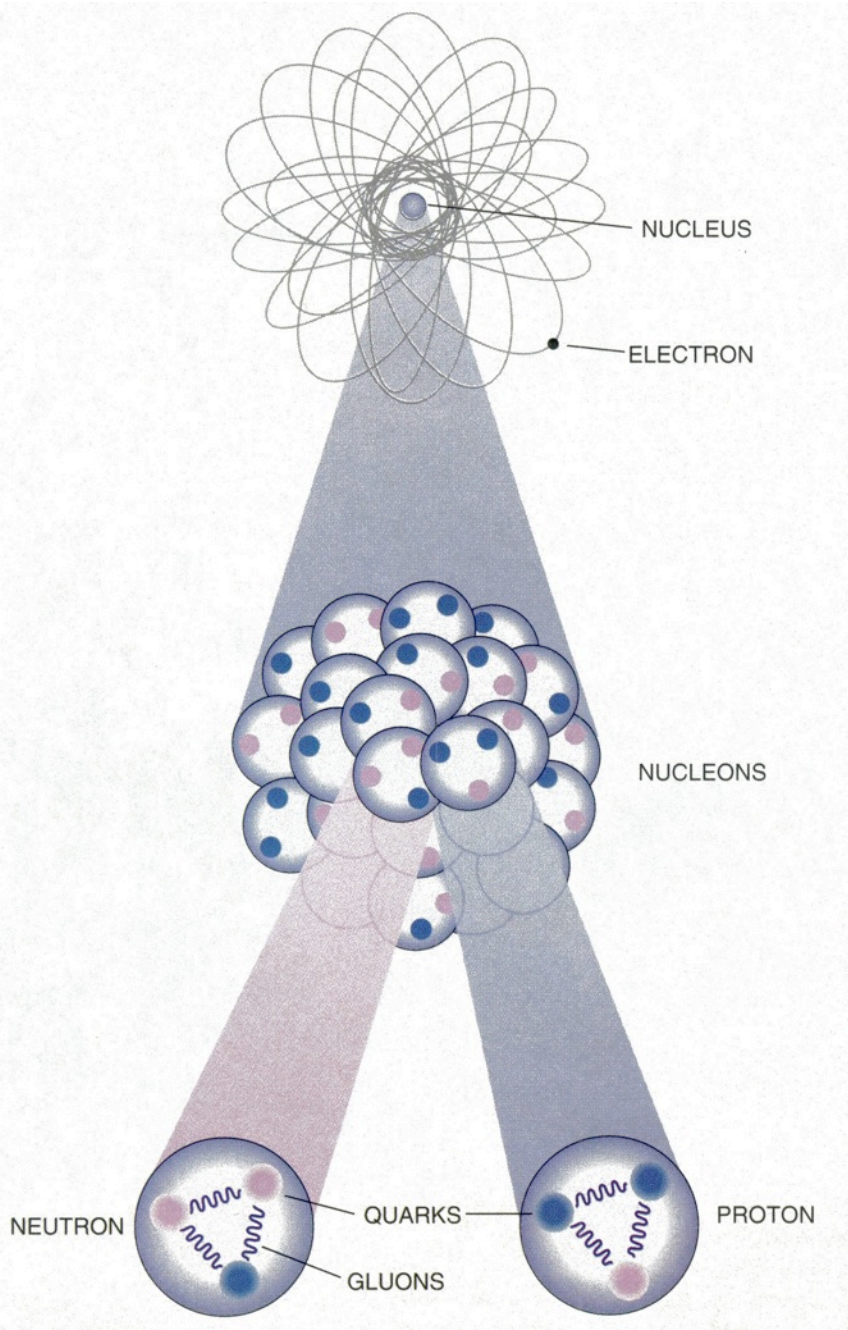
$$p = (u + u + d) \quad M_{(u)} \approx 3 \text{ MeV}$$

$$M_{(d)} \approx 6 \text{ MeV}$$

$$2 M(u) + M(d) \approx 12 \text{ MeV}$$

$$M_p = 938.27 \text{ MeV}$$

$$\Delta M \approx 926 \text{ MeV}$$



$$M_{(\text{neutron})} = M_n = 1.008664904 \text{ u}$$

$$M_{(\text{proton})} = M_p = 1.007276470 \text{ u}$$

$$M_{(\text{electron})} = M_e = 0.000548580 \text{ u}$$

$$M_{(\text{átomo de Hidrogênio})} = M_H = 1.007825032 \text{ u}$$

$$M_p + M_e = 1.007815050 \text{ u}$$

$$M_{(16O)} = 14\,899.17 \text{ MeV}$$

$$8M_p + 8M_n = 15\,022.64 \text{ MeV}$$

$$\Delta M = 123.47 \text{ MeV}$$

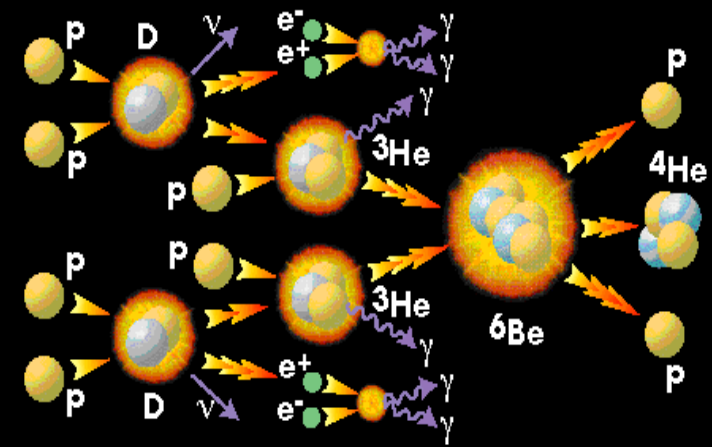
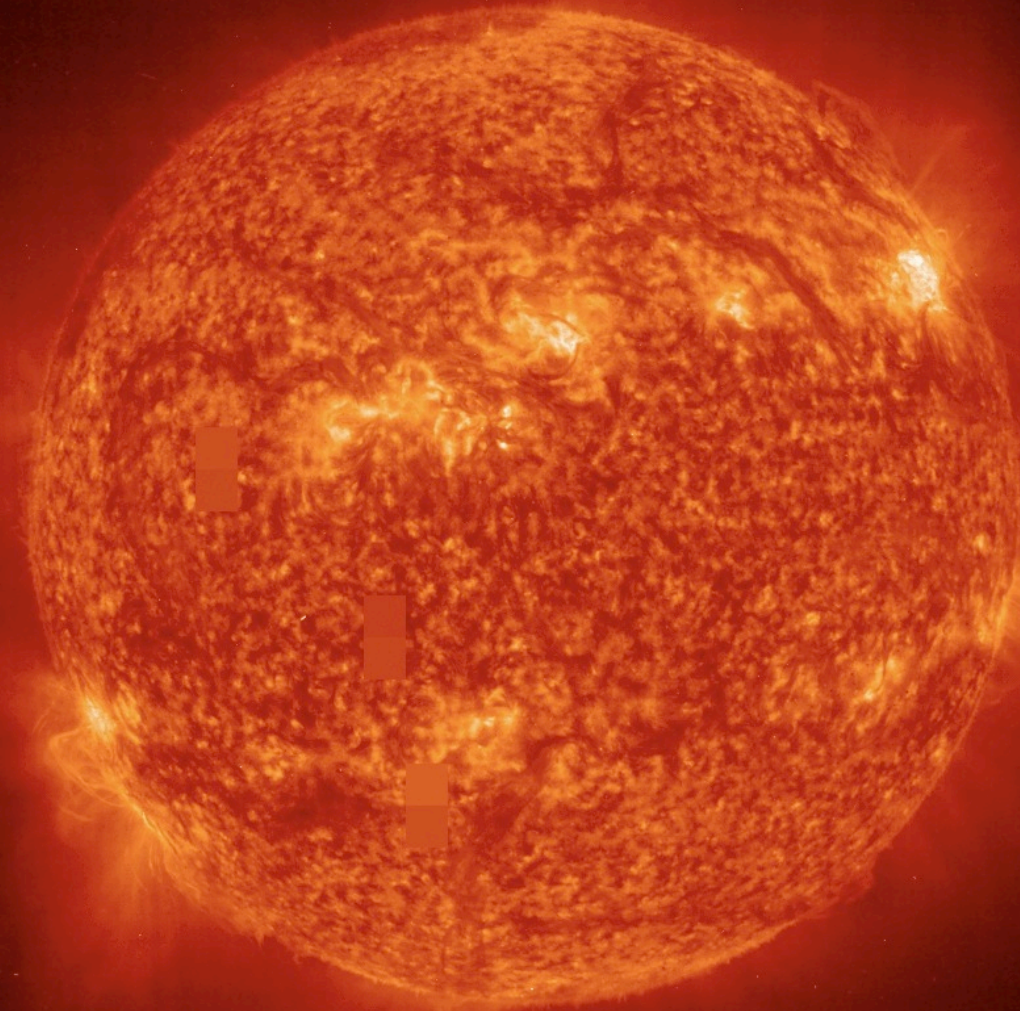
$$p = (u + u + d) \quad M_{(u)} \approx 3 \text{ MeV}$$

$$M_{(d)} \approx 6 \text{ MeV}$$

$$2 M(u) + M(d) \approx 12 \text{ MeV}$$

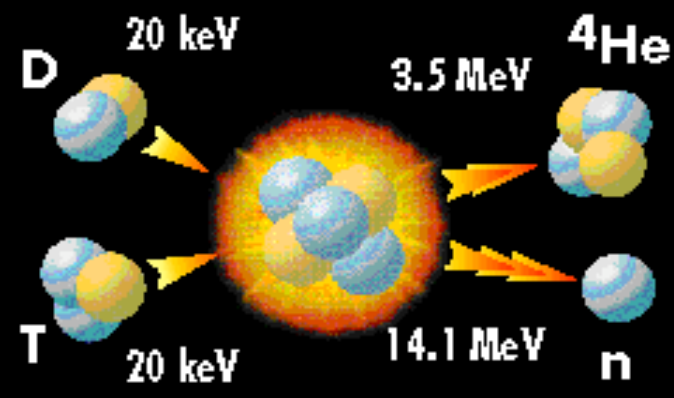
$$M_p = 938.27 \text{ MeV}$$

$$\Delta M \approx 926 \text{ MeV}$$

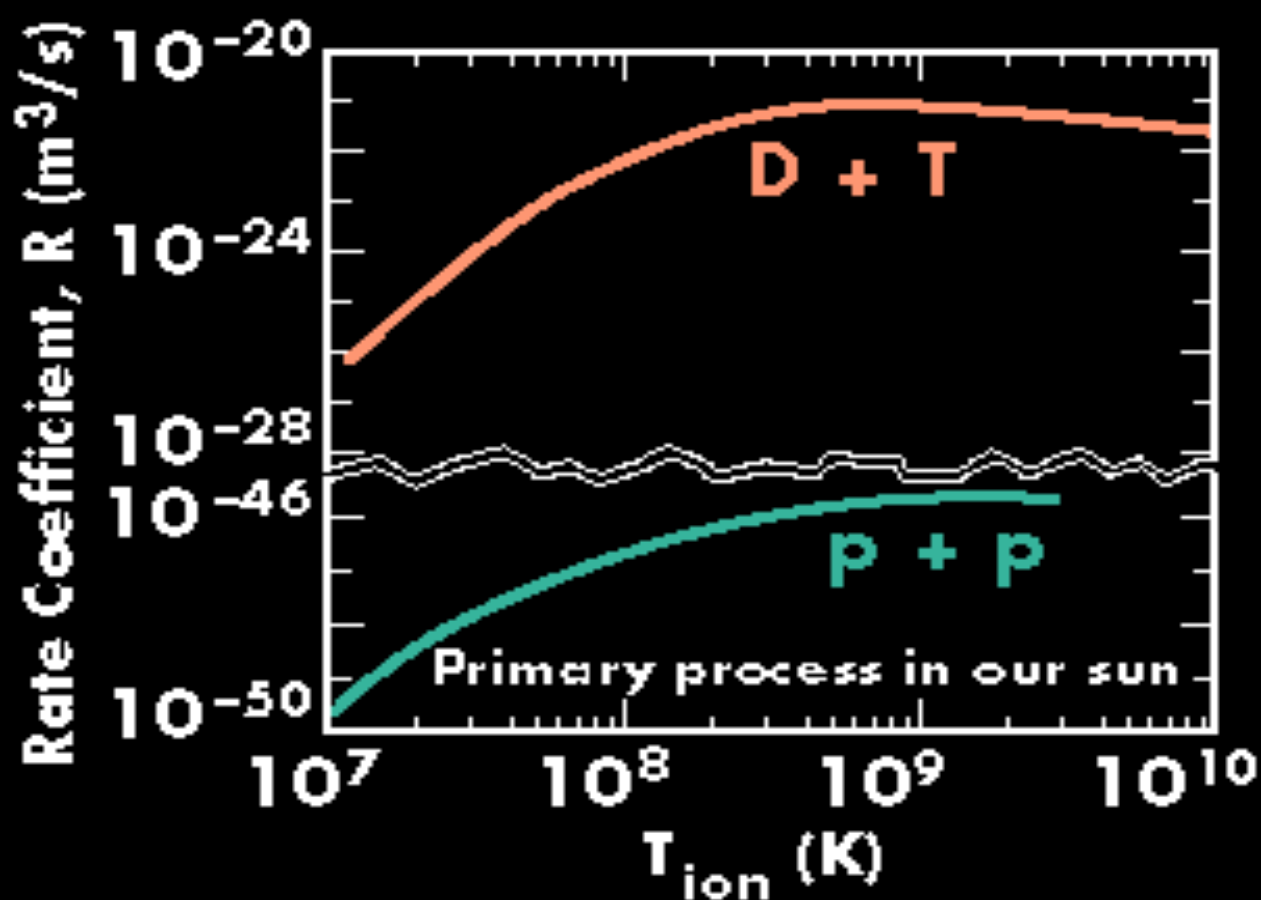


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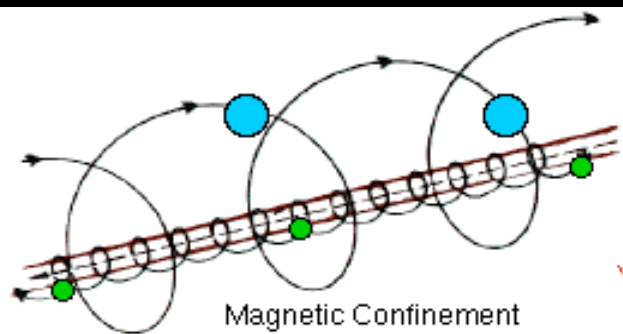
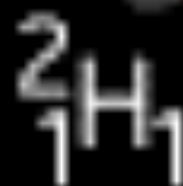
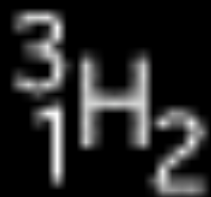
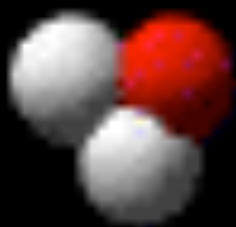
Reactants Fusion Products



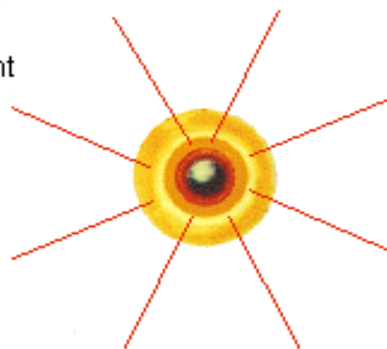
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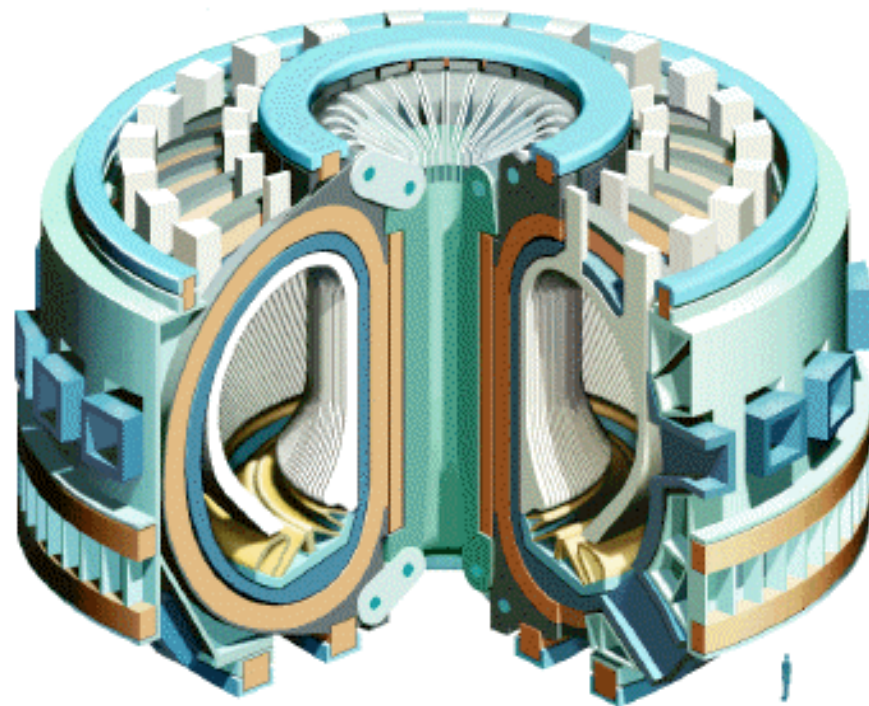
Copyright © 1996 Contemporary Physics Education Project.

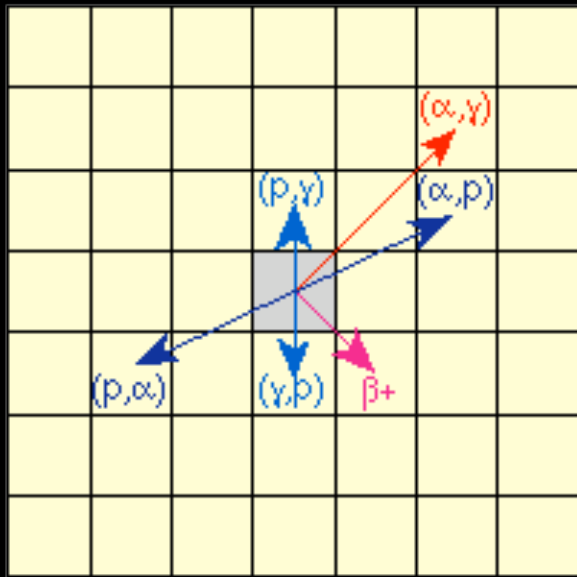


Gravitational Confinement
in the Sun and Stars

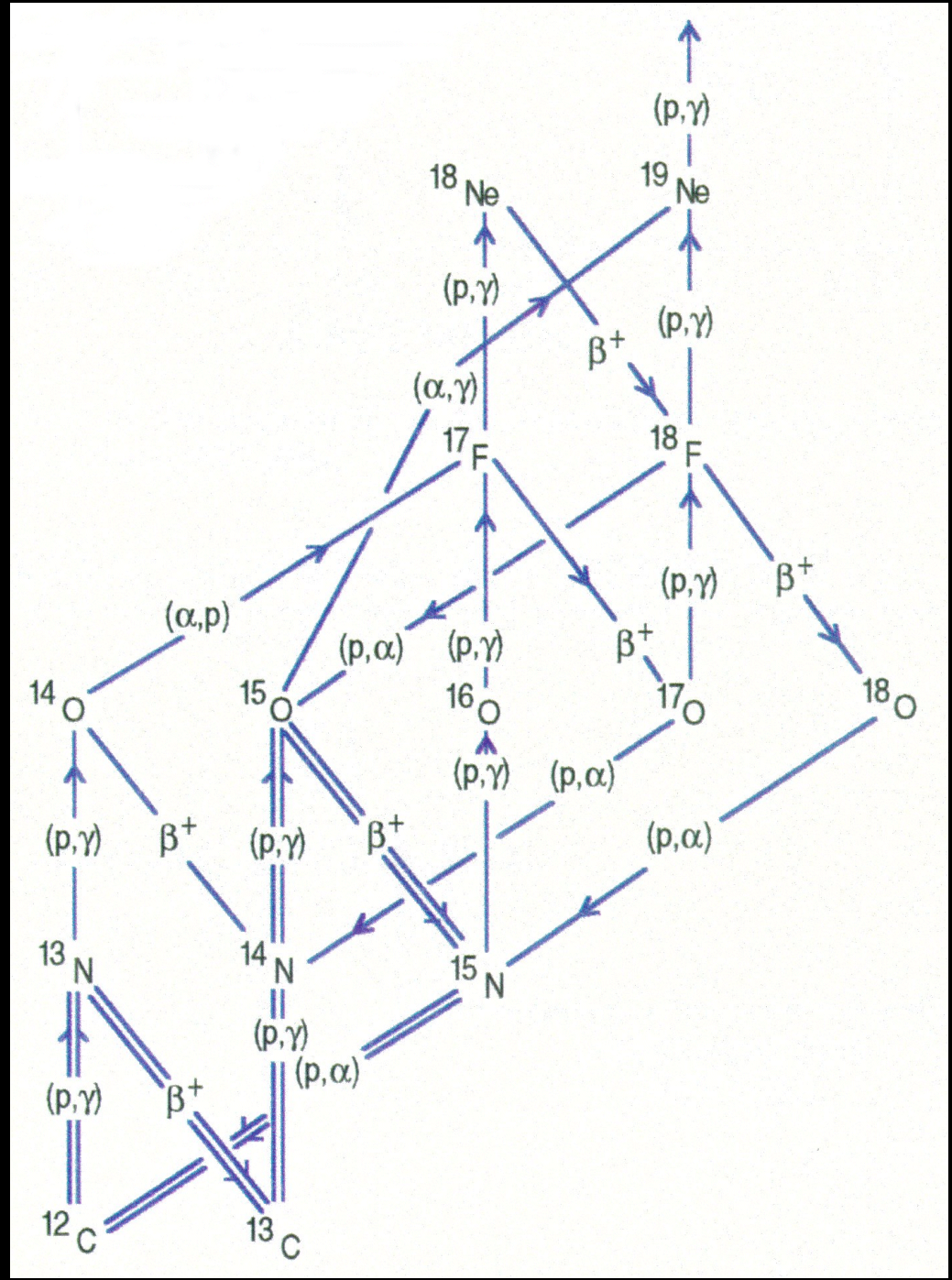
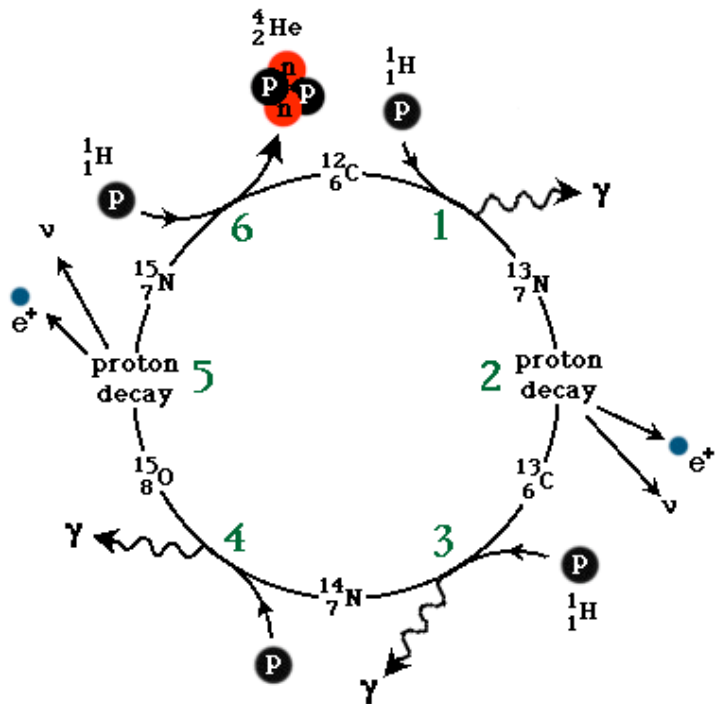


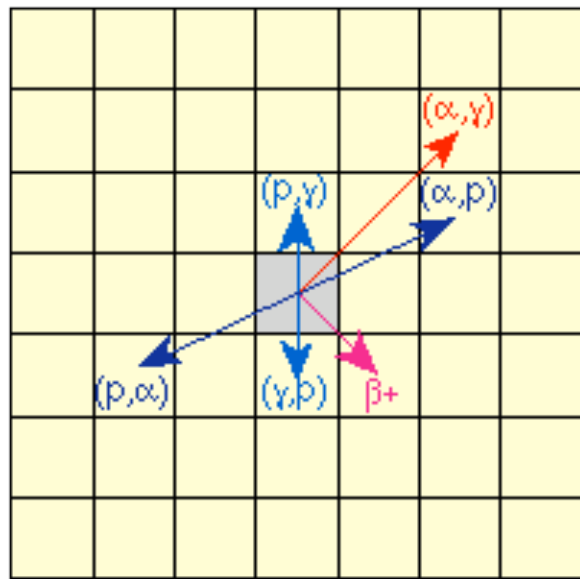
Inertial Confinement
Using Lasers



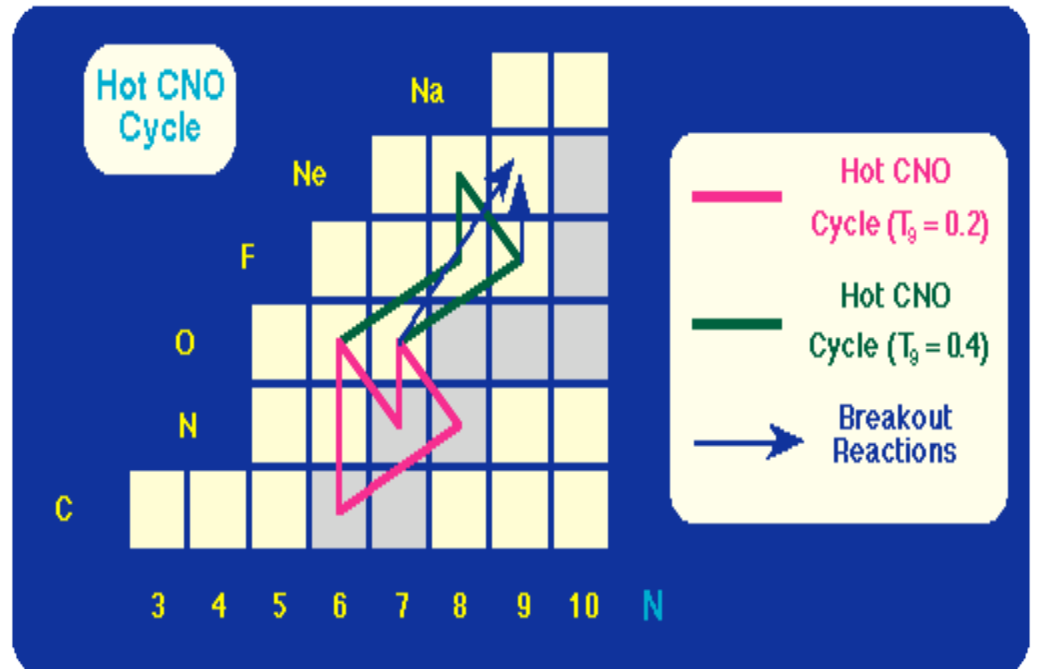
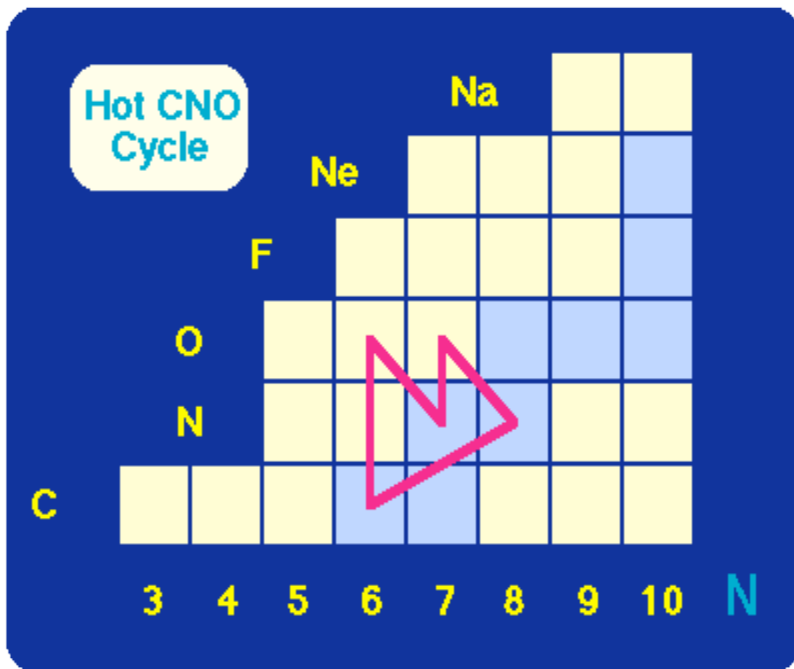
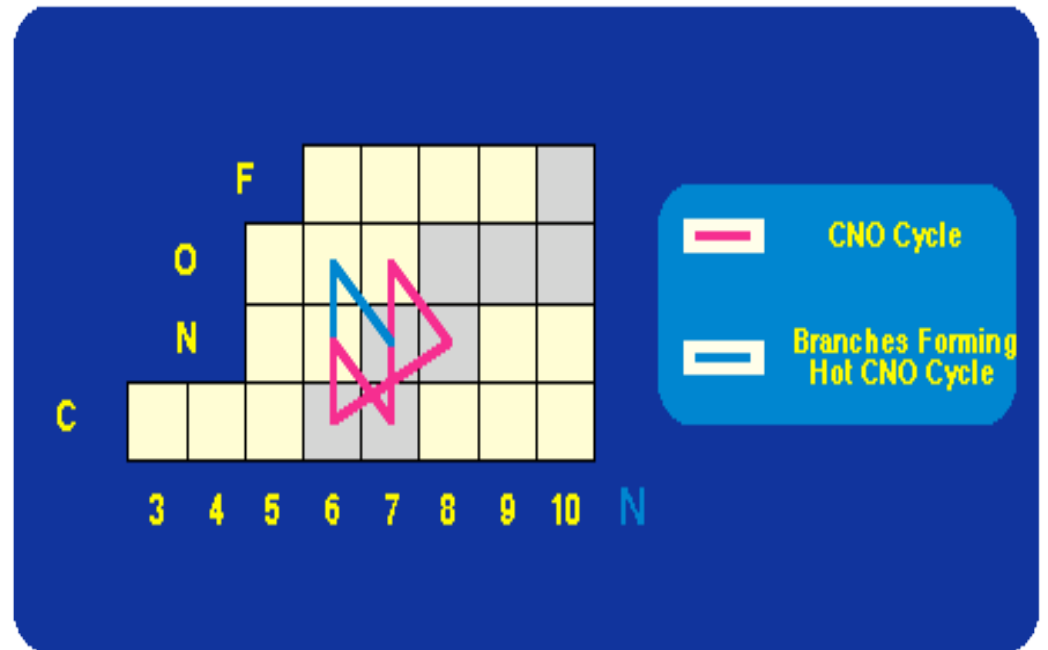


Carbon-Nitrogen-Oxygen Cycle

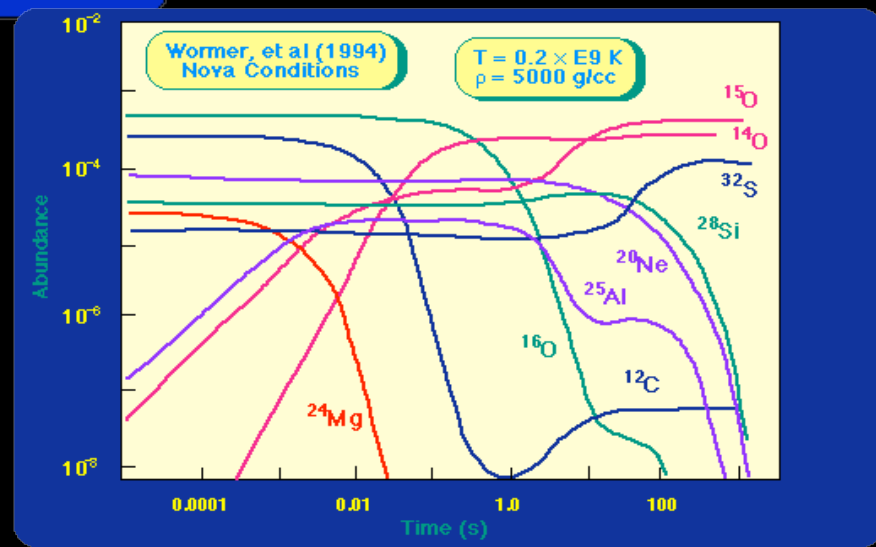
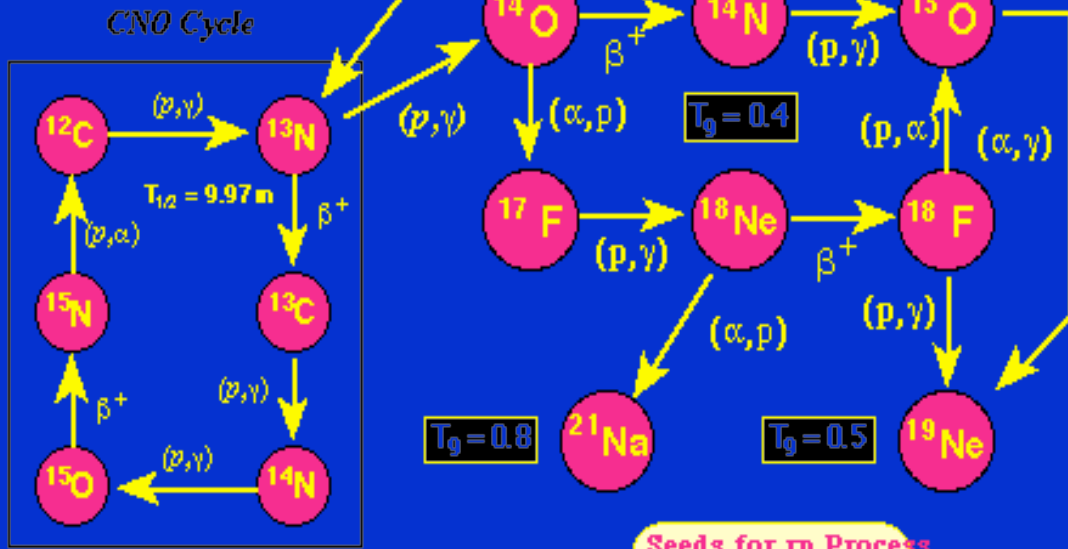


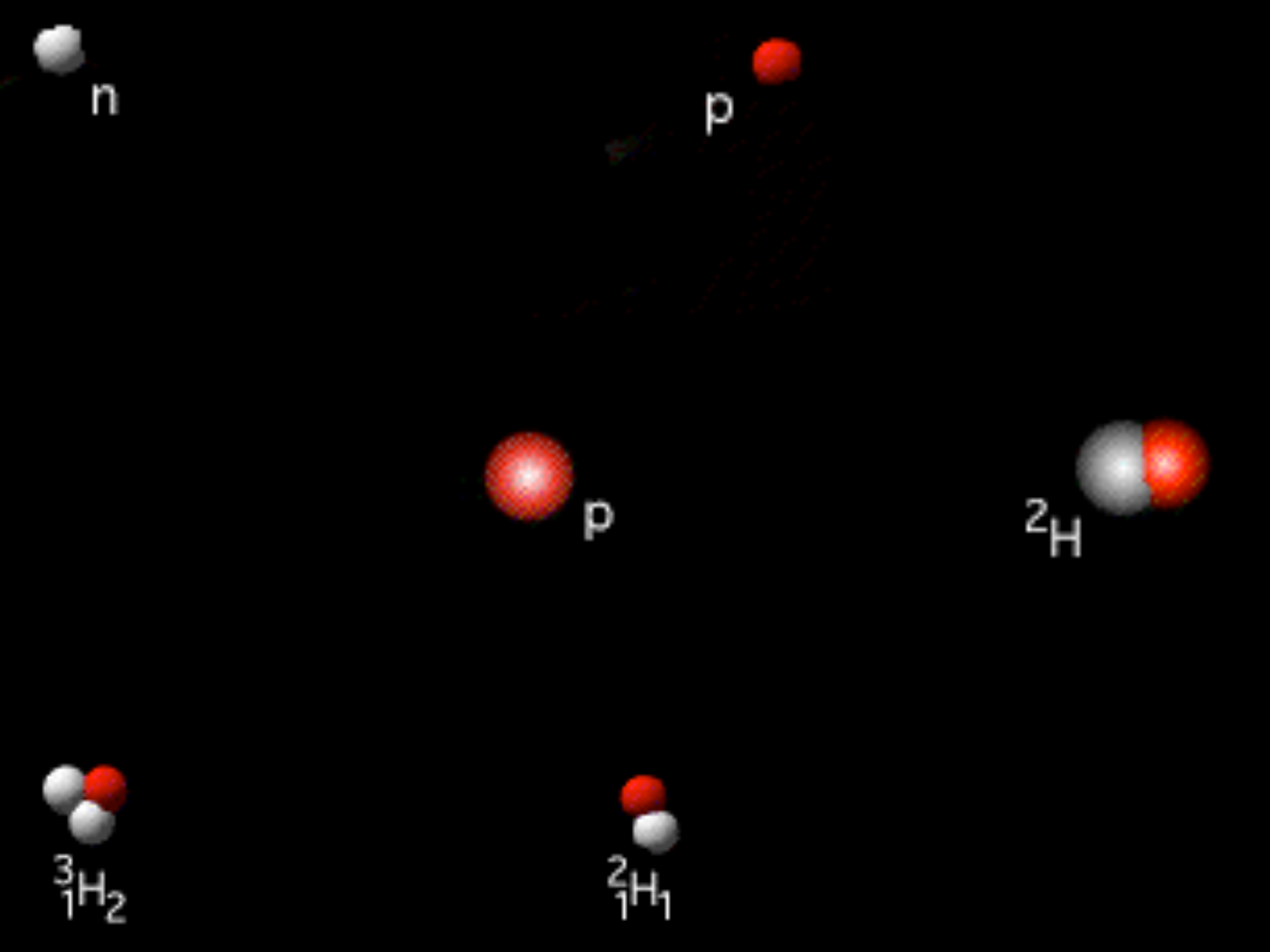


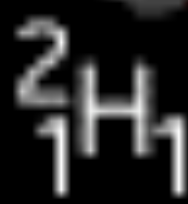
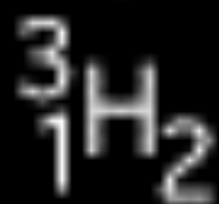
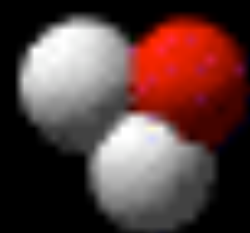
Reaction Vectors



The Hot CNO Cycle



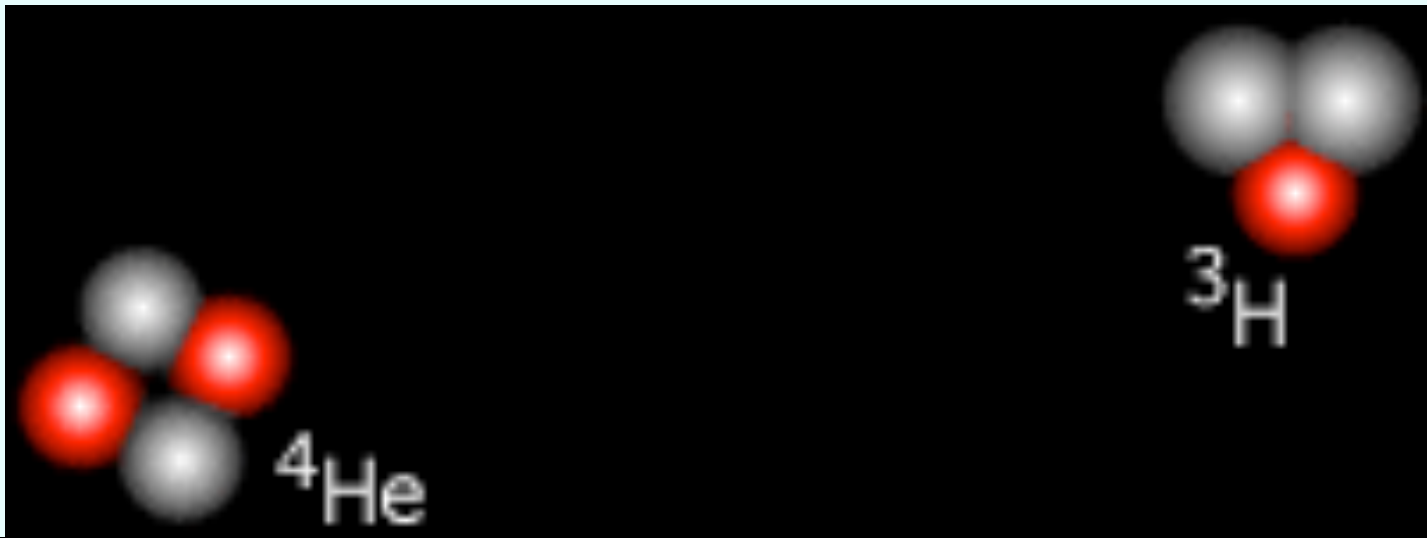


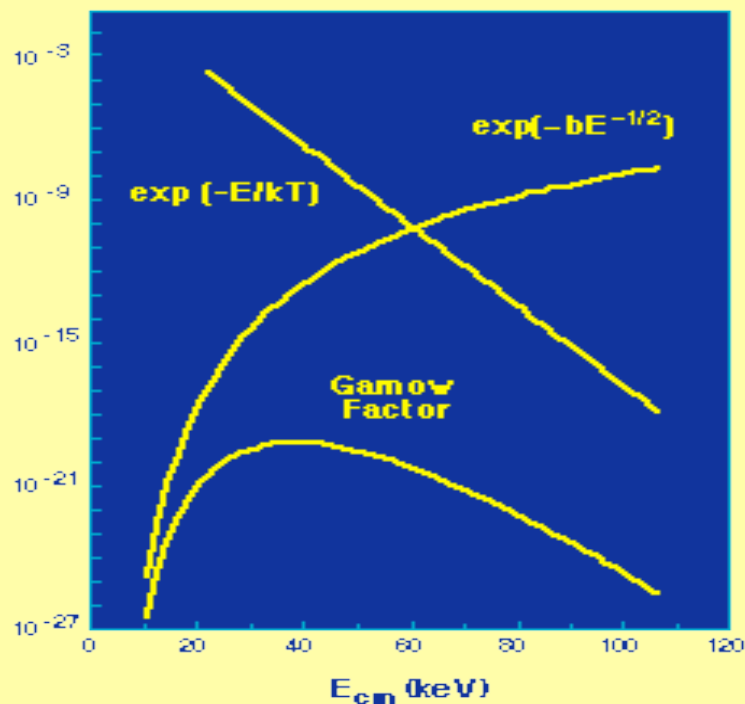


ESPALHAMENTO



TUNELAMENTO





Gamow Factor

For charged-particle reactions there is an optimum energy for reactions that represents a tradeoff between a Coulomb barrier penetration factor that increases exponentially with the square root of energy, and the Boltzmann factor that decreases exponentially with energy. The maximum of the product of these two functions is termed the *Gamow Peak*.

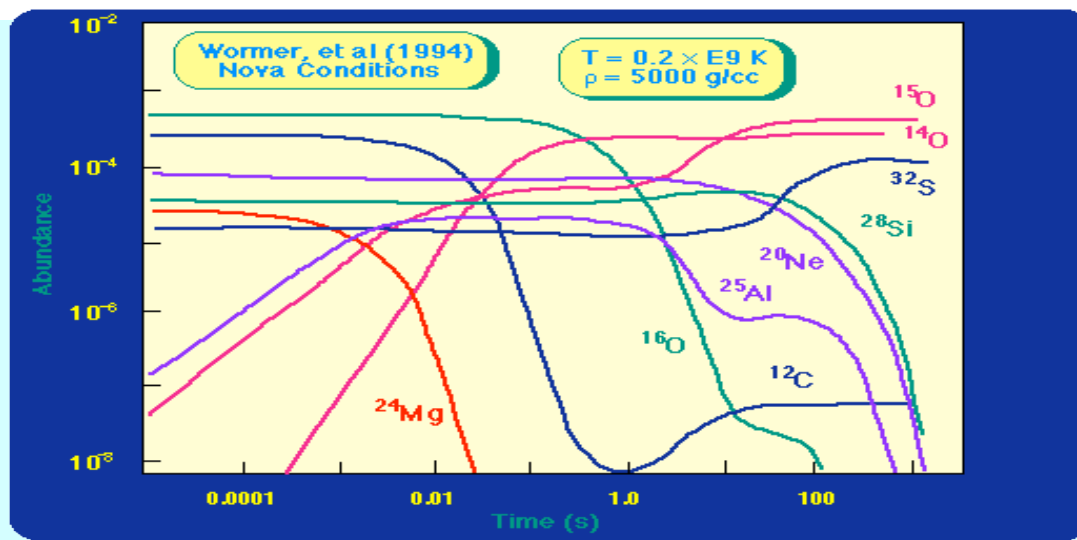
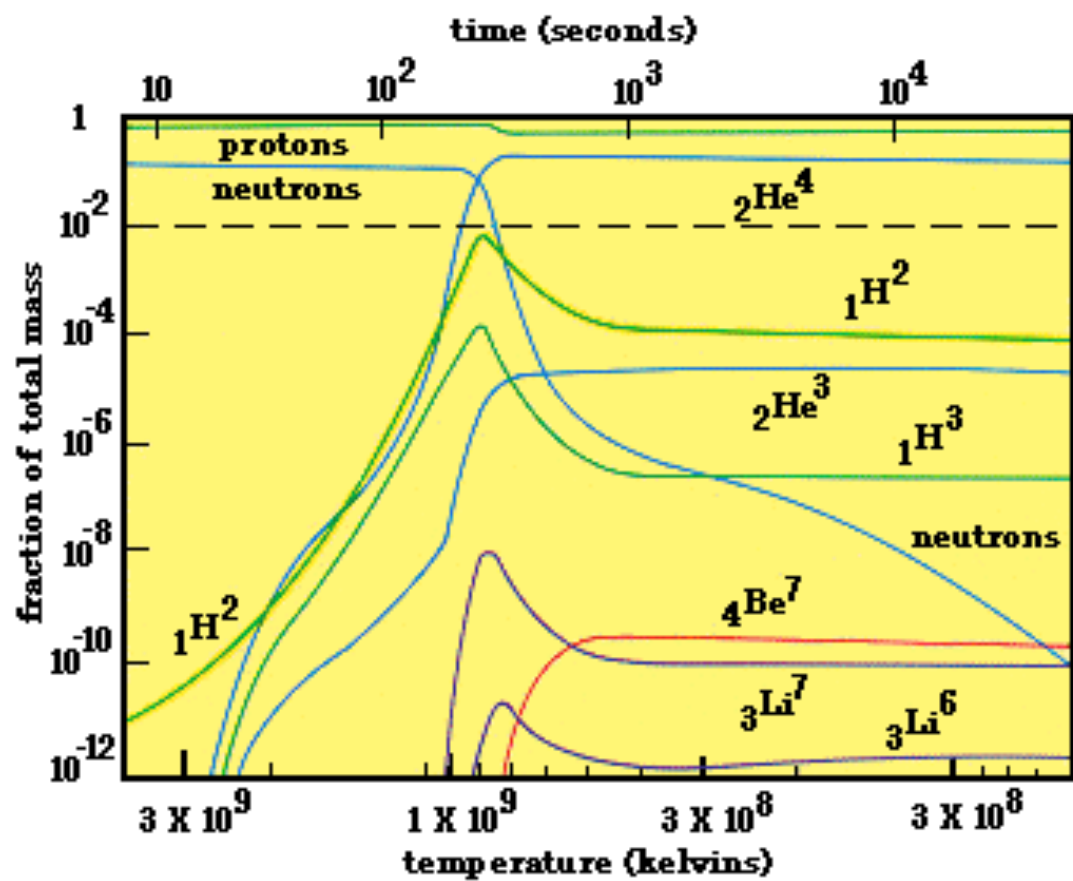
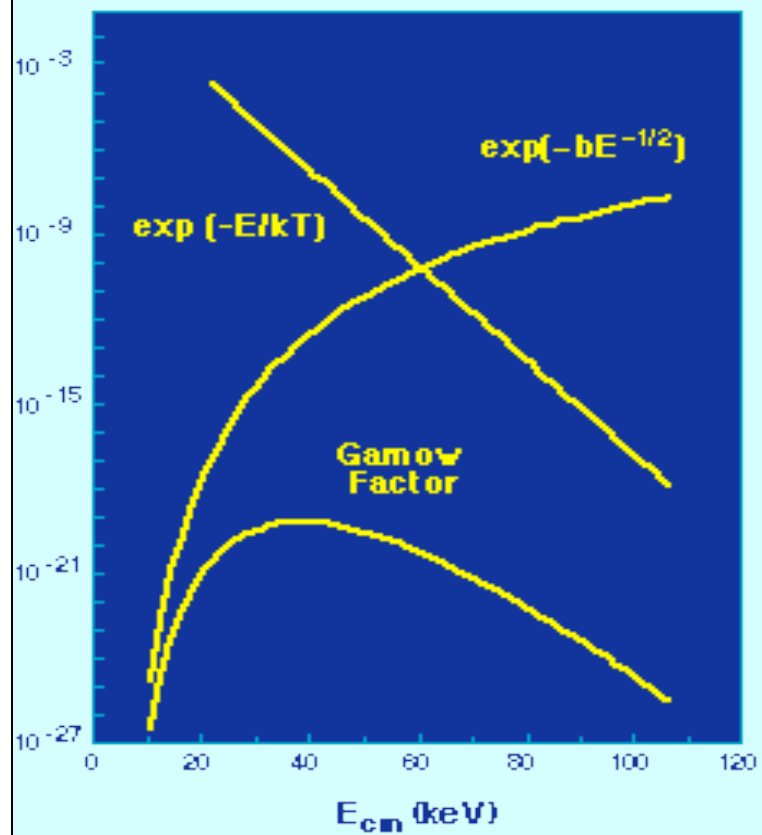
Element Production Networks

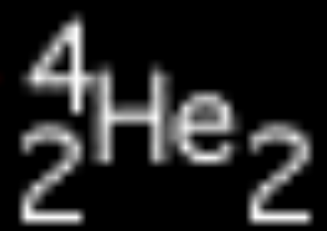
Cross Sections:
$$\sigma = \frac{\text{Number of Reactions/Target/ Second}}{\text{Incoming Flux}} = \frac{r/n_j}{n_k v}$$

Reaction Rates:
$$\langle j, k \rangle \equiv \langle \sigma v \rangle_{j, k} = \left(\frac{8}{\mu \pi} \right)^{1/2} (kT)^{-3/2} \int_0^{\infty} E \sigma(E) e^{-E/kT} dE$$

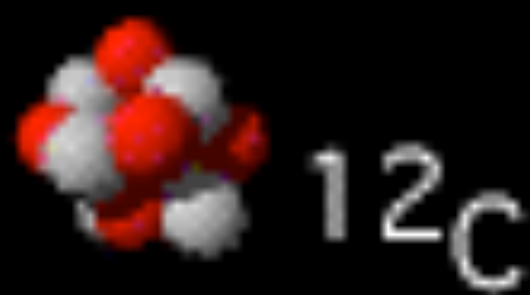
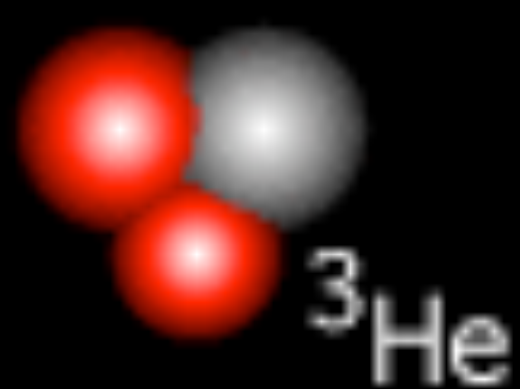
Differential Equations for Nuclear Abundances Y :

$$\dot{Y} = \sum_j c_j^i Y_j + \sum_{j, k} c_{jk}^i Y_j Y_k + \sum_{j, k, \ell} c_{j k \ell}^i Y_j Y_k Y_\ell$$

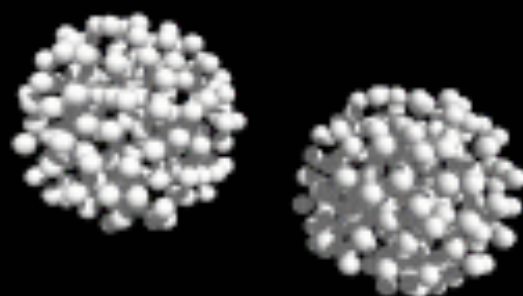




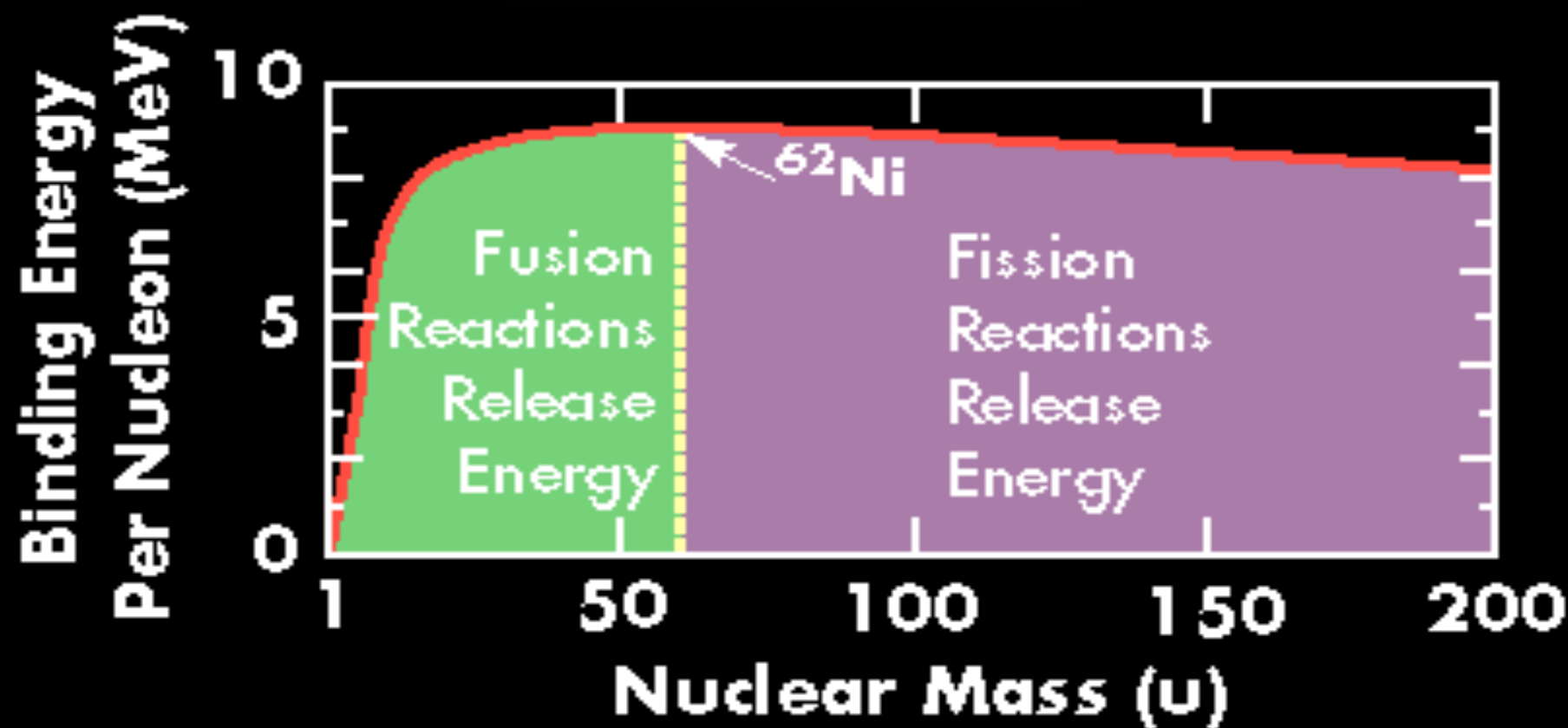


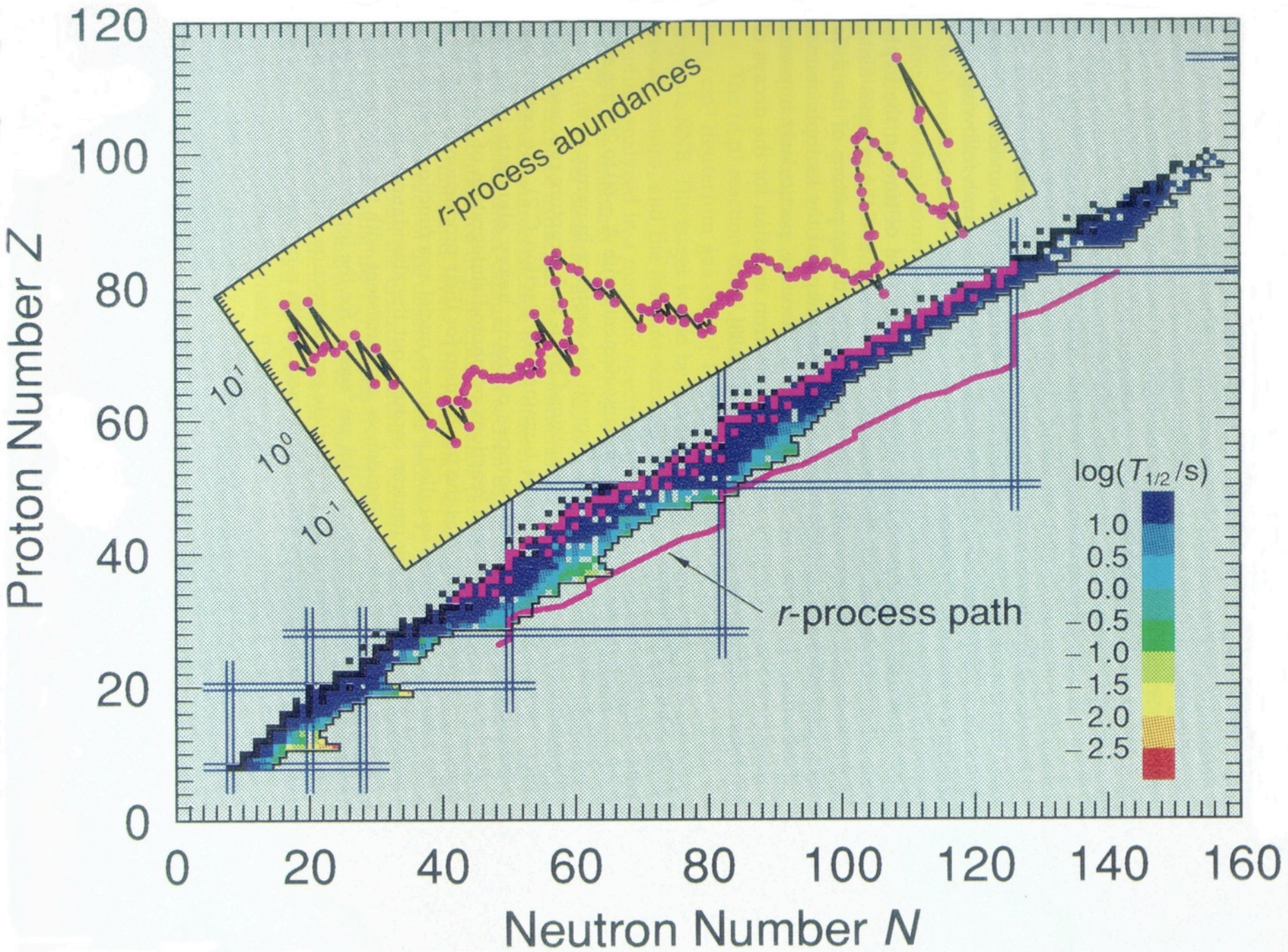


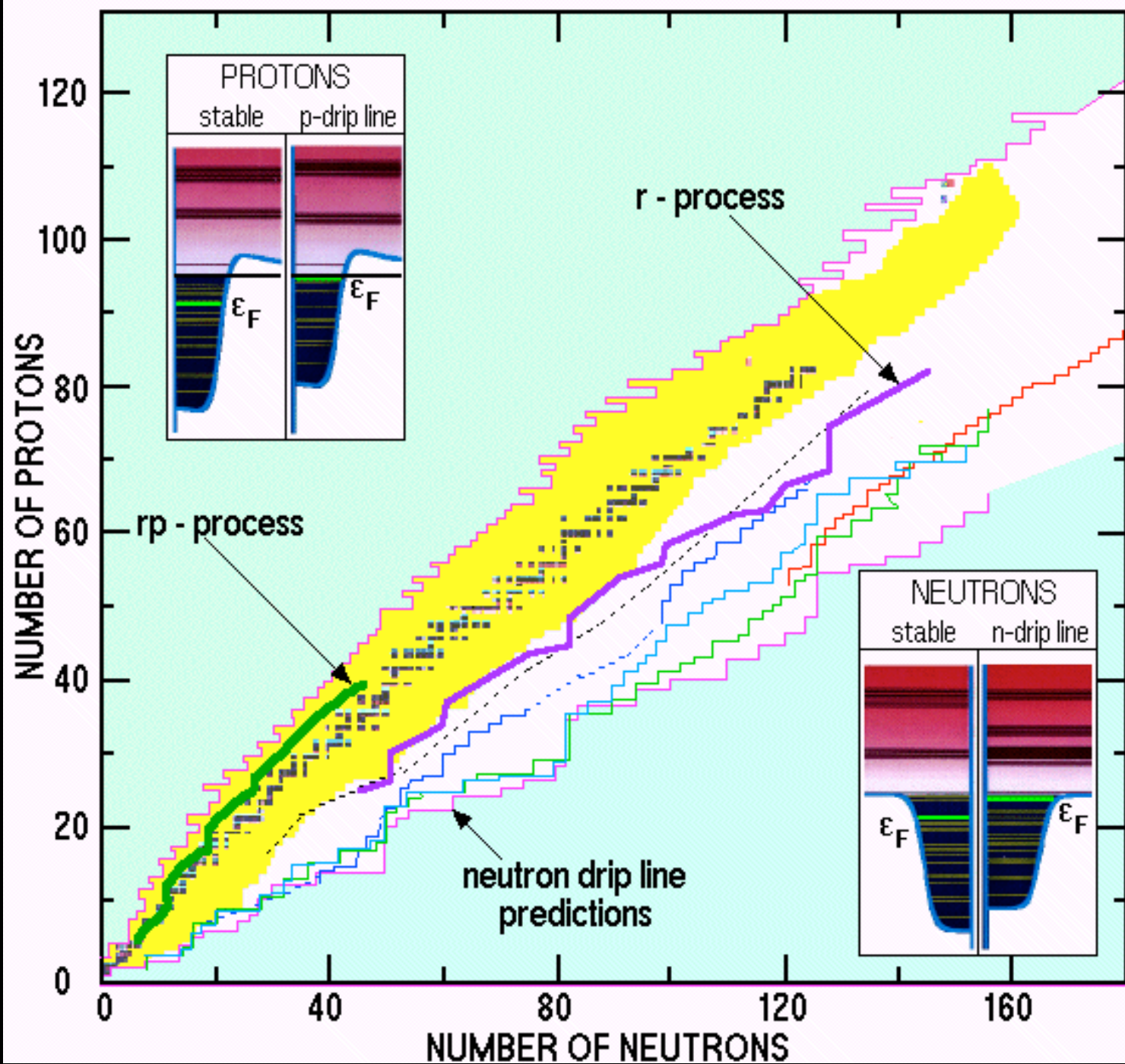


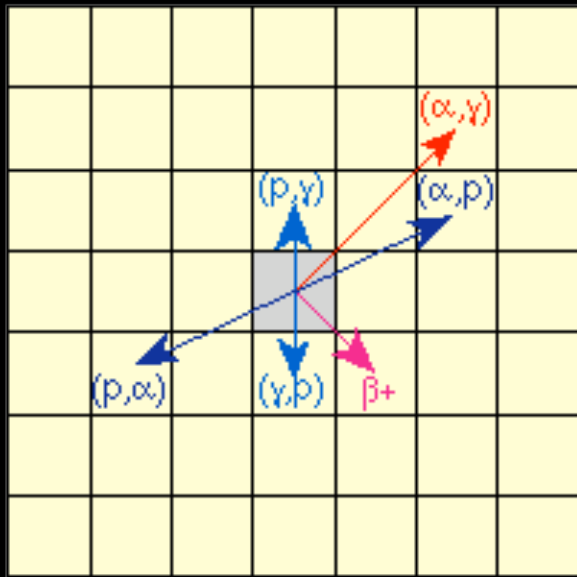


©2000 Prentice Hall

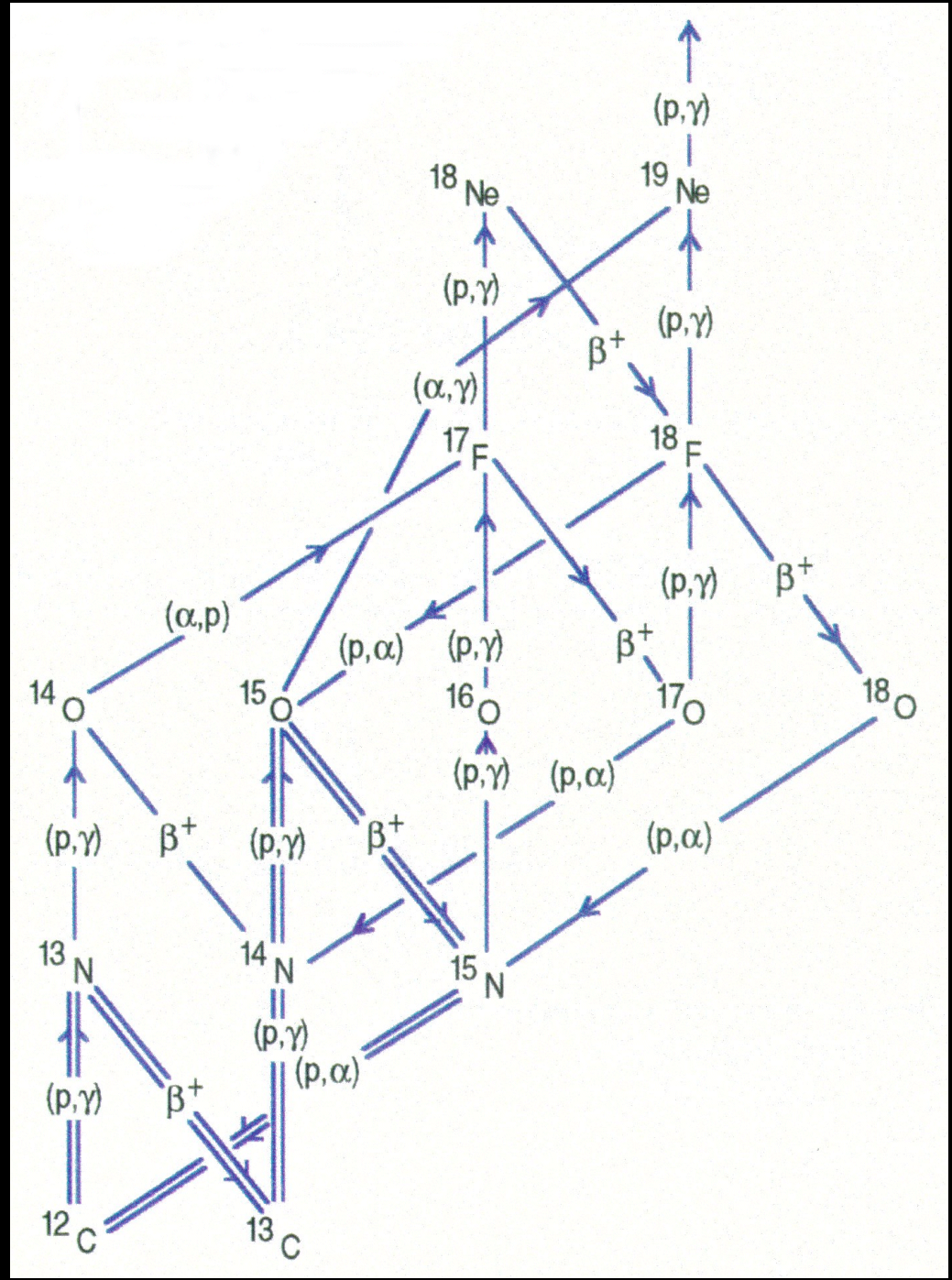
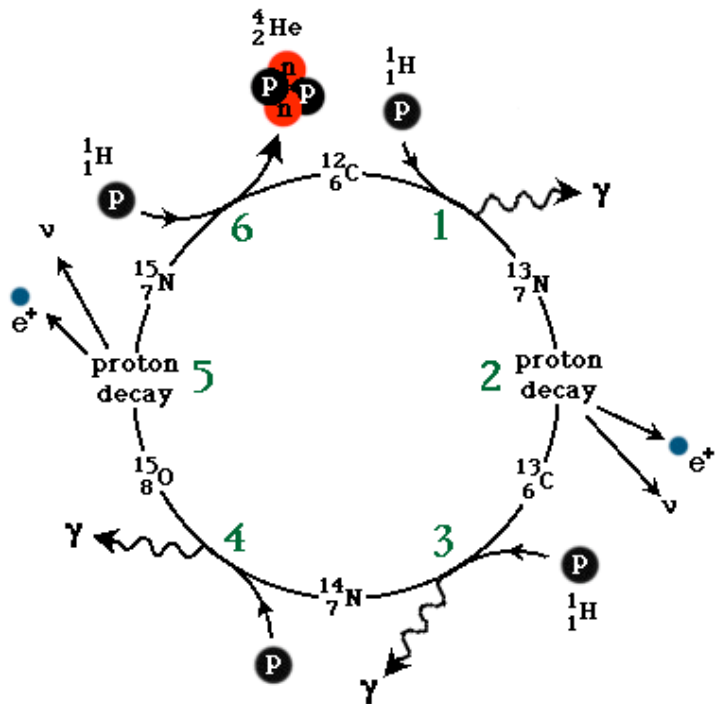


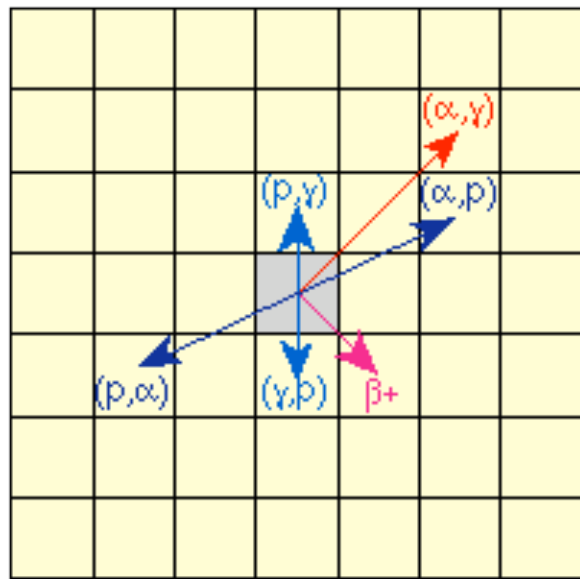




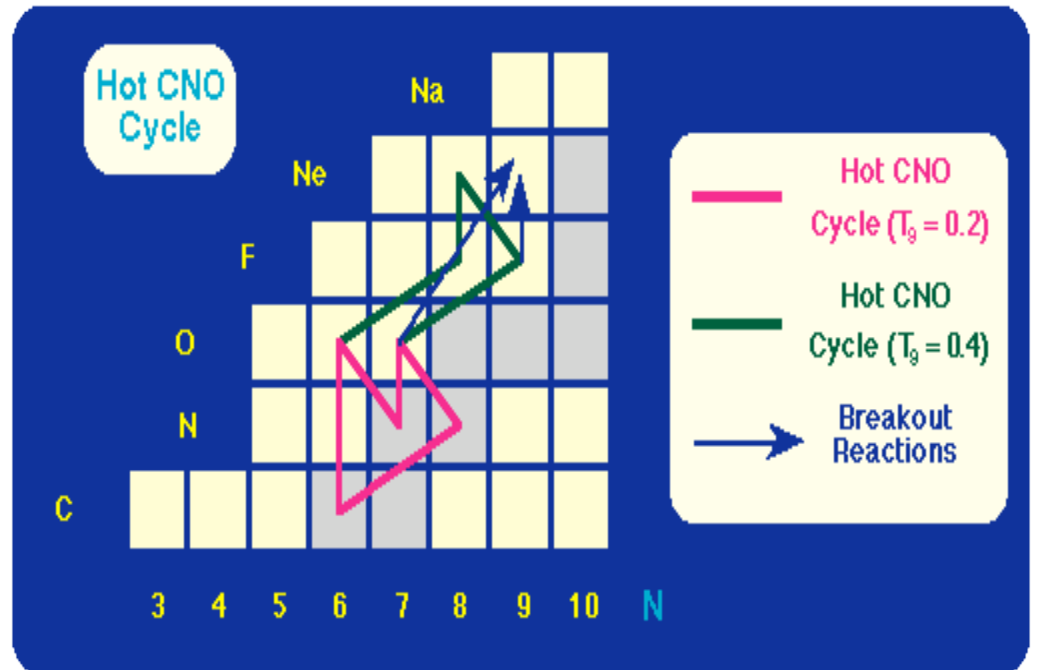
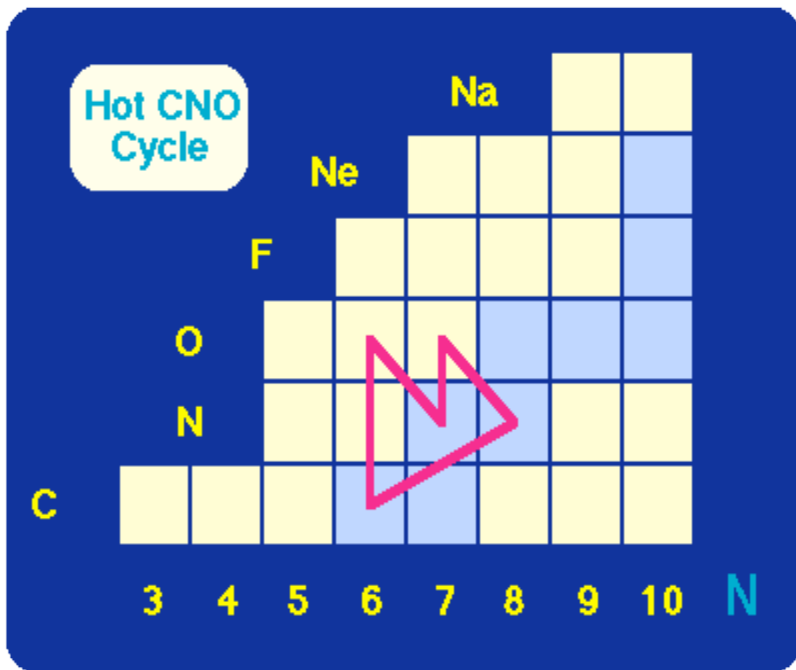
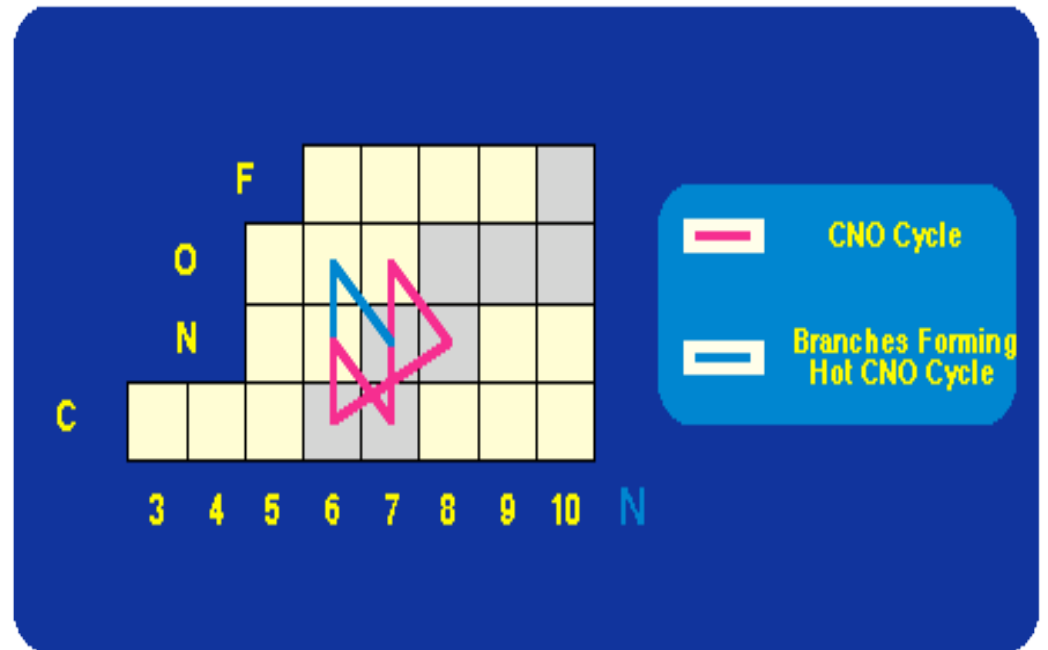


Carbon-Nitrogen-Oxygen Cycle

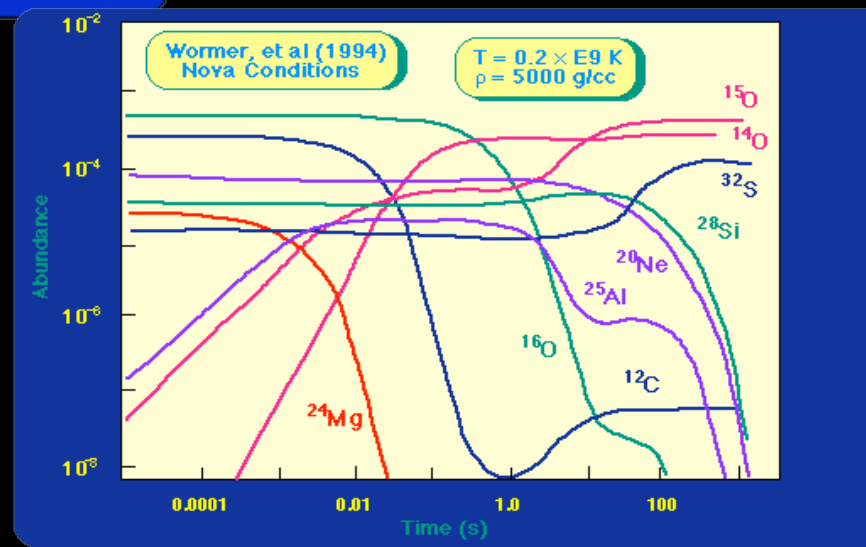
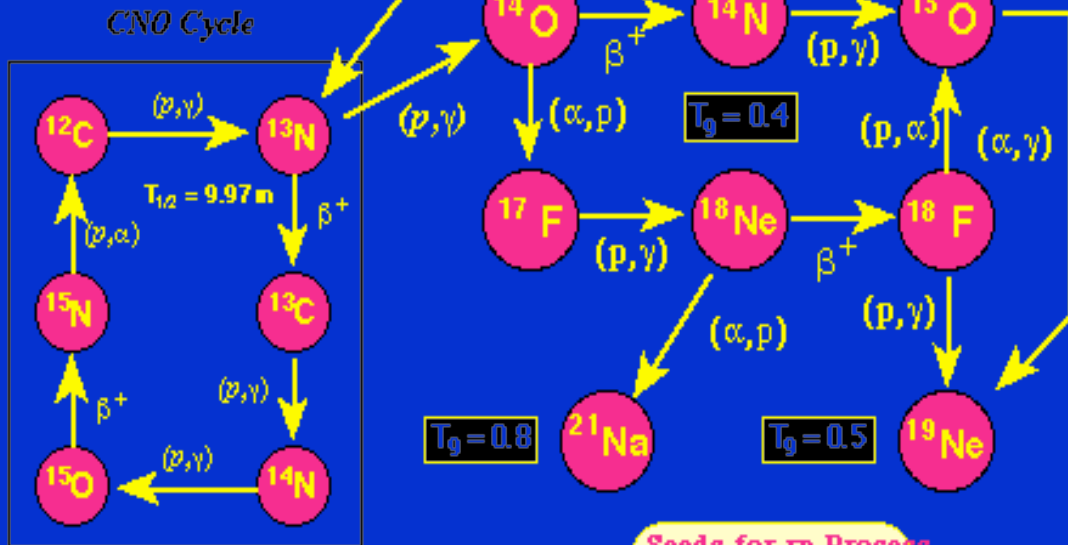




Reaction Vectors



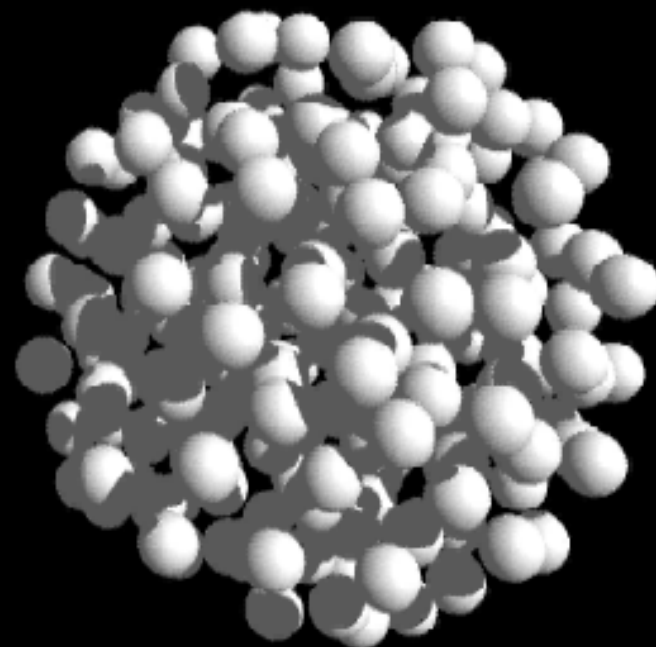
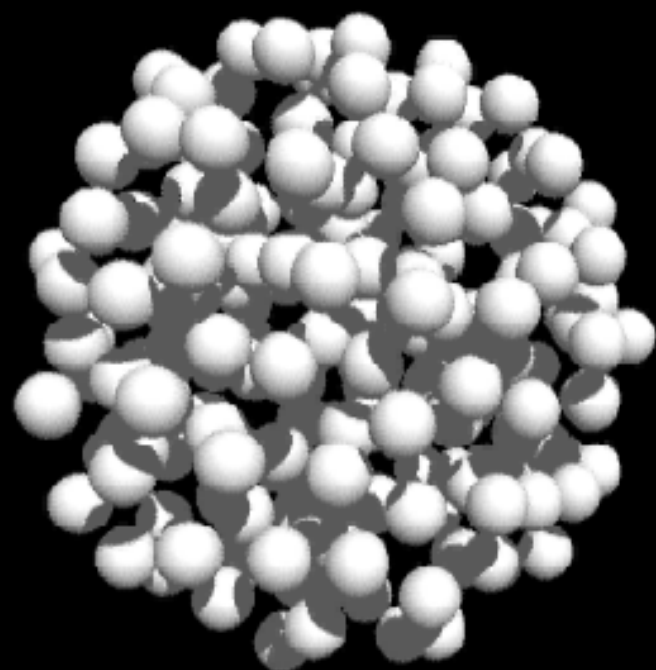
The Hot CNO Cycle



ESTRUTURA NUCLEAR

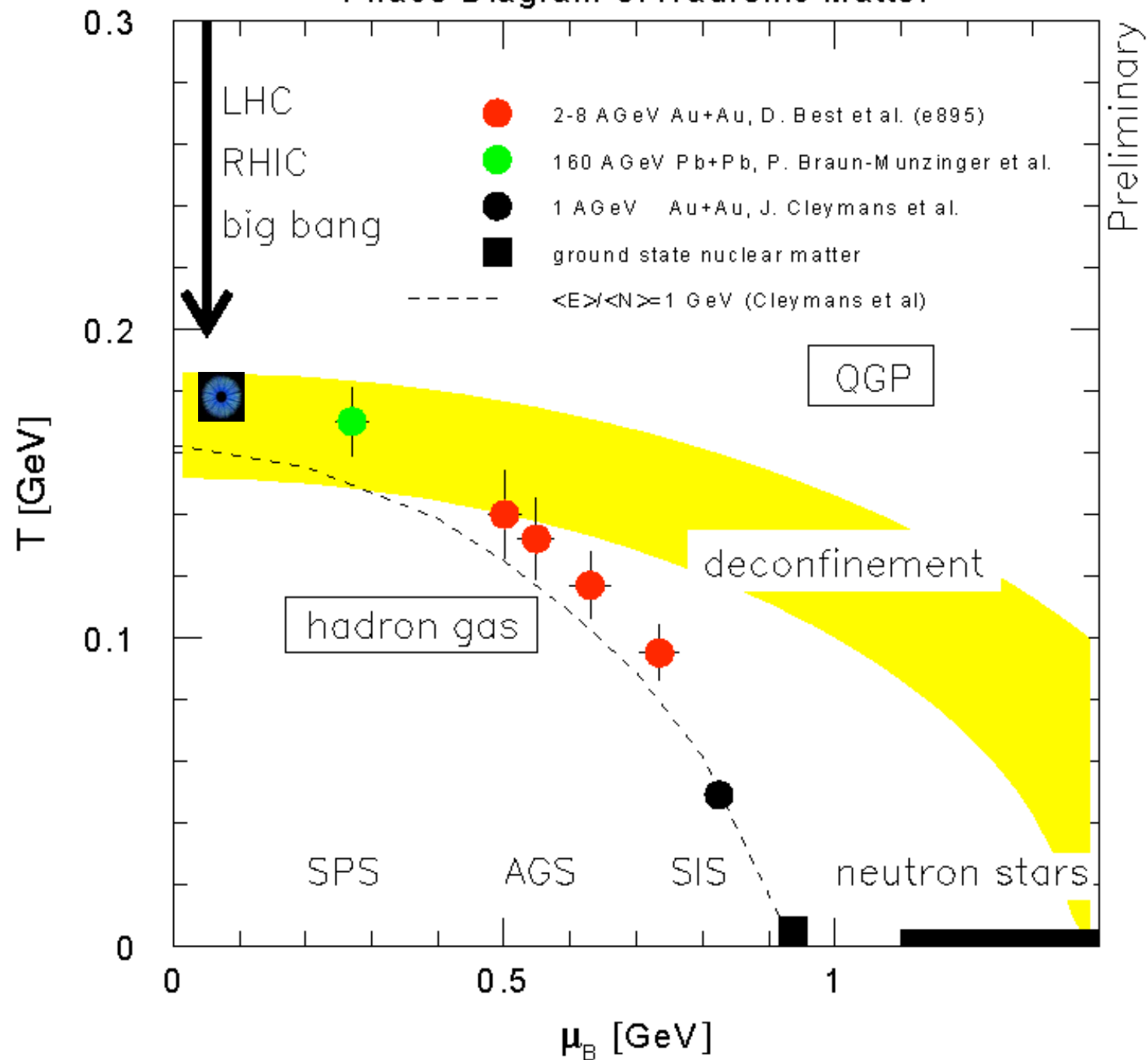
MECANISMO DE REAÇÕES

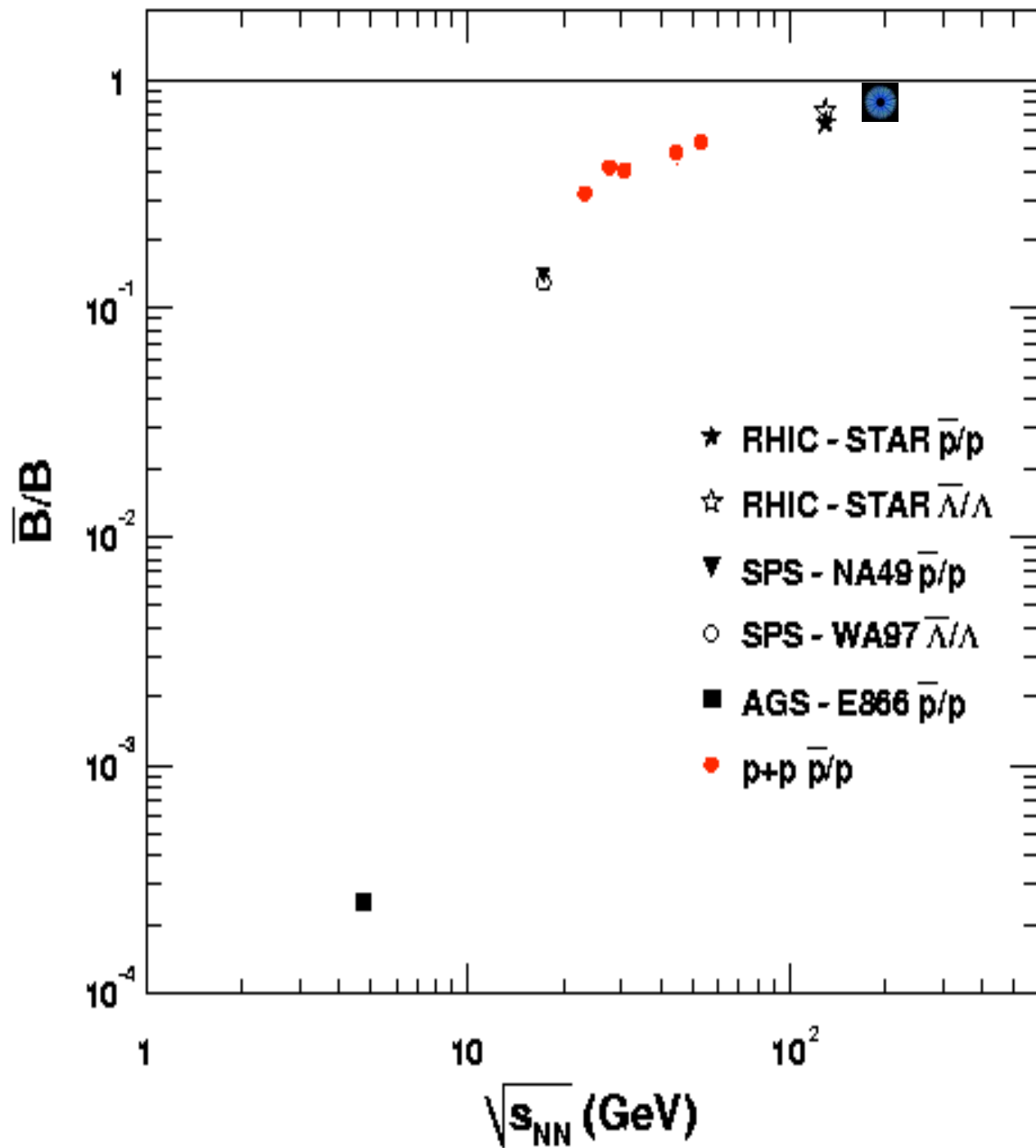
DINÂMICA DE REAÇÕES

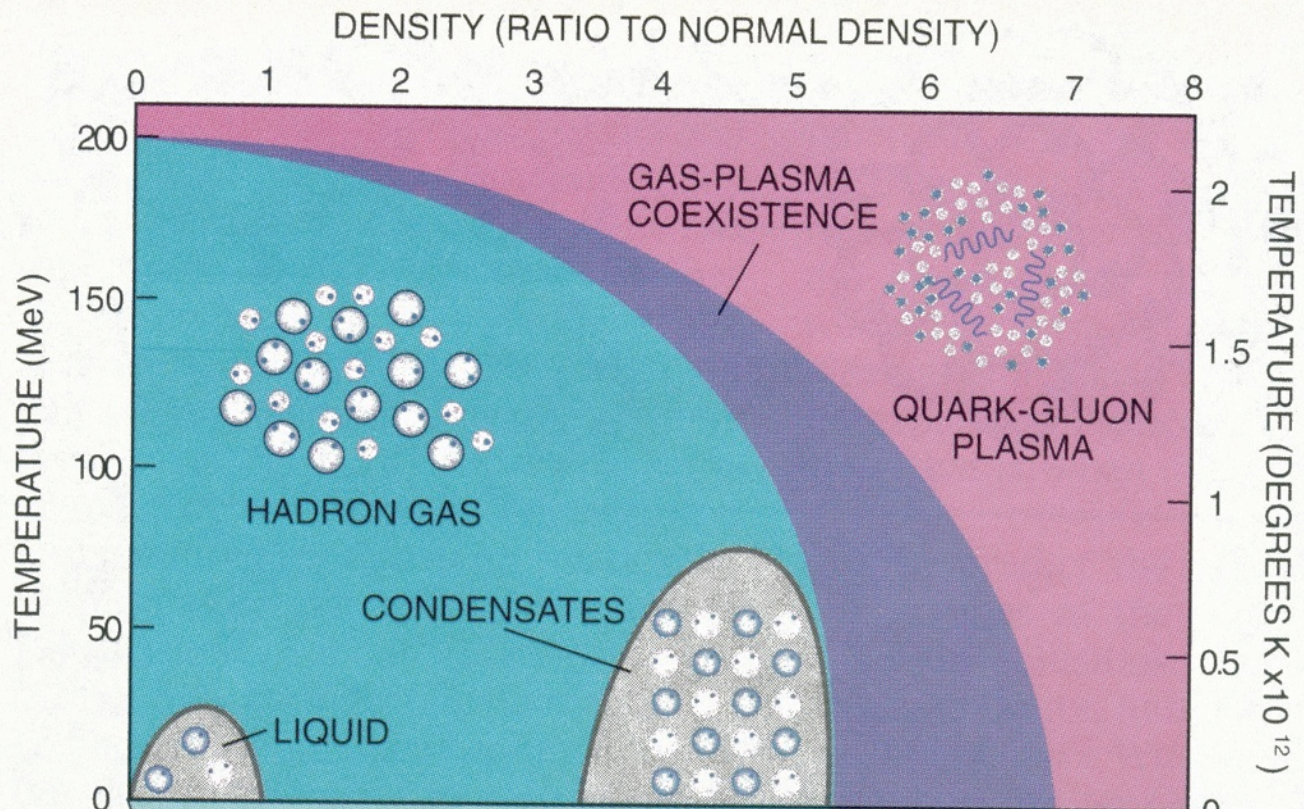




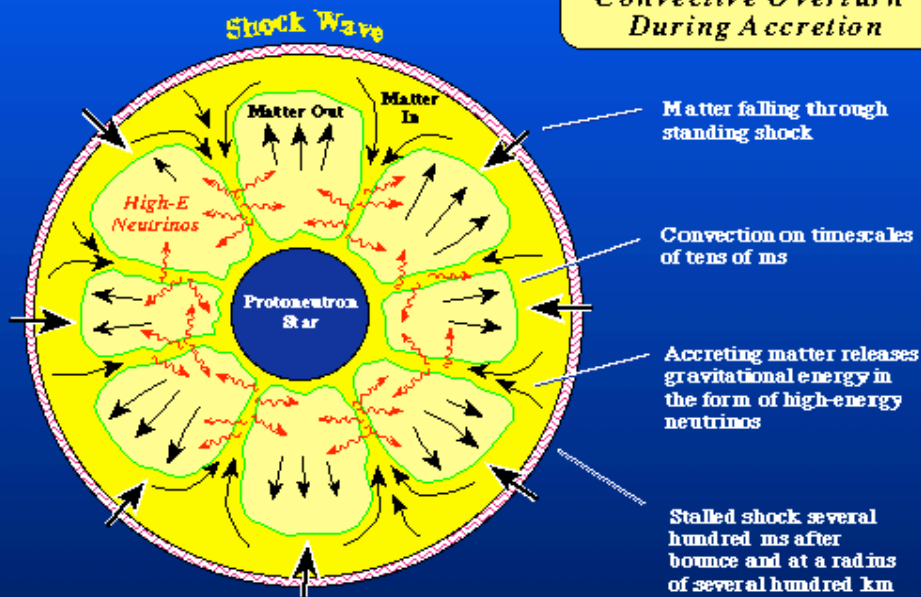
Phase Diagram of Hadronic Matter



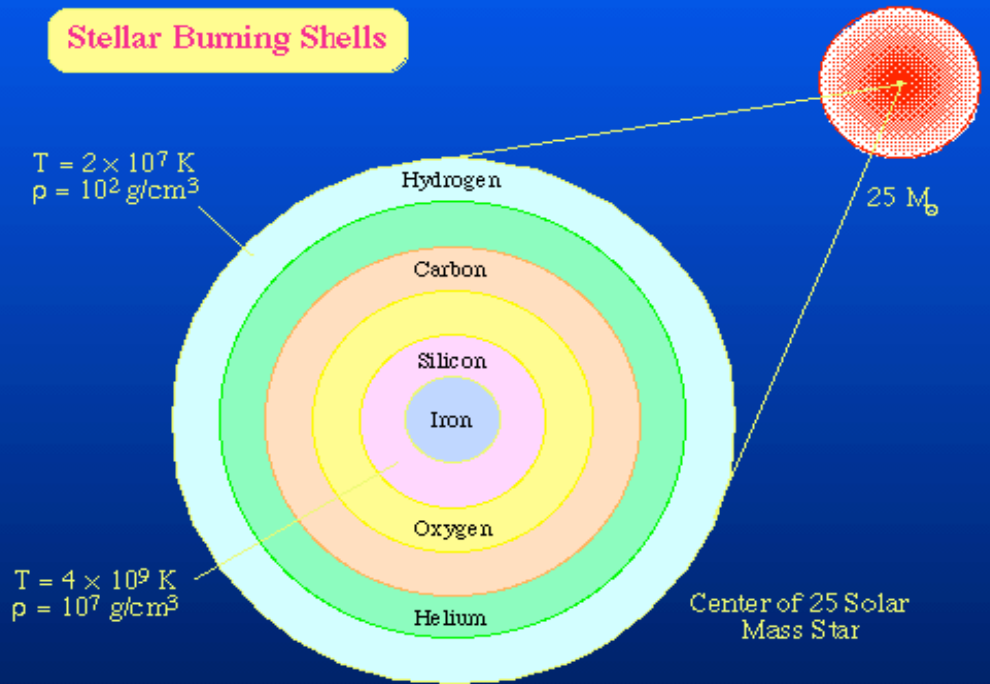




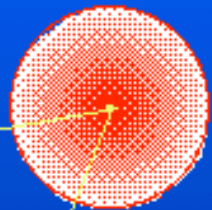
Convective Overturn During Accretion



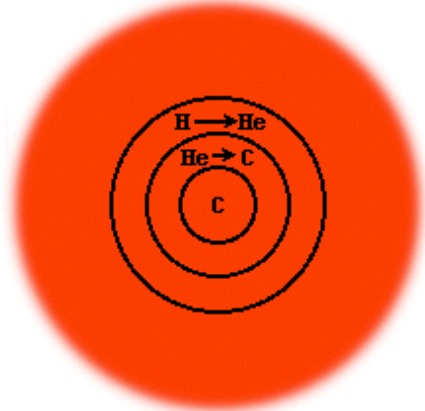
Stellar Burning Shells



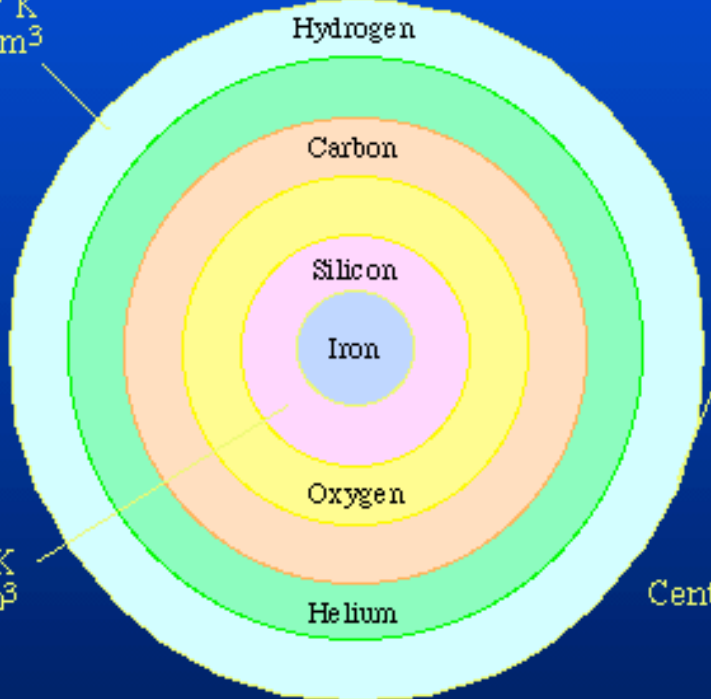
Stellar Burning Shells



25 M_☉

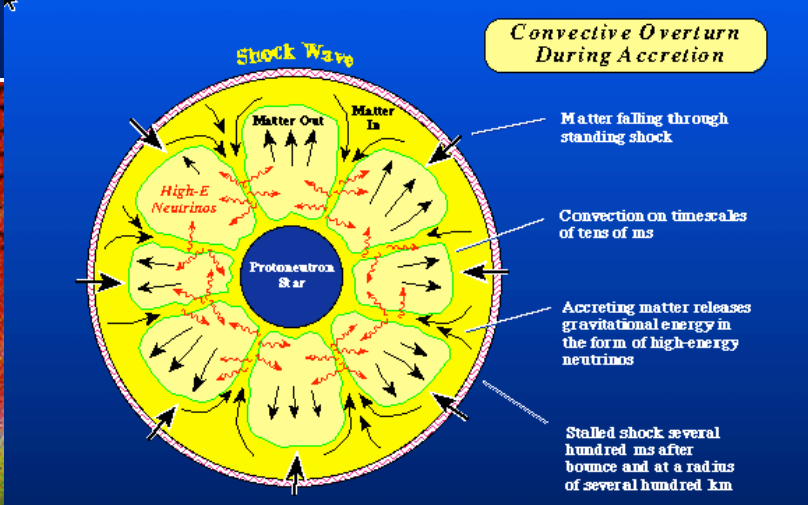
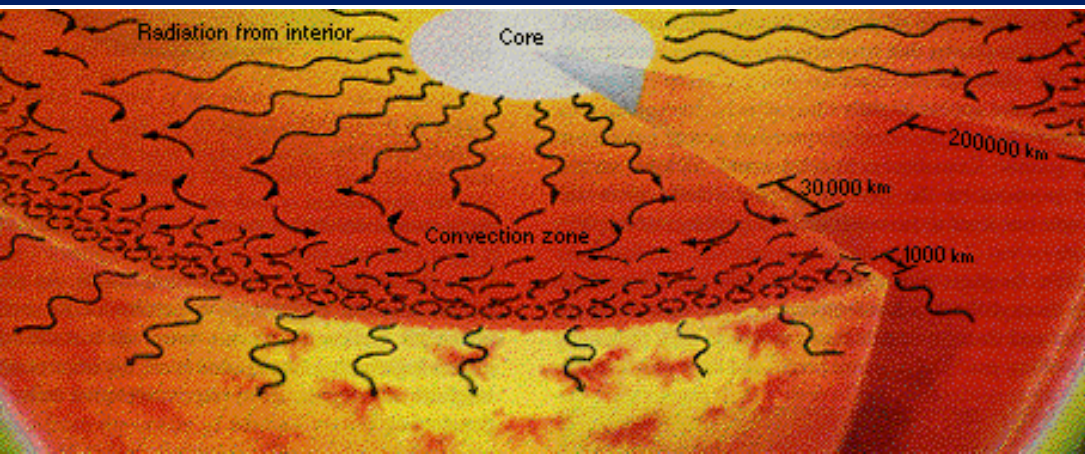
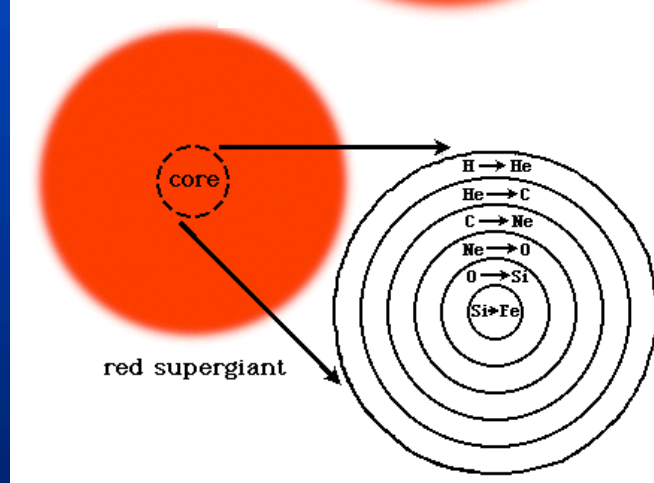


$T = 2 \times 10^7 \text{ K}$
 $\rho = 10^2 \text{ g/cm}^3$

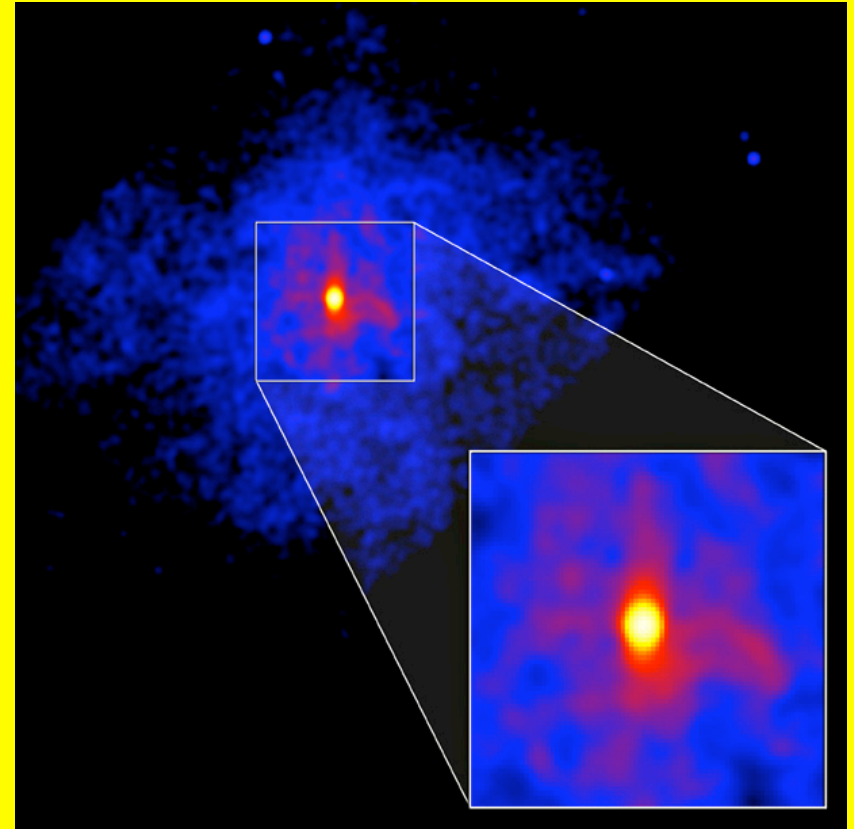
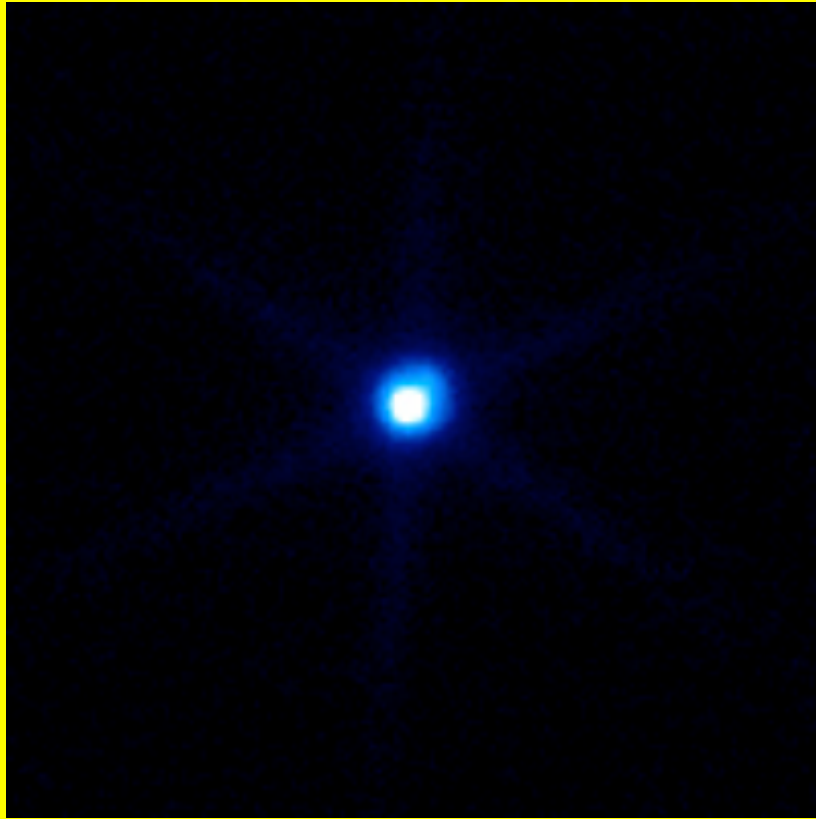


$T = 4 \times 10^9 \text{ K}$
 $\rho = 10^7 \text{ g/cm}^3$

Center of 25 Solar Mass Star



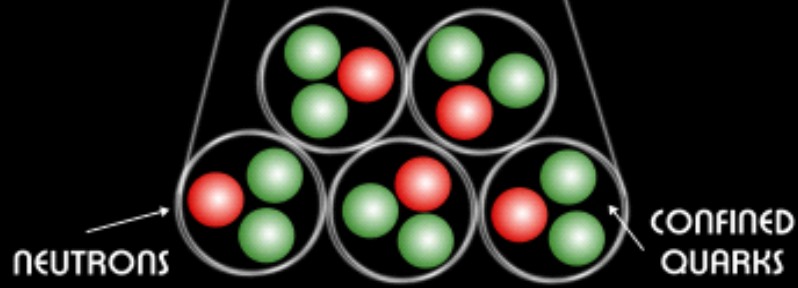
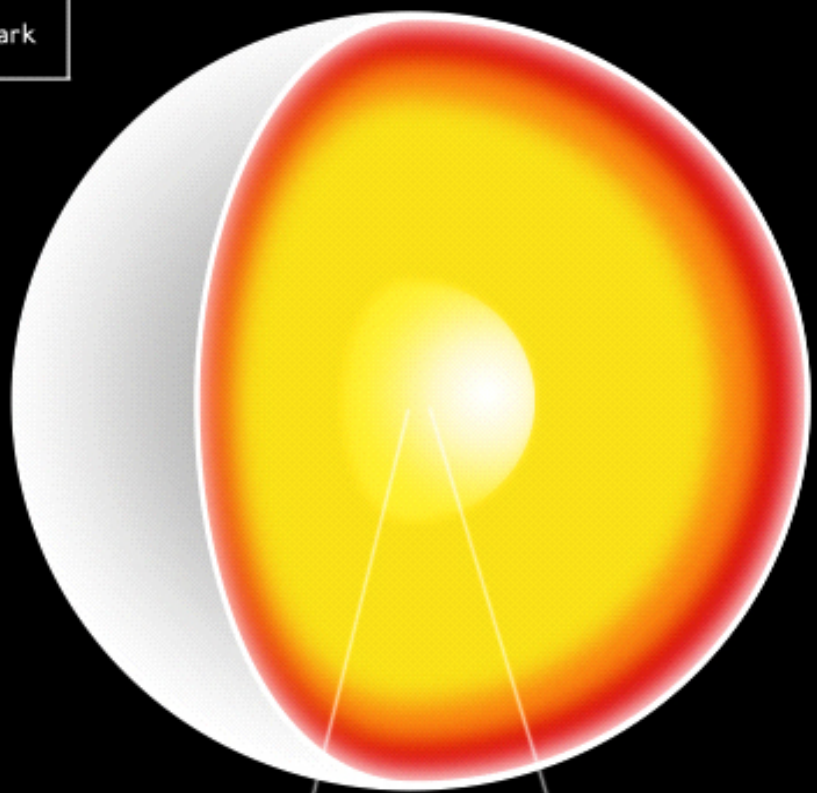
X-rays May Reveal New Form of Matter



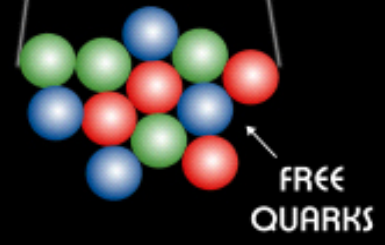
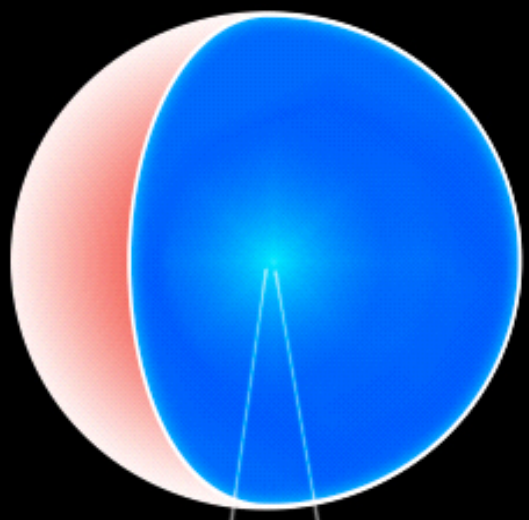
Chandra observations of RX J1856.5-3754 and the pulsar in 3C58 suggest that the matter in these collapsed stars is even denser than nuclear matter, the most dense matter found on Earth. This raises the possibility that **these stars are composed of free quarks** or crystals of sub-nuclear particles, rather than neutrons. One exciting possibility, predicted by some theories, is that the neutrons in the **star have dissolved at very high density into a soup of "up," "down" and "strange" quarks to form a "strange quark star"** which would explain the smaller radius.

- Up Quark
- Down Quark
- Strange Quark

Neutron Star



Strange Quark Star



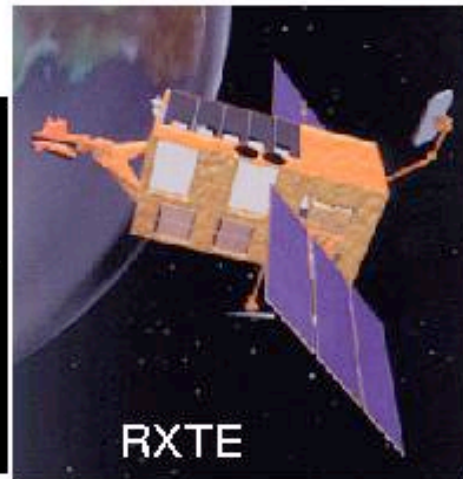
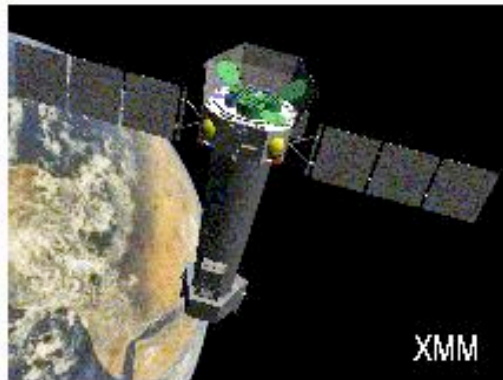
Unlocking the Mysteries of the Universe with state-of-the-art Telescopes



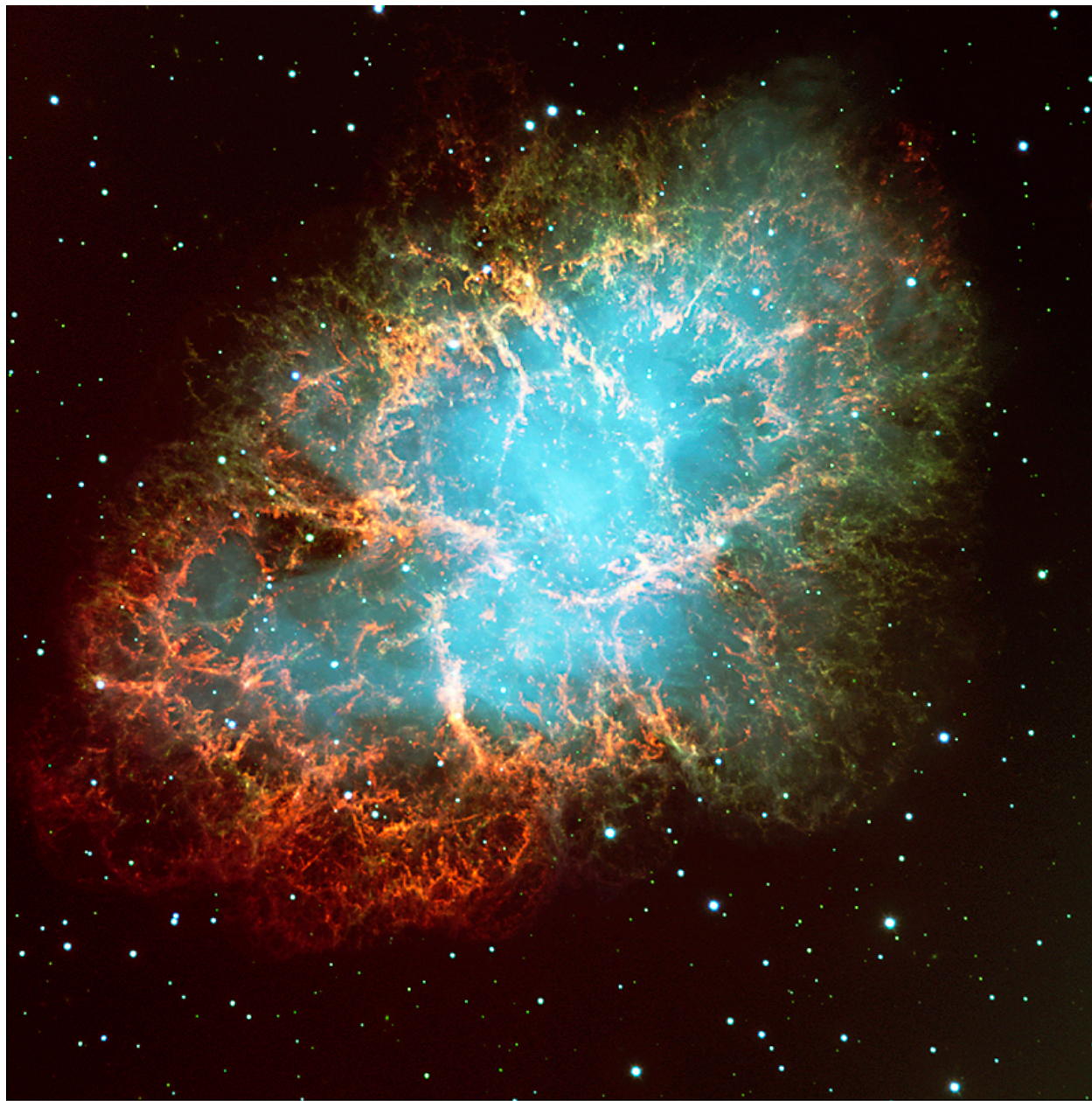
Search for 'exotic' pulsars



First lone NS, RXJ 185635-3754



First accreting ms pulsar, J1808-369



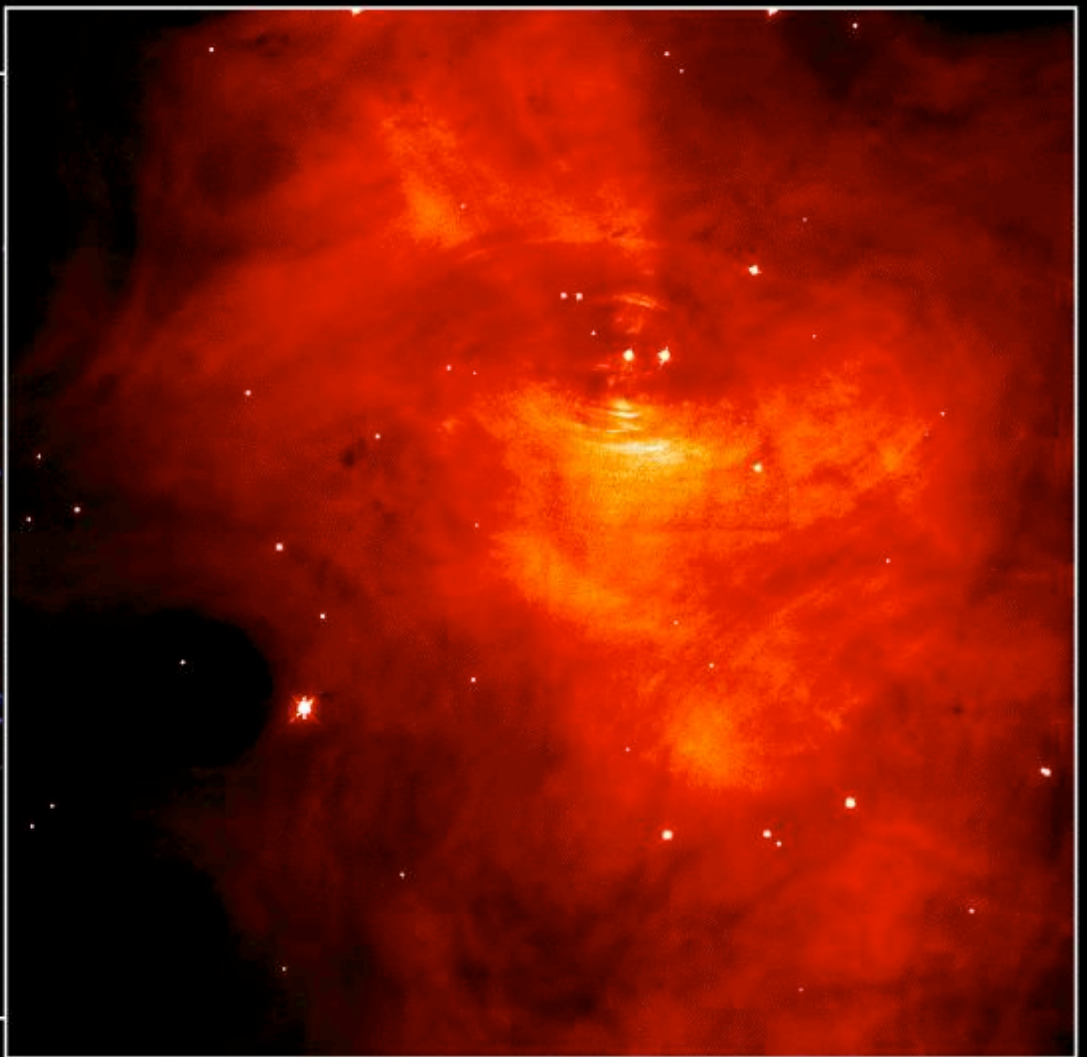
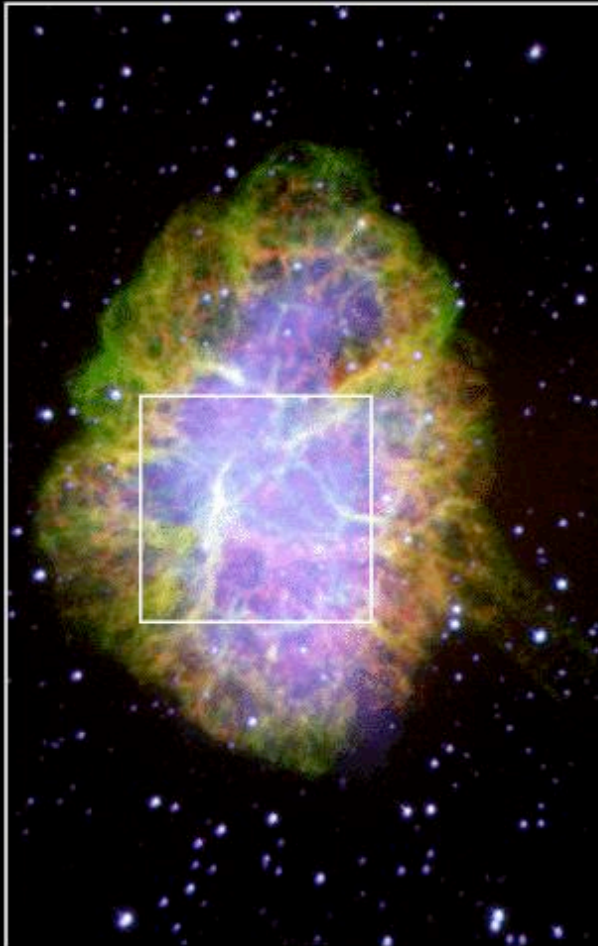
The Crab Nebula in Taurus (VLT KUEYEN + FORS2)

ESO PR Photo 40f/99 (17 November 1999)

© European Southern Observatory



Crab Nebula

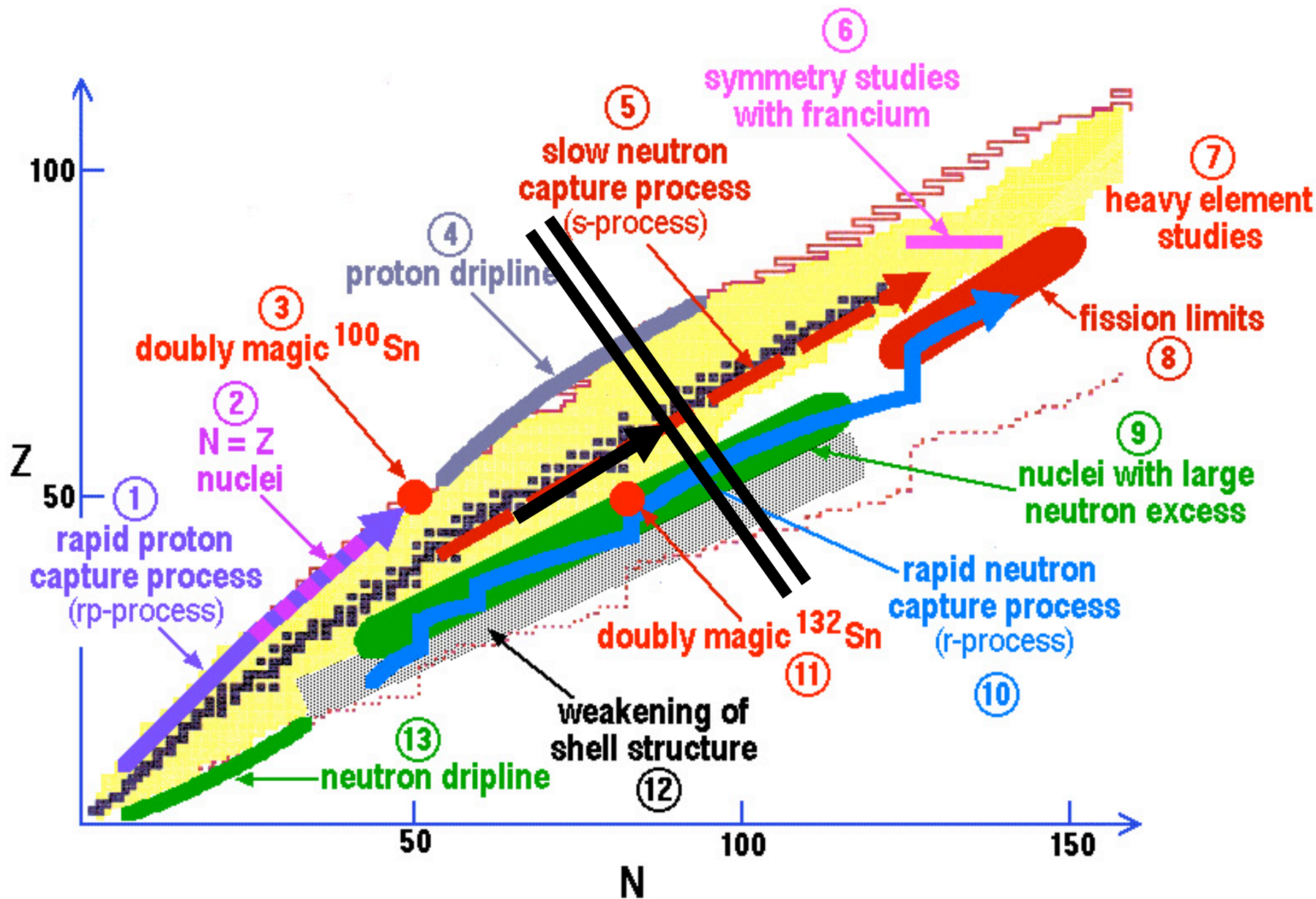


Palomar

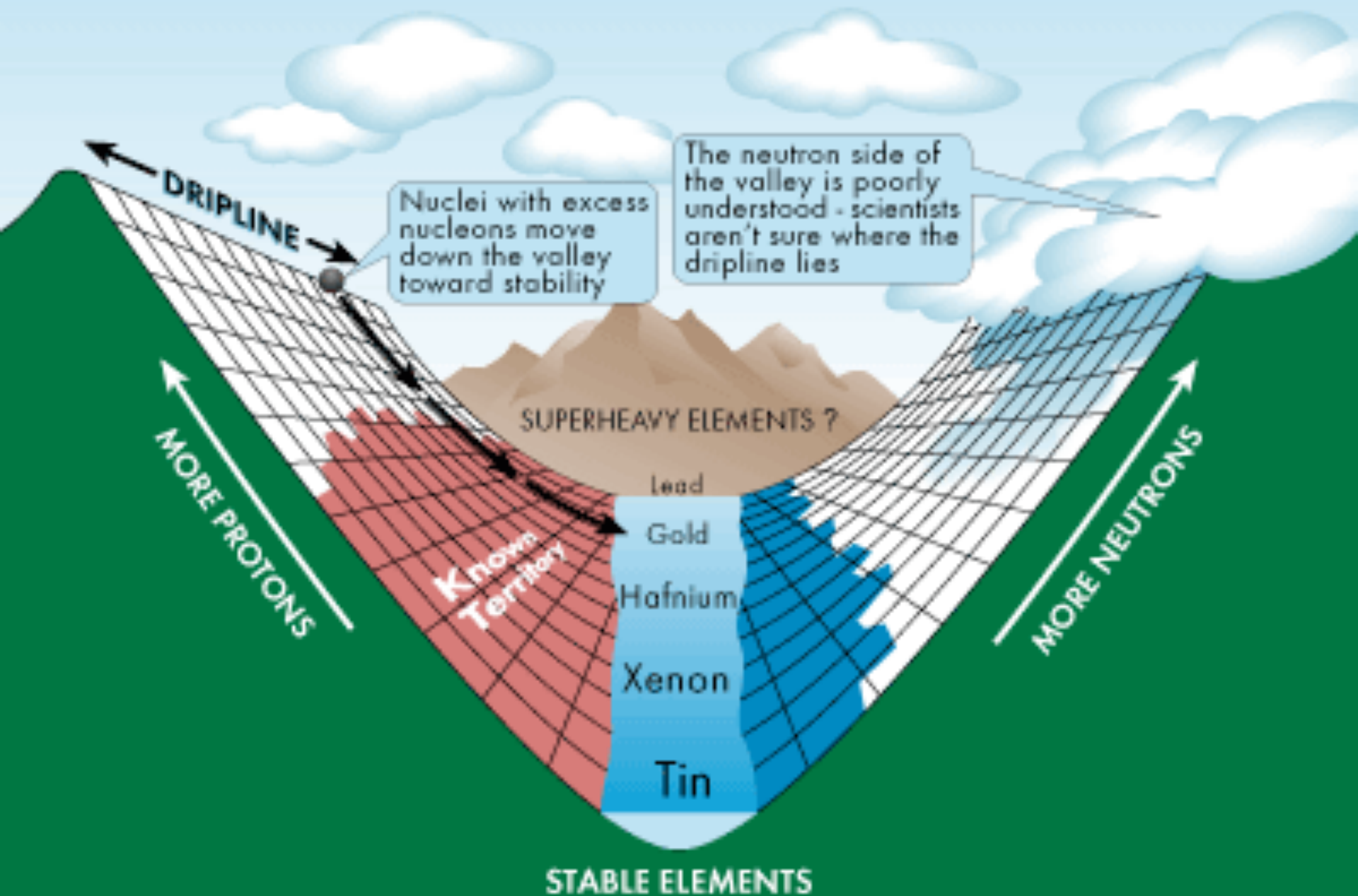
PRC96-22a · ST Sci OPO · May 30, 1996

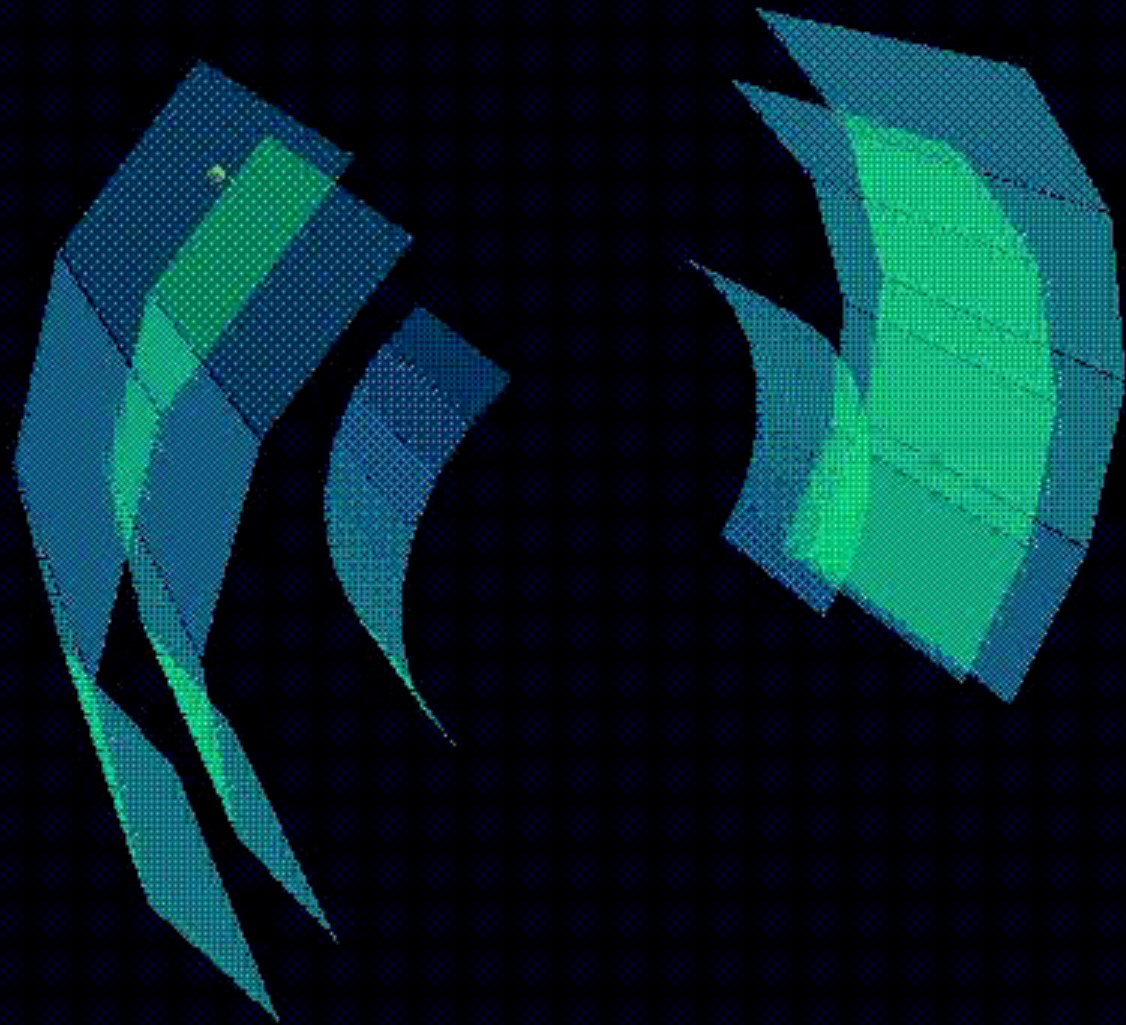
J. Hester and P. Scowen (AZ State Univ.) and NASA

HST · WFPC2

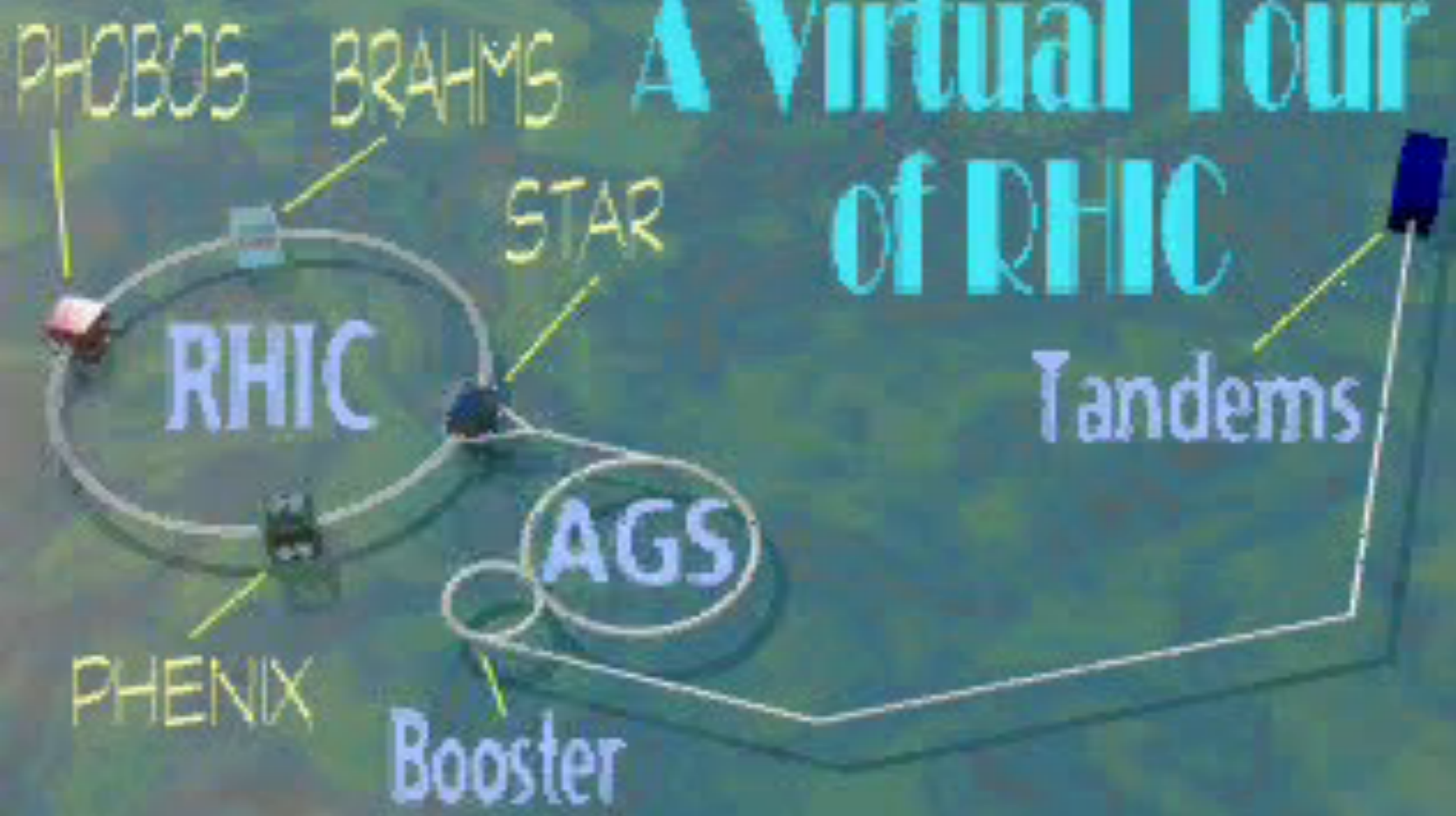


THE VALLEY OF STABILITY

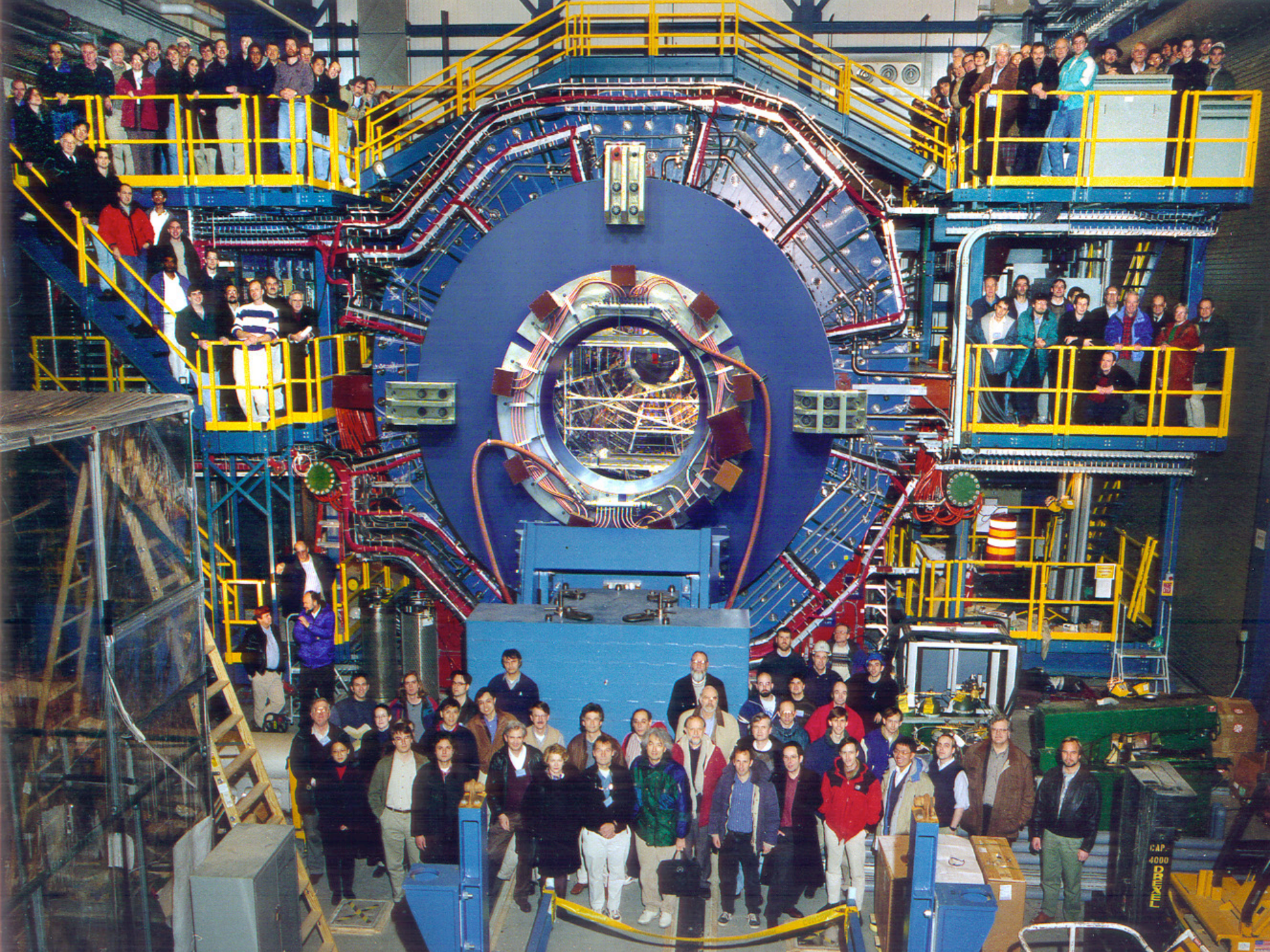


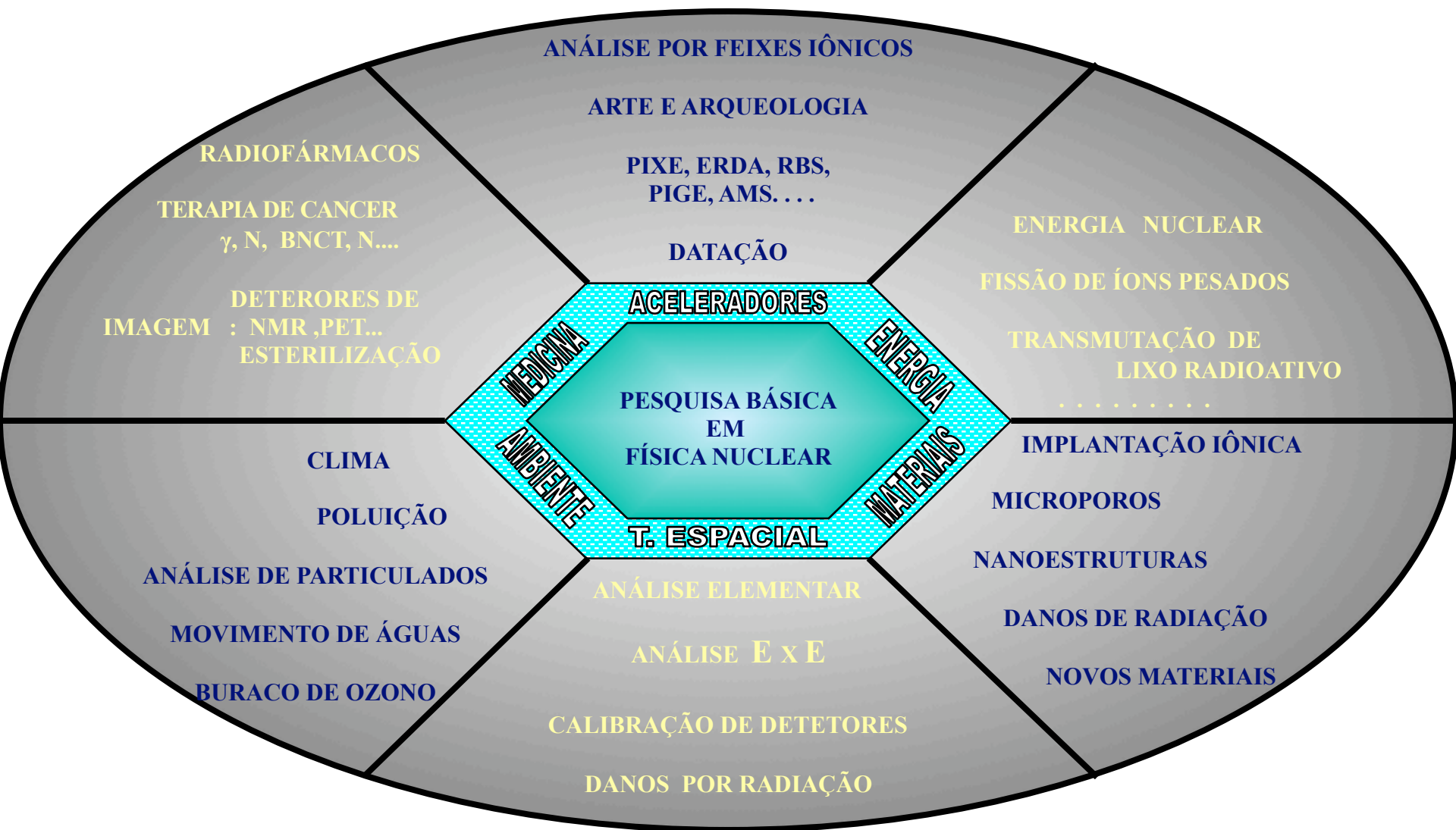


A Virtual Tour of RHIC



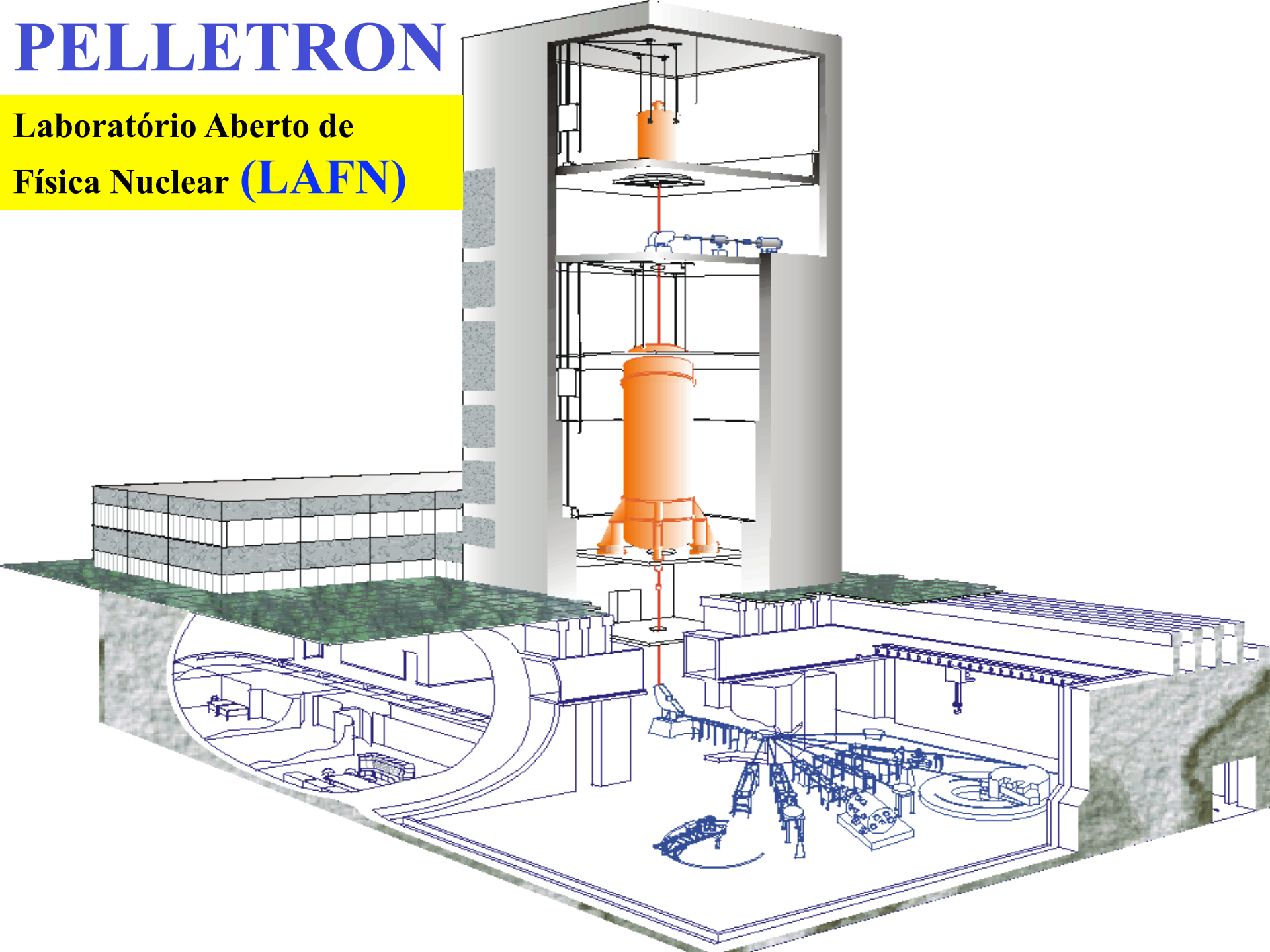
Animation by Jeffrey T. Mitchell





PELETRON

Laboratório Aberto de Física Nuclear (LAFN)

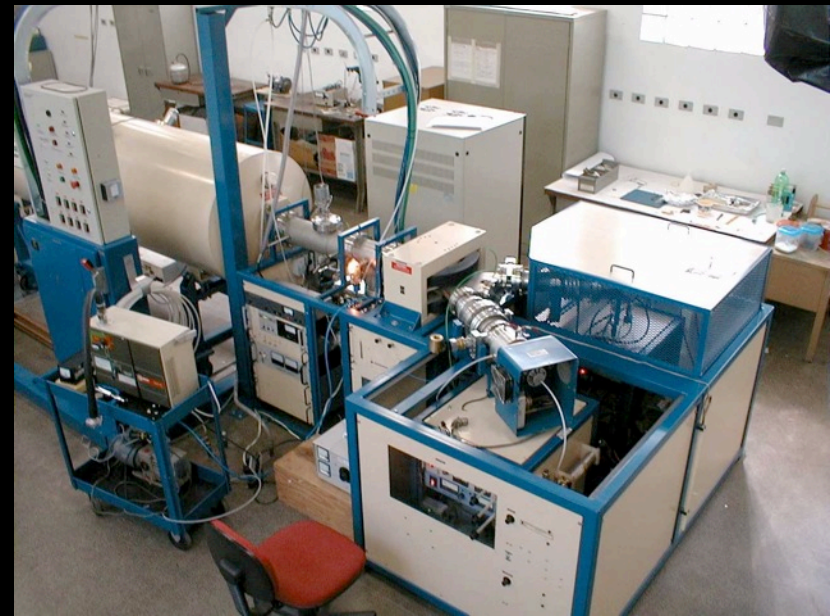




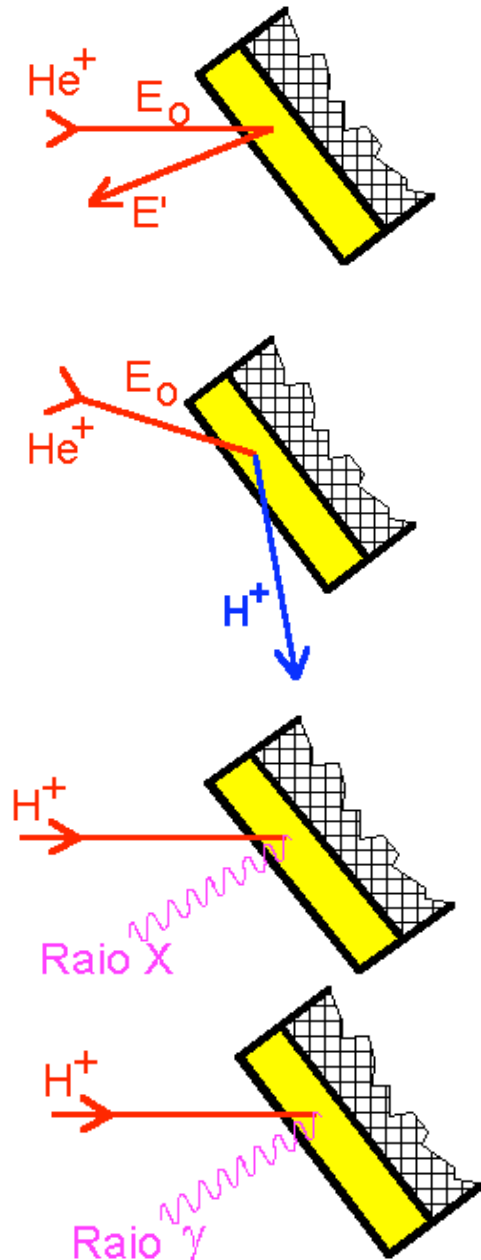
**Acelerador Pelletron,
tandem, $V_{\text{max}} = 1.7$
MV com *stripper*
gasoso de N_2**

**Fonte de íons tipo RF modelo Alphasross
com célula de troca de carga com Rb, para
feixes de He e H.**

**Fonte SNICS (Fonte de íons negativos por
sputtering de Césio), para feixes de H, C, O,
Cl, Si...**



Métodos Analíticos no LAMFI - USP



RBS -

Rutherford Backscattering Spectrometry

Concentração e perfil em profundidade

Medida absoluta em átomos/cm²

Sensibilidade < 10¹² Au/cm²

Rápido (10 min)

Sensível à topografia da camada e interface

FRS - Forward Recoil Spectrometry

ERS - Elastic Recoil Spectrometry

ERDA - Elastic Recoil Detection Analysis

(Concentração e perfil em profundidade)

Medida absoluta em átomos/cm²

Sensibilidade < 10¹² Au/cm²

Rápido (10 min)

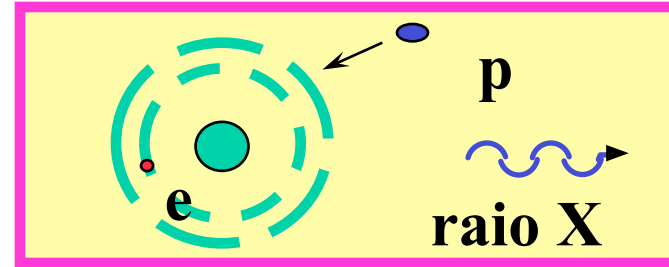
Sensível à topografia da camada e interface

PIXE - Particle Induced X-ray Emission

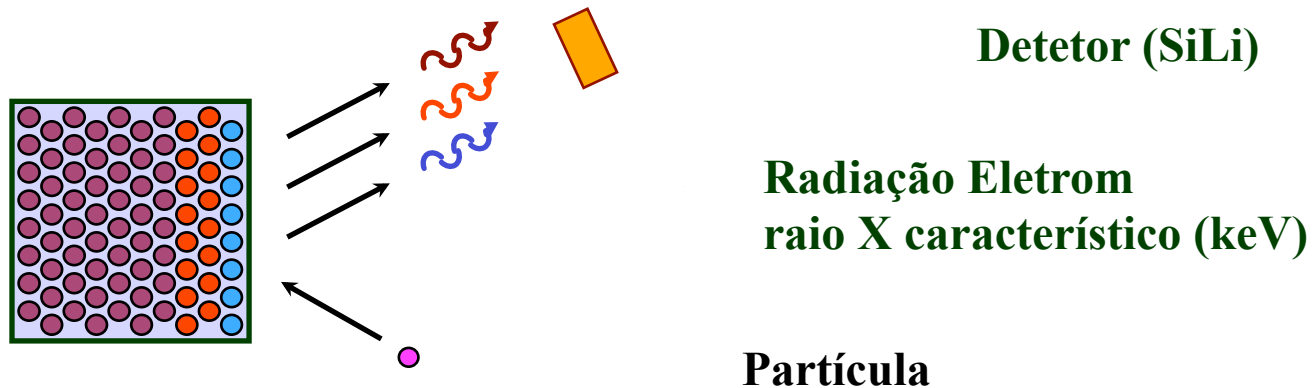
Medida absoluta em átomos/cm²

Alta resolução para elementos vizinhos

PIXE



- **Particle Induced X-ray Emission (proton)**



Características

Medidas absolutas em átomos/cm²

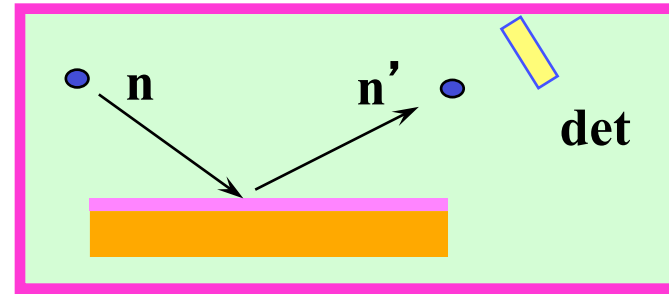
Alta sensibilidade (ppm)

Alta resolução para elementos vizinhos

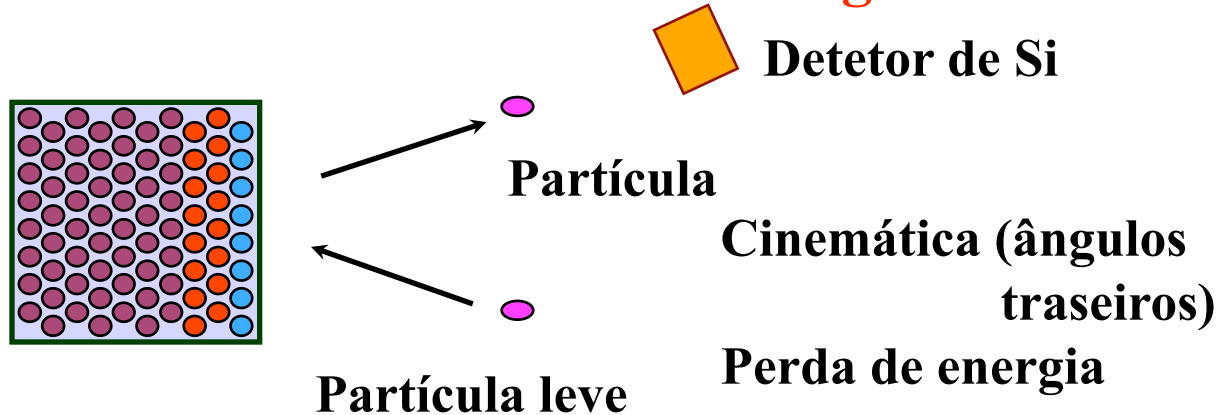
Eficiente para elementos mais pesados que Si

Rápido (10 min)

RBS



- **Rutherford BackScattering**



Características

Concentração e perfil em profundidade

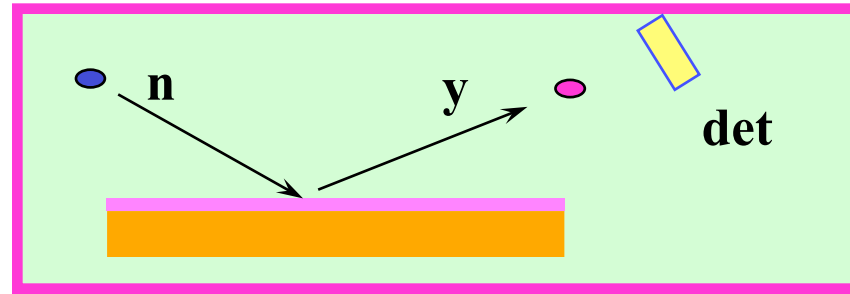
Medida absoluta em átomos/cm²

Sensibilidade < 10¹² Au/cm²

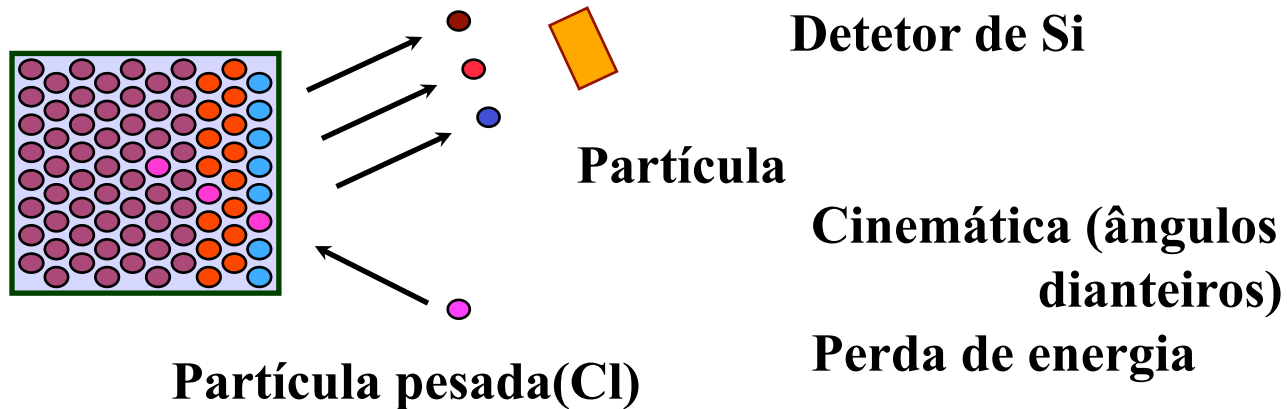
Sensível a topografia da interface

Eficiente para Z acima de Si

ERDA/ FRS



Elastic Recoil Detection Analysis



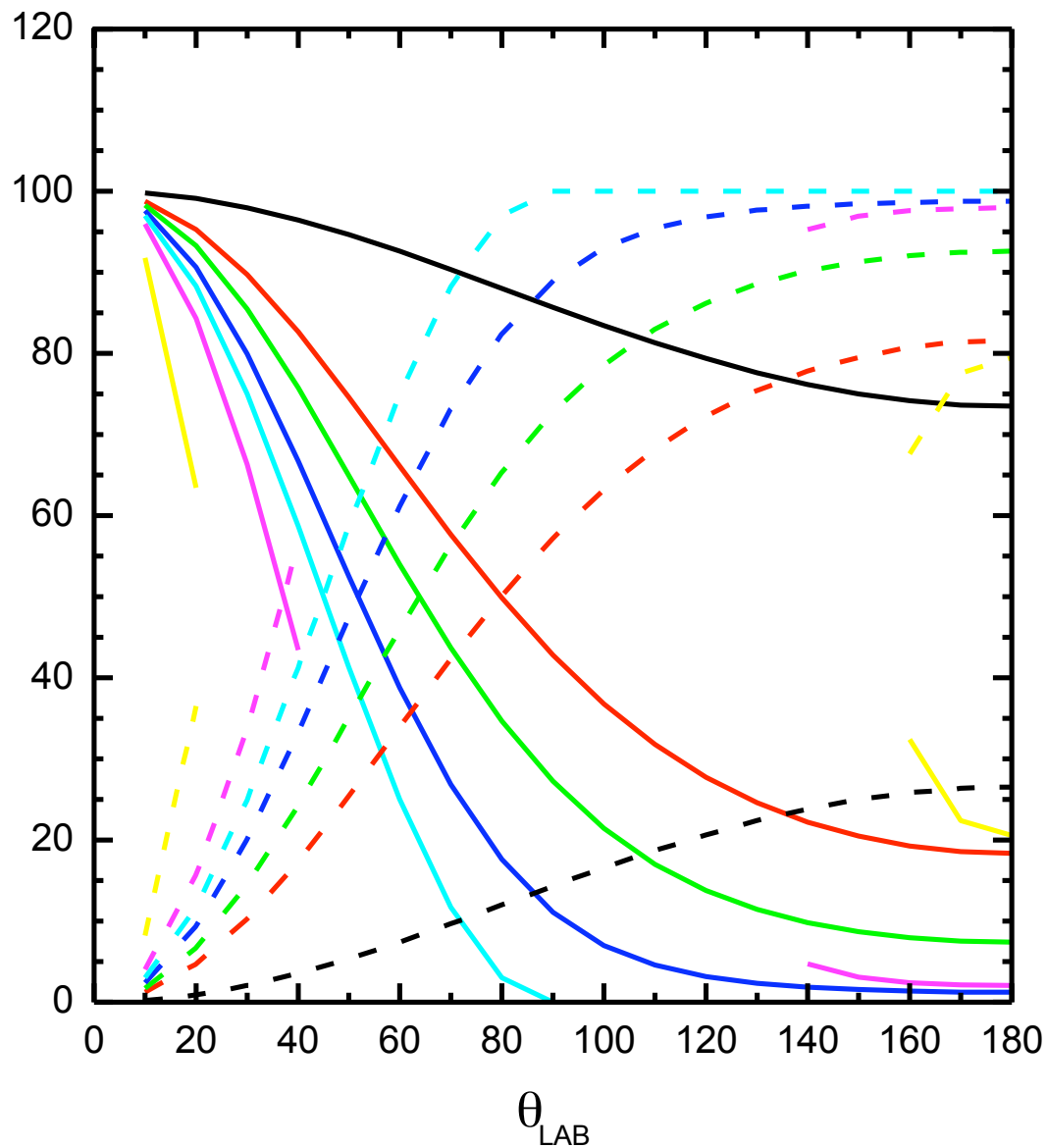
Características

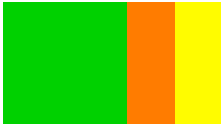
Concentração e perfil em profundidade

Medida absoluta em átomos/cm²

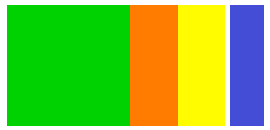
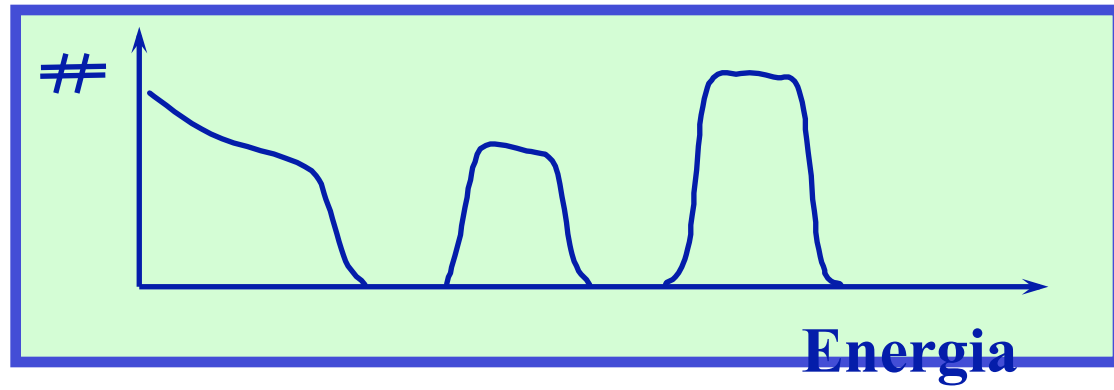
Eficiente para Z abaixo de Si (até H)

Rápido (10 min)

E_{OUT} (MeV) E_{RES} (MeV)

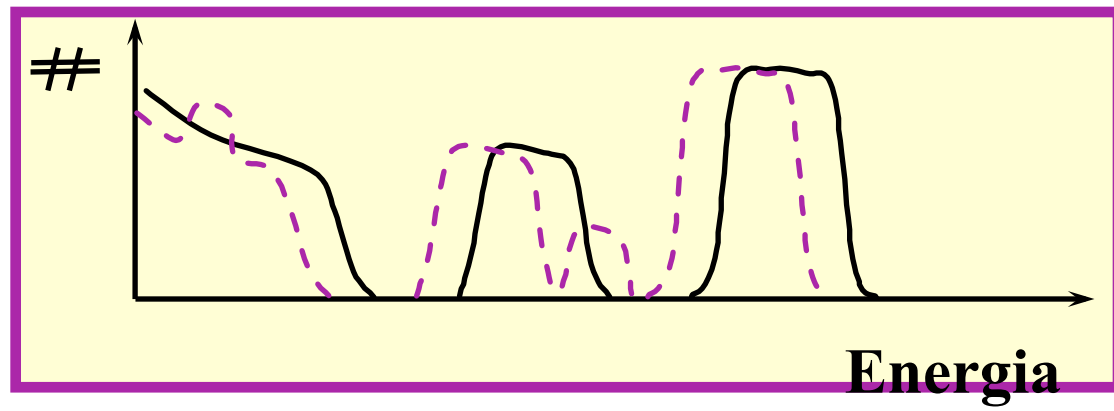


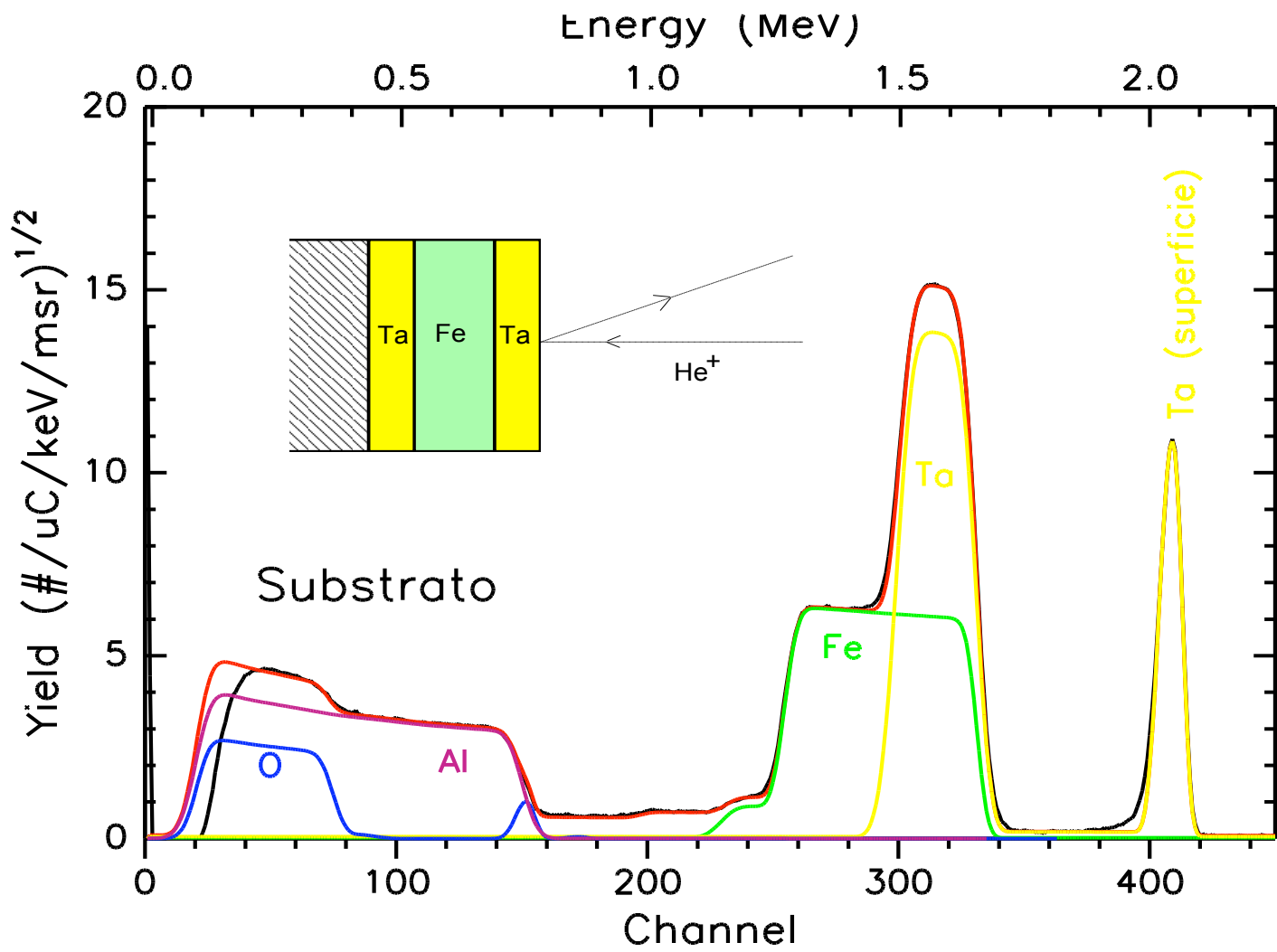
Si Ti Au



TiN

Si Ti Au





Espectro RBS típico de um filme tri-camada Ta/Fe/Ta sobre Al_2O_3 . As condições experimentais foram: RBS He^+ , 2.3MeV. Detecção a 170° , filme inclinado de 55° . Dados em linha preta, a simulação teórica com programa RUMP em vermelho e componentes em outras cores.

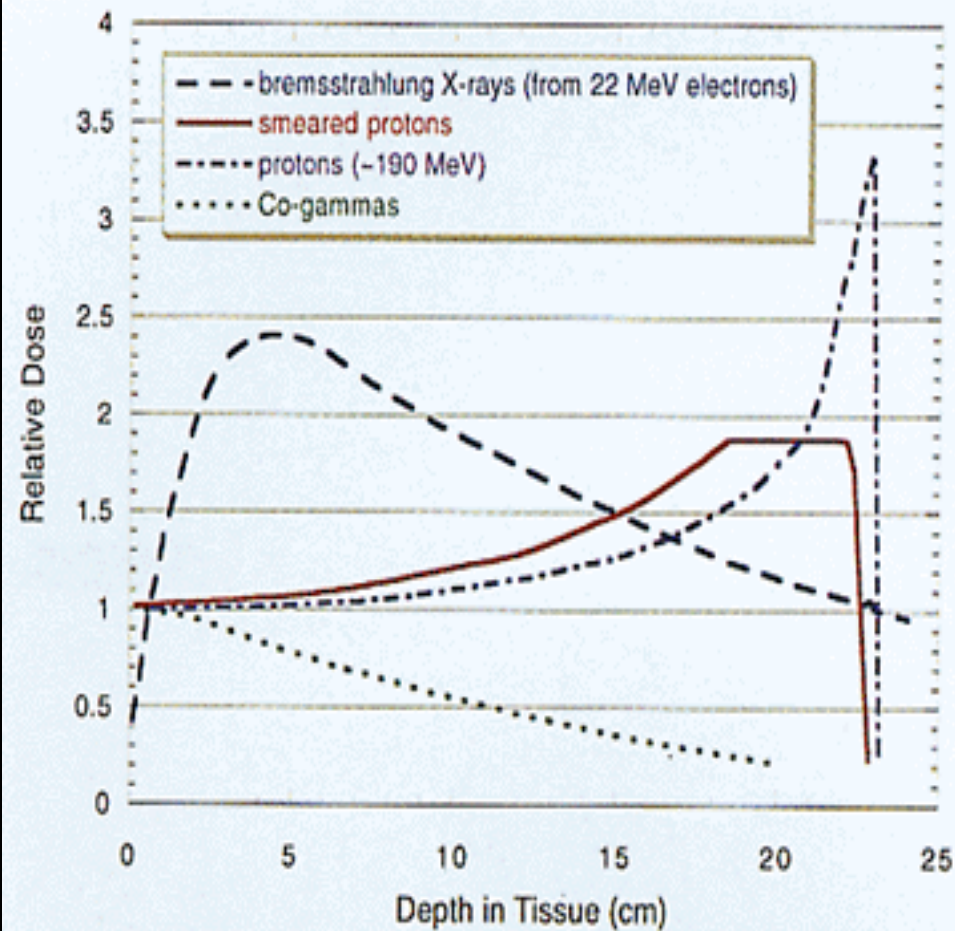
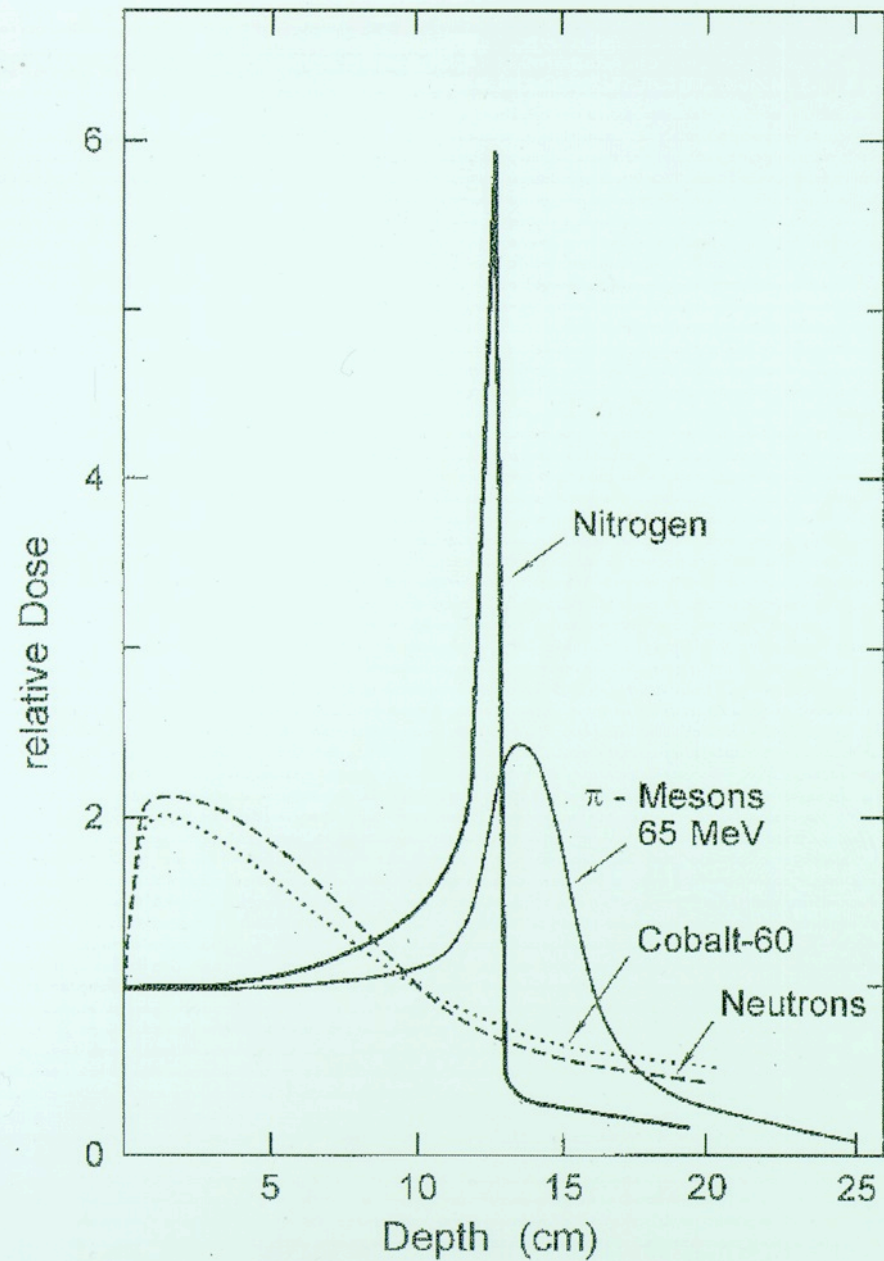
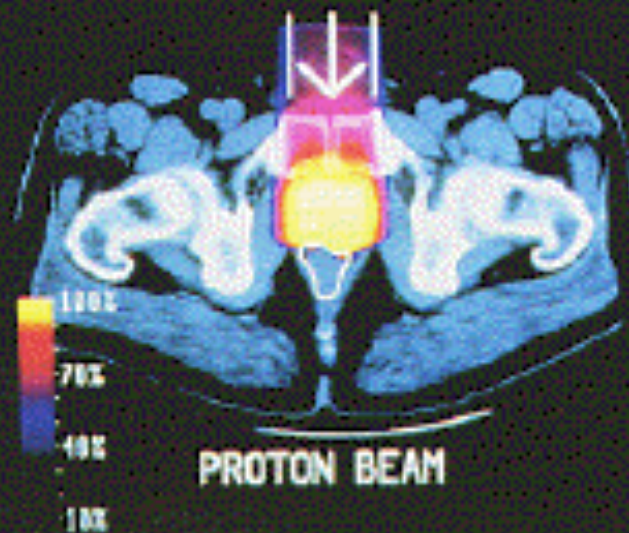
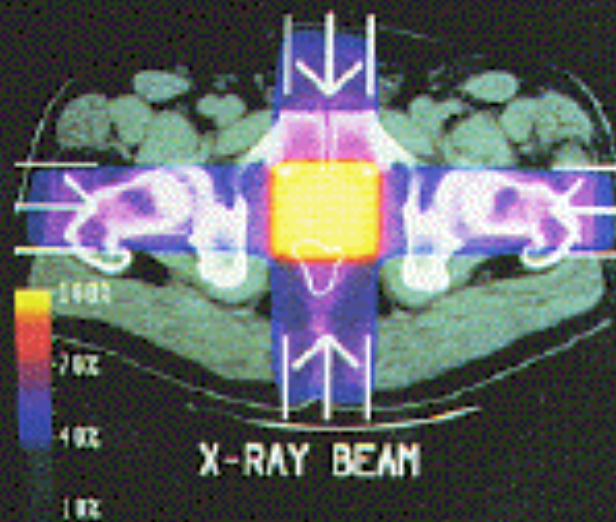
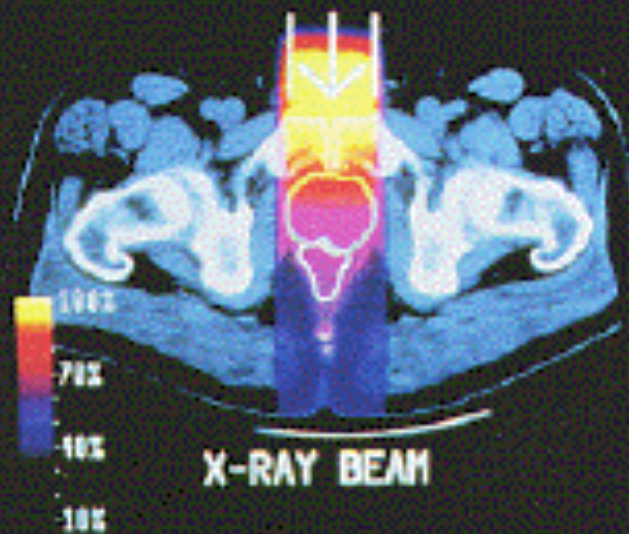
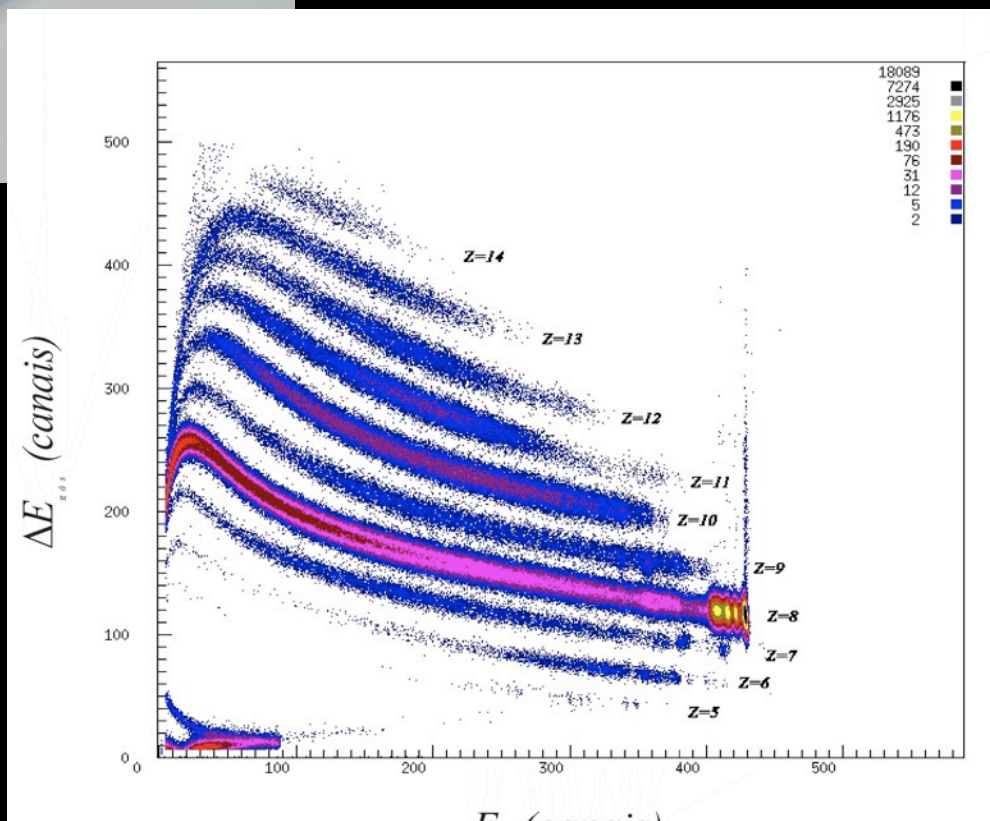


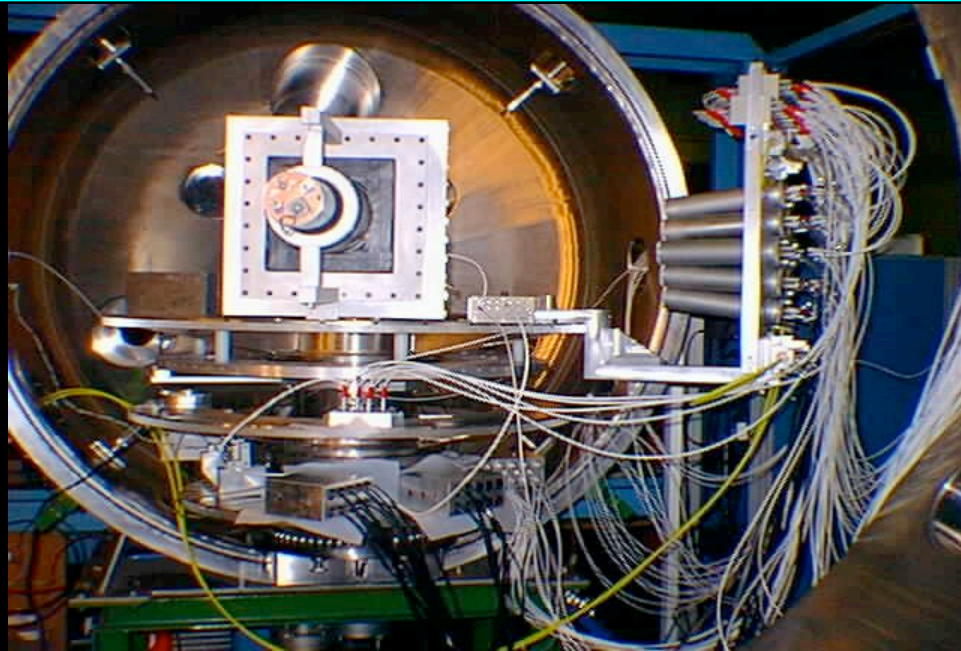
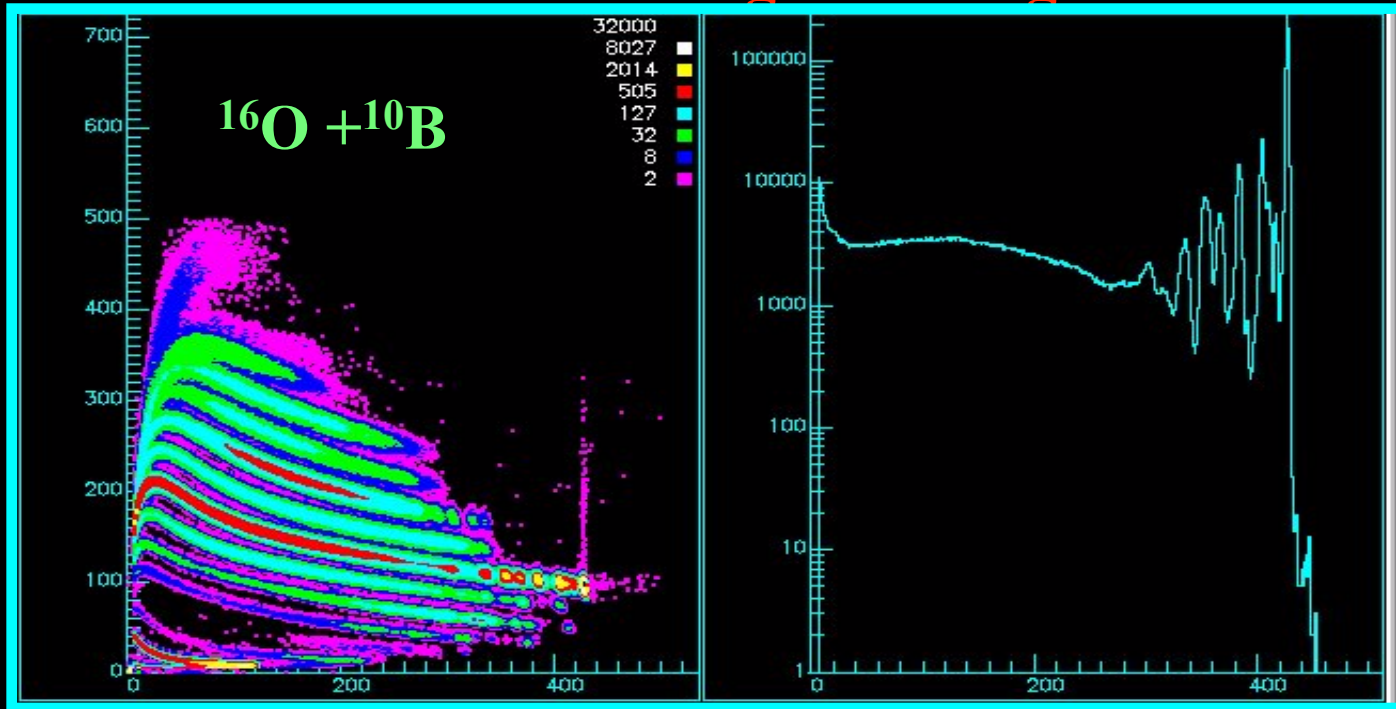
Figure VIII.1: Relative dose at various depths in tissue for bremsstrahlung X-rays generated by 22 MeV electrons (dashed curve), for 190 MeV protons (dot-dashed curve), for protons with smeared energy distribution to provide a plateau dose profile (solid curve), and from γ -rays produced by a cobalt 60 source (dotted curve).







Perda de energia x Energia



Poluição do Ar em São Paulo



São Paulo em um dia de inverno com baixos níveis de poluição



São Paulo em um dia de inverno com altos níveis de poluição

Fotos no topo do edifício da Faculdade de Medicina da USP, na Avenida Doutor Arnaldo.

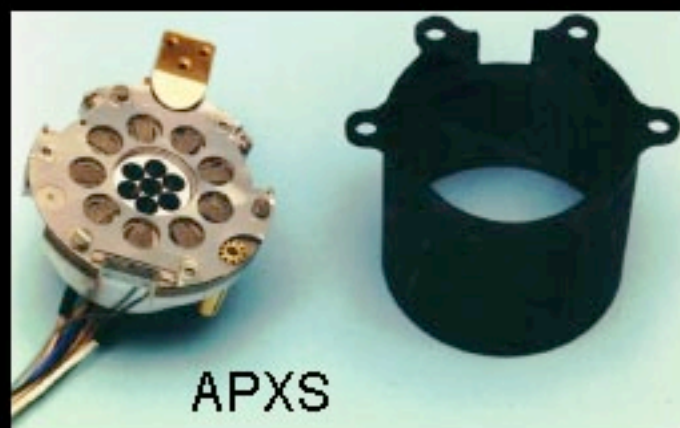
Mars Pathfinder Project

JPL

Pathfinder Science Payload



ASIIMET



MPF
R. Anderson



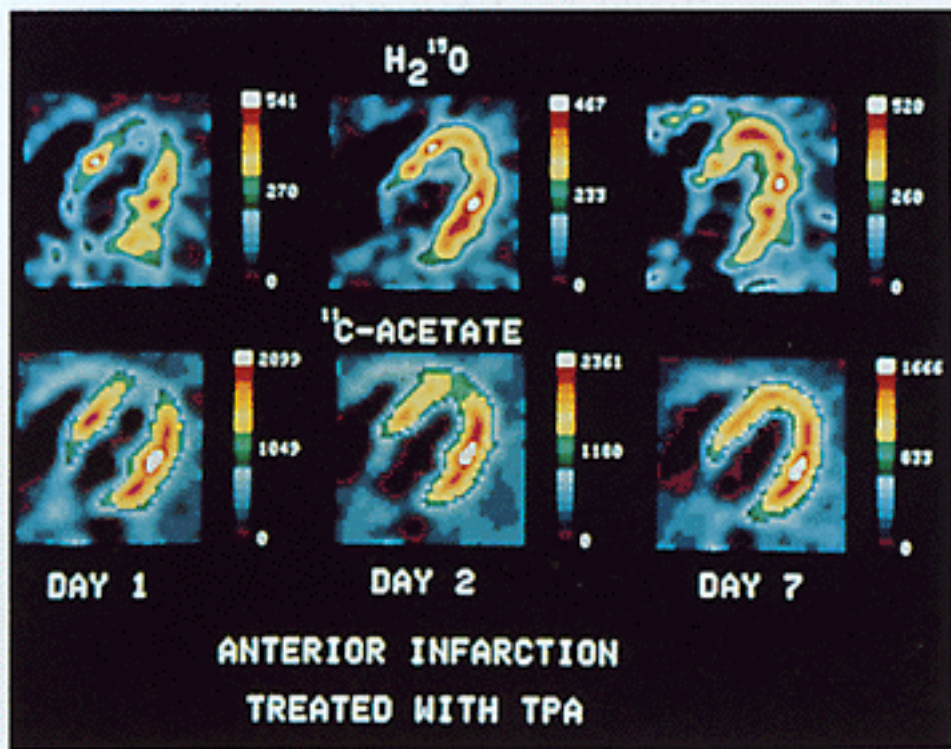
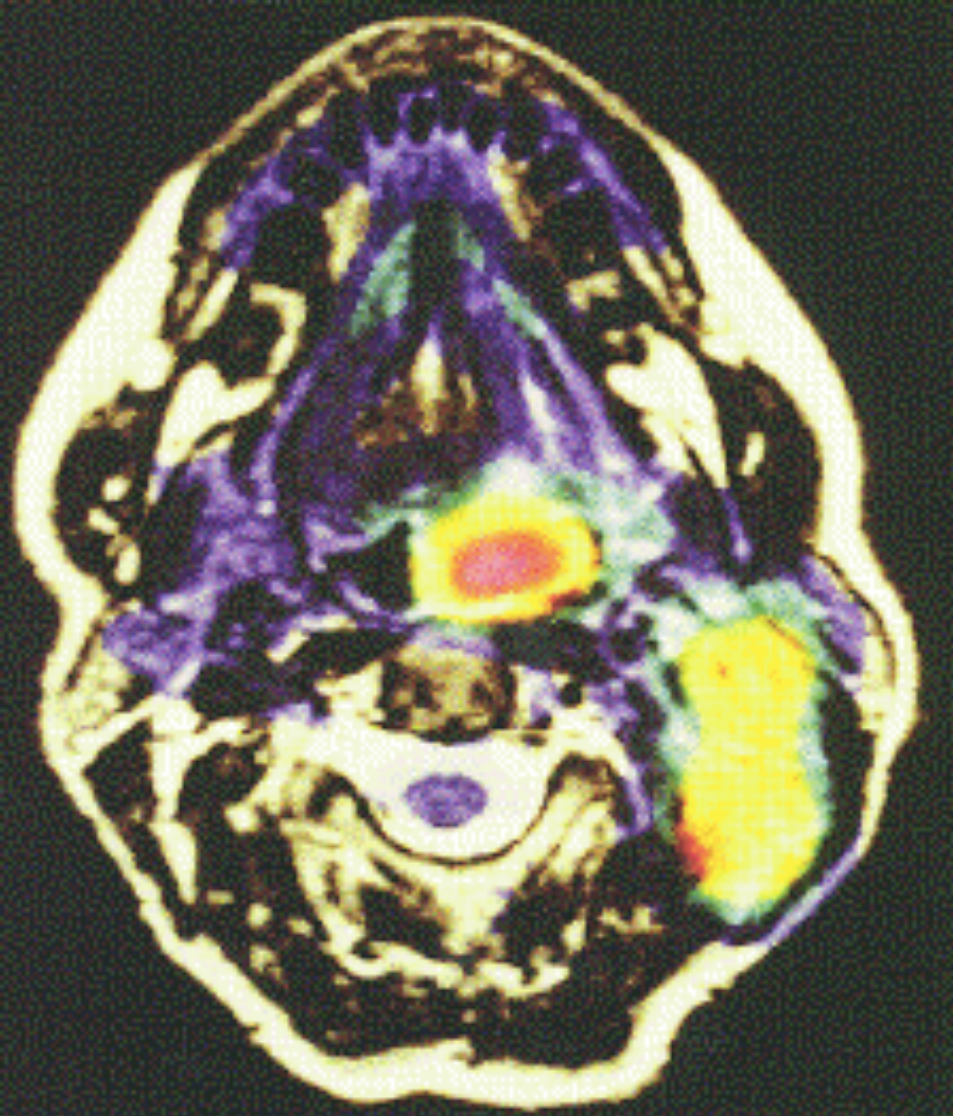
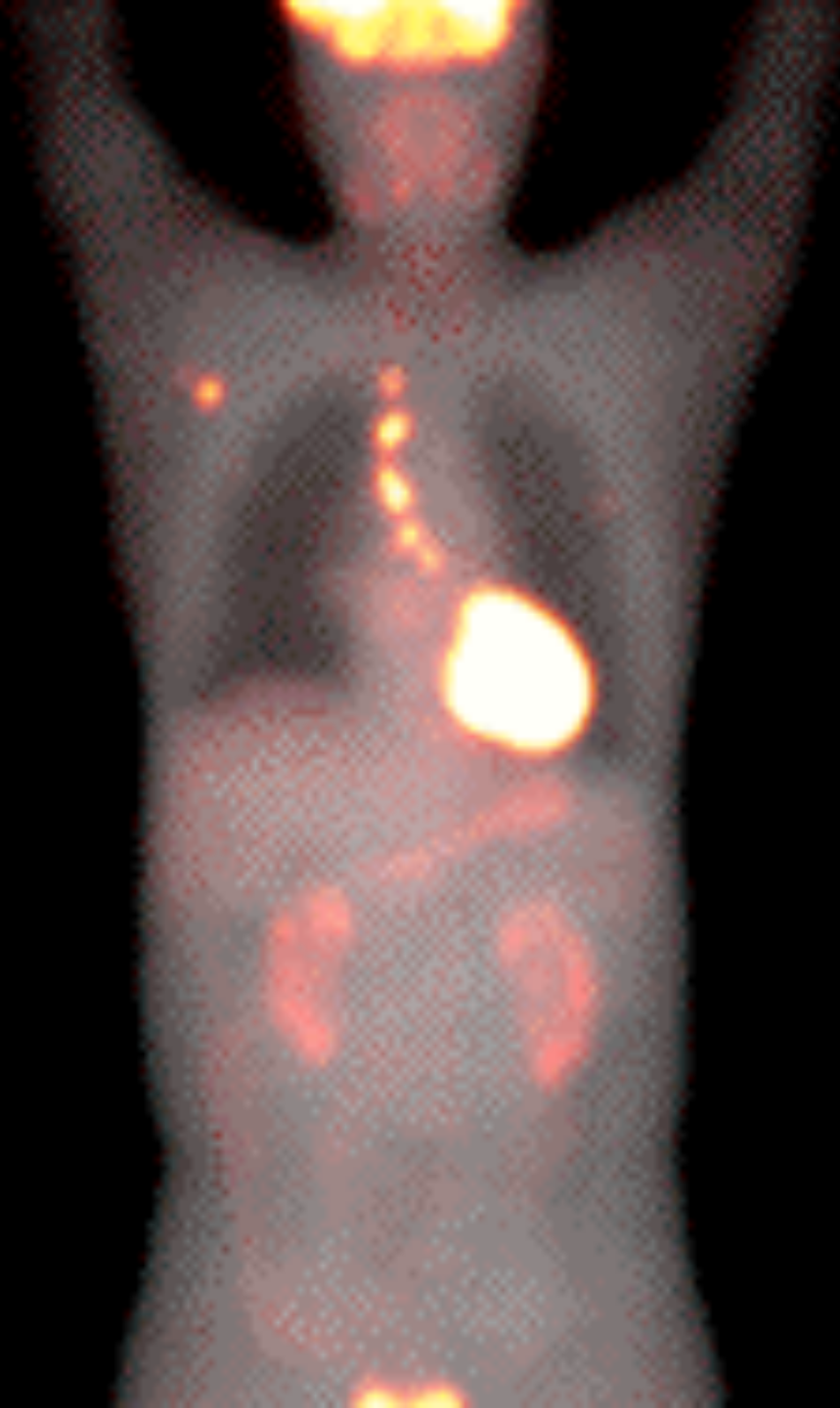


Figure VIII.2: The figure shows PET pictures of the heart of a patient with acute myocardial infarction treated with a thrombolytic agent. The top row shows scans after administration of water containing oxygen-15 nuclei to trace the blood flow. The bottom row shows tomograms obtained after administration of acetate containing carbon-11 nuclei to trace the heart's metabolism, i.e. its rate of oxygen usage. The defects are clearly visible on day 1, both in the impaired blood flow (top left) and the impaired metabolic use of oxygen (bottom left). Recovery of blood circulation has taken place on day 2 (top center) and is maintained through day 7 (top right). Defects in the metabolism are still visible on day 2 (bottom center); full recovery is seen by day 7.



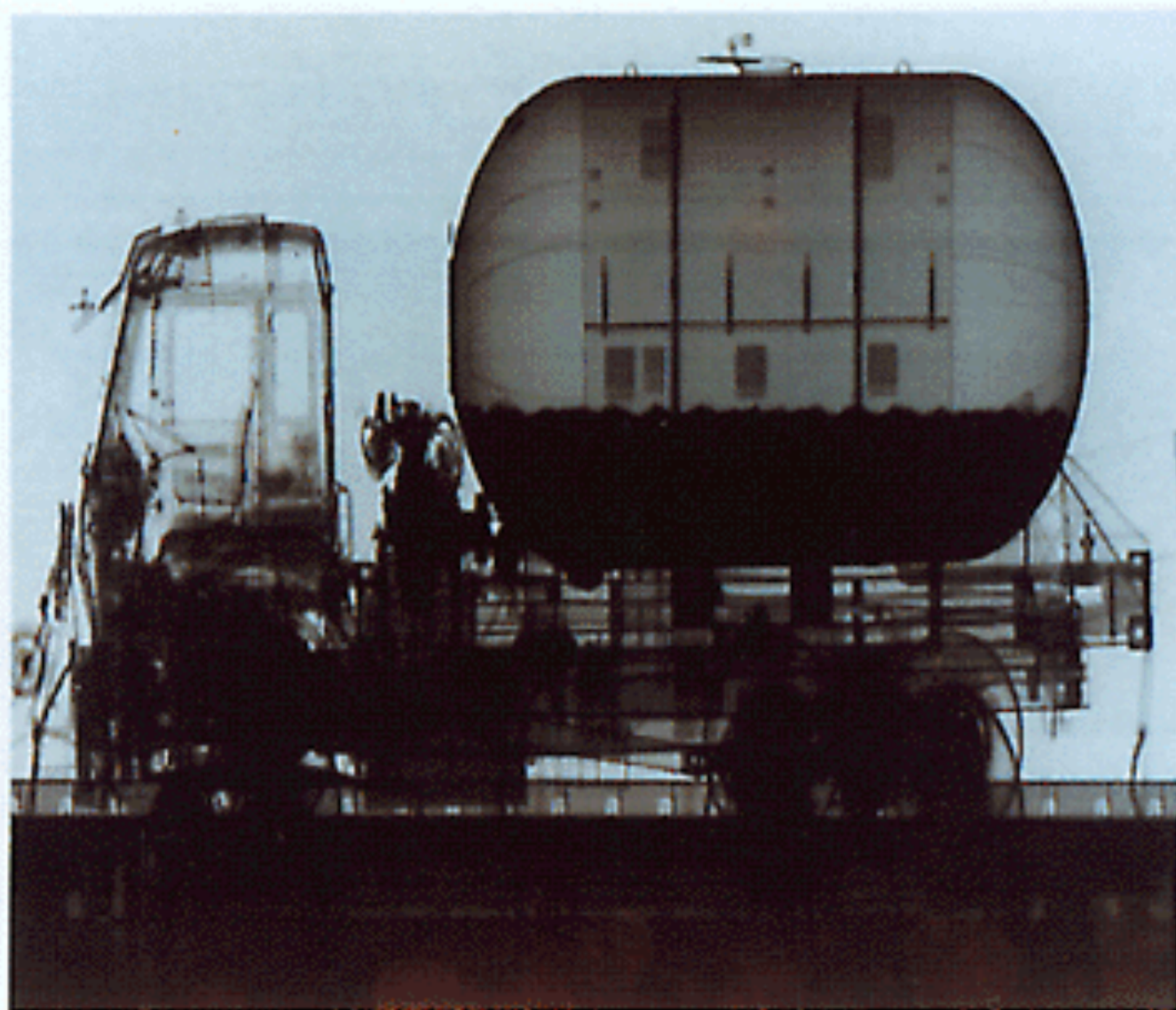


Figure VIII.4: Large-scale X-ray scan of a tanker truck revealing the fill level of the tank. Such systems can be used to detect car-bombs at highly vulnerable transportation centers.

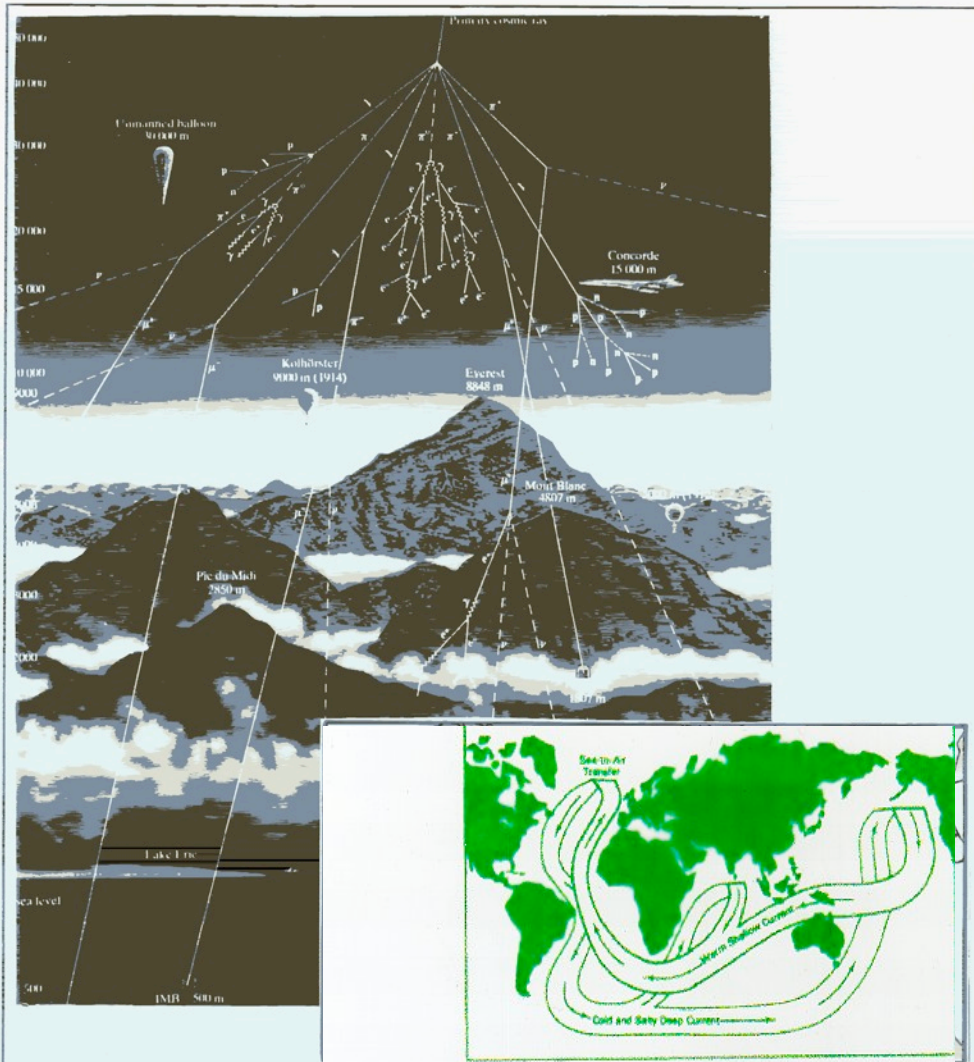
BNCP (“Boron-Neutron Capture Therapy”)

TERAPIA com NEUTRON e I. P. (LET)

MARCADORES MONOCLONAIS e REATIVOS

.....



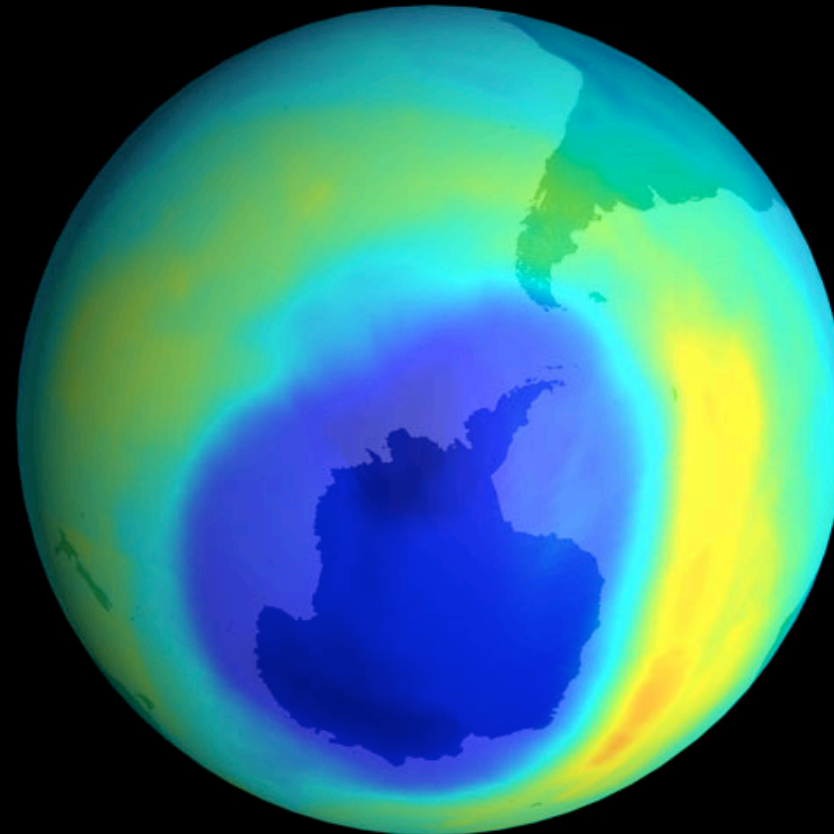


Cosmic-ray induced nuclear reactions in the atmosphere and the earth's crust generate long-lived radioisotopes that serve as wide-spread tracers and dating tools in various studies of geophysical and environmental importance. It is now possible, for example, to use accelerator mass spectrometry (AMS), a technique that was recently developed at nuclear physics accelerators, for highly sensitive carbon-14 dating to measure the age distribution of carbon-dioxide dissolved in oceanwater. This provides important information for understanding oceanic circulation patterns and their influence on the weather, and insights into the phenomenon of global warming.

$7,10\text{Be}$

Isótopos de Be criados na estratosfera em colisões de raios cósmicos em átomos de ^{14}N . Átomos de Be são ligados com facilidade em aerossóis.

Ozone • September 6, 2000 • Total Ozone Mapping Spectrometer (TOMS)



AMS- Accelerator Mass Spectrometry

MÉTODO ANALÍTICO DE IDENTIFICAÇÃO

1) Identificação e contagem de isótopos
($> 10^5$ átomos/amostra)

TRAÇADORES

2) Identificação e determinação de razões isotópicas (10^{-15} a 10^{-12})

CRONOLOGIA

ISÓTOPOS DE INTERESSE:

Raros

Radioativos com $T_{1/2}$ longa (10^4 a 10^7 a)

^{14}C 5,73Ka

^{36}Cl 301Ka

^{10}Be 1.5Ma

^{129}I 16Ma

^{26}Al 720Ka

IRRADIAÇÃO DE ALIMENTOS

γ , e, n, p Esterilização

Tomate - Preservado por um período de 2 a 3 semanas.



Banana - Preservada por um período de 3 a 4 semanas.



Mamão Papaya - Preservado por um período de 2 a 3 semanas.



Cebola - A brota é inibida por mais de 6 meses.



Identification of Microorganisms for the Analysis of Images Obtained by Neutron Radiography

J.D.R.Lopes^(a), V.Crispim^{(b)*} and C.Lage^(c)

ABSTRACT

The main difficulty to identify infectious microorganisms is the required time to obtain a reliable result, a minimum of 72 hours. We propose a reduction to about 5 hours through the technique of neutron radiography. Samples containing the bacillus *Escherichia coli* and the cocci *Staphylococcus epidermidis* were incubated with B^{10} , layered on SSNTD (CR-39) surface and irradiated in the J-9 channel from the Argonauta Reactor (IEN/CNEN) with a flux of thermal neutrons at a rate of 2.2×10^5 n/cm².s. Images were observed in an optical microscope after exposure of the plates to chemical development of the latent alpha-tracks. Analysis of the images revealed morphological differences between the species, conferring the technique the perspective to use in microbial diagnosis.



Figure 1 - Neutron radiographic image of bacilli *Escherichia coli*



Figure 2 - Electronic micrograph of *E. coli*. Copyright Dennis Kunkel, University of Hawaii



Figure 3 - Neutron radiographic image of cocci *Staphylococcus epidermidis*



Figure 4 - Electronic micrograph of *S. epidermidis*. Copyright Dennis Kunkel, University of Hawaii

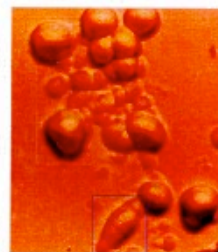


Figure 5 - Neutron radiographic image of both *E. coli* and *S. epidermidis* mixed in the same sample



Figure 6 - Agglomeration of alpha particles tracks originated from the $^{10}\text{B}(n,\alpha)^7\text{Li}$ reaction in CR-39

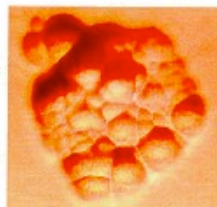


Figure 7 - Neutron radiographic image of an agglomeration of cocci *S. epidermidis*



Figure 8 - Neutron radiographic image of an agglomeration of bacilli *E. coli*

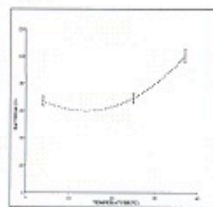


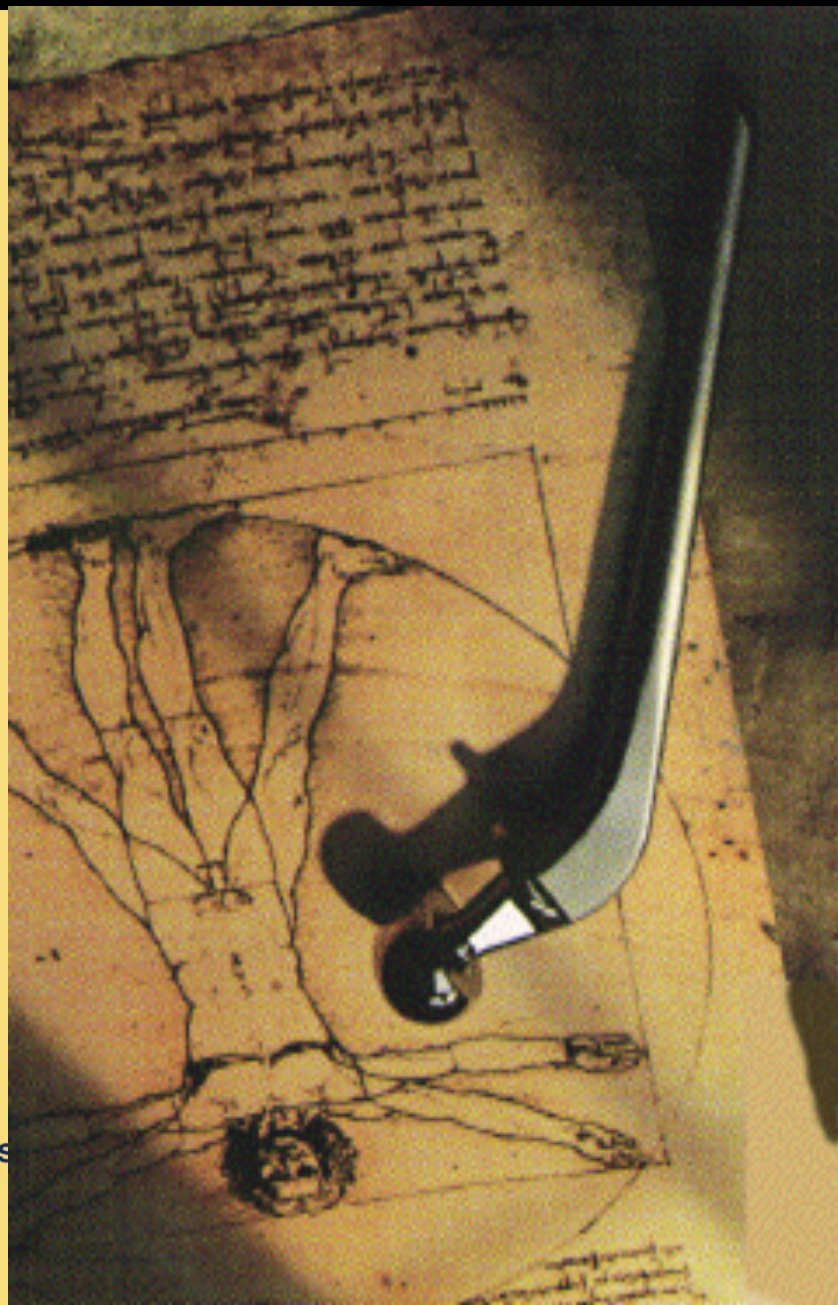
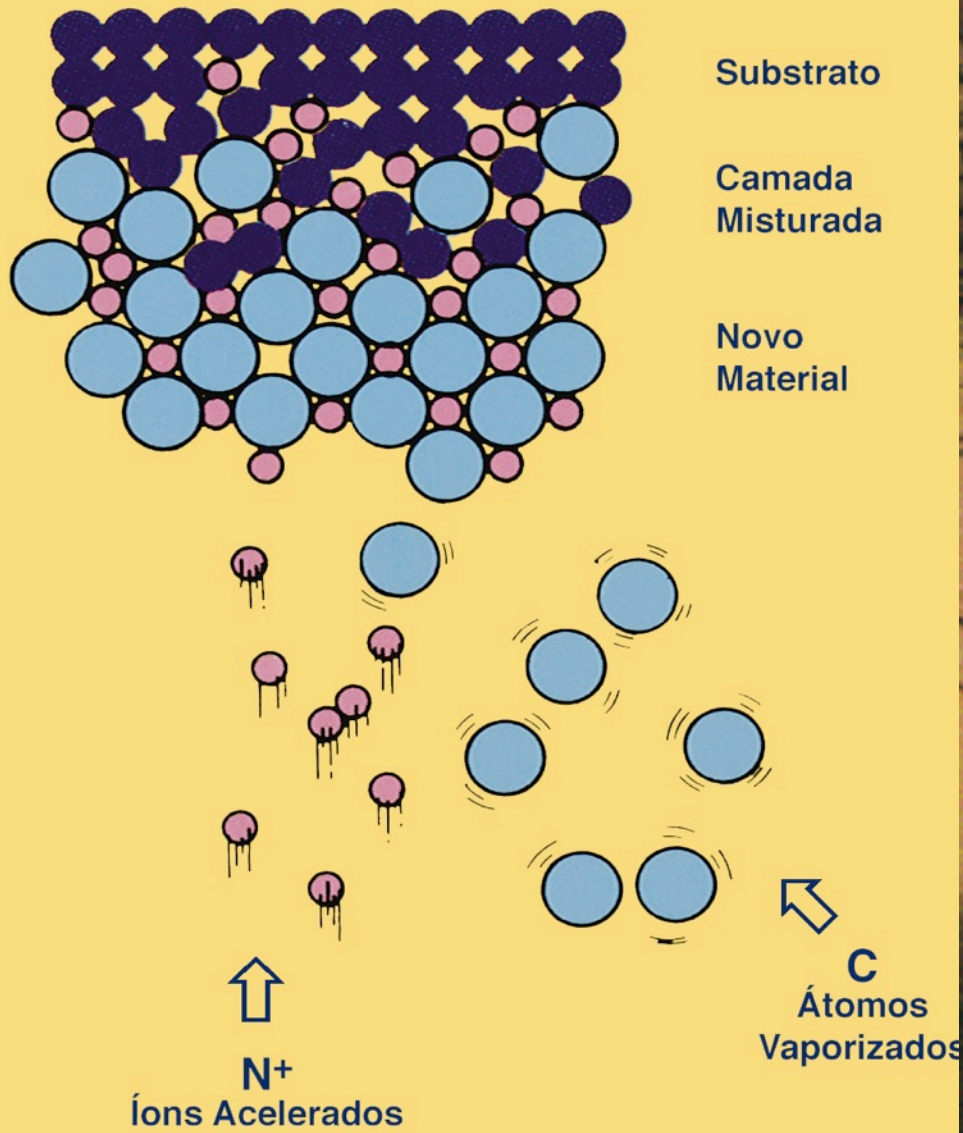
Figure 9 - Percentage of the number of scored in each sample in CR-39 strips as a function of the temperature of incubation with boron.

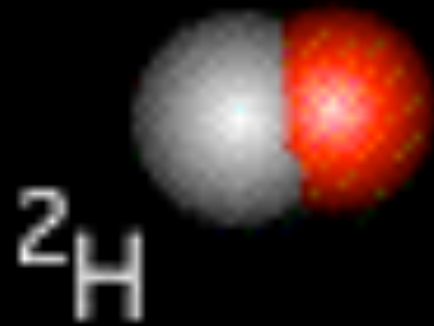
CONCLUSIONS

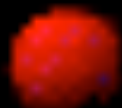
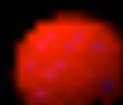
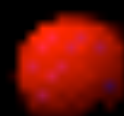
In summary, data presented in this paper indicate that the neutron radiography technique has a good perspective in applied Microbiology, serving as a tool for identification of microorganisms. Tests are now being performed to establish protocols for identification of other types of bacteria.

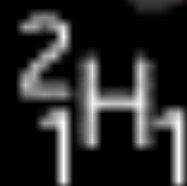
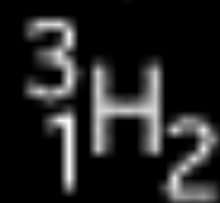
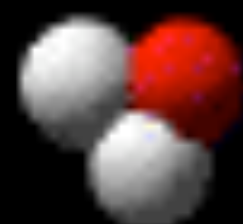
*Corresponding author's

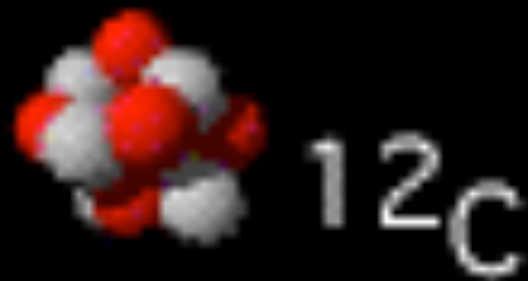
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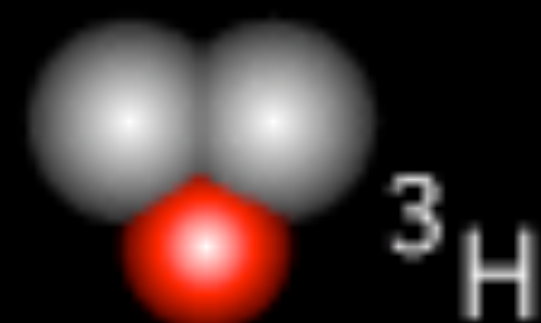














${}_{98}^{252}\text{Cf}_{154}$



n



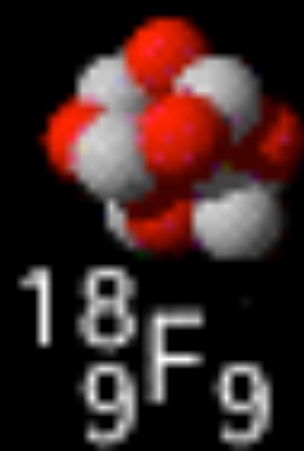
${}_{92}^{235}\text{U}_{143}$

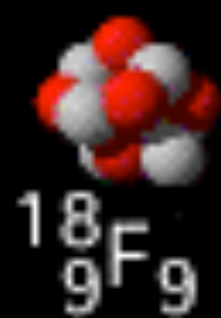


${}^{70}_{30}\text{Zn}_{40}$



${}^{208}_{82}\text{Pb}_{126}$





e



${}_{39}^{90}\text{Y}$
41

